



THREE ISOFLAVANONES FROM ERYTHRINA ORIENTALIS

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Key Word Index—*Erythrina orientalis*; Leguminosae; roots; isoflavanone; orientanols D–F.

Abstract—Three new isoflavanones, orientanols D–F, were isolated from the roots of *Erythrina orientalis* L., together with the two known isoflavanones, bidwillon A and bidwillon B. Their structures were elucidated on the basis of spectroscopic evidence. © 1998 Elsevier Science Ltd. All rights reserved

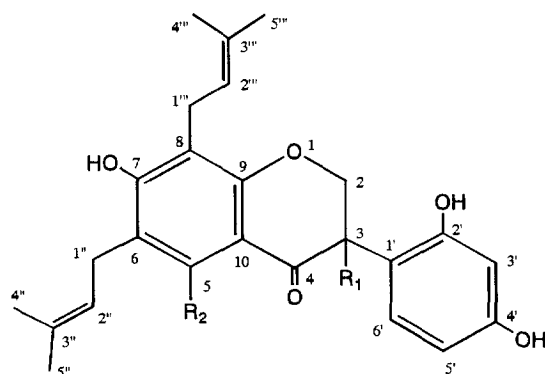
INTRODUCTION

We have reported [1] the isolation and structural determination of two pterocarpanes from the roots of *Erythrina orientalis*. Further investigation of the roots of *E. orientalis* has now led to the isolation of three novel isoflavanones (2–4), named orientanols D–F, along with two previously reported isoflavanones, bidwillon A (eriotrichin B) (1) [2, 3] and bidwillon B (5) [2].

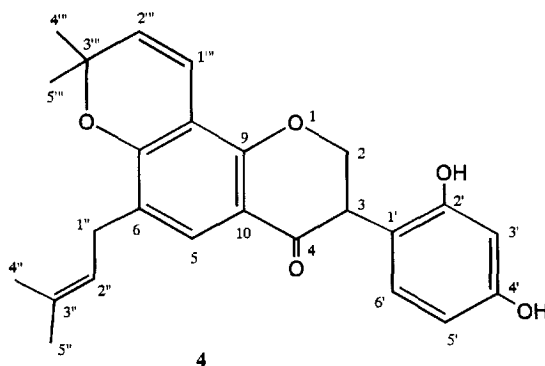
RESULTS AND DISCUSSION

Compound (1) was identified as bidwillon A by comparison of its spectral data (IR, UV and ^1H NMR) with the corresponding literature values [2, 3] and, in the ^{13}C NMR spectrum, the chemical shifts were shown to be identical with those described in the literature except for signals at δ 123.0 (C-6) and 115.4 (C-10), which were assigned by the COLOC spectrum (Table 2). This compound was characterized previously from *Erythrina* $\times$ *bidwillii* [2] and *E. eriotricha* [3].

Orientanol D (2) was obtained as a colourless oil and the molecular formula was confirmed to be $\text{C}_{25}\text{H}_{28}\text{O}_6$ by HRMS (m/z 424.1875). The IR spectrum showed the presence of a conjugated carbonyl (1650 cm^{-1}) and hydroxyl (3400 cm^{-1}) groups. The ^1H NMR spectrum revealed signals of an oxymethylene group (δ 4.37 and 4.92), ascribable to H-2 of a 3-hydroxyisoflavanone [4], four aromatic protons (δ 6.29, 6.32, 7.29 and 7.51) and two prenyl groups (δ 1.66, 1.69, 1.74, 1.77, 3.32, 3.40, 5.18 and 5.31) (Table

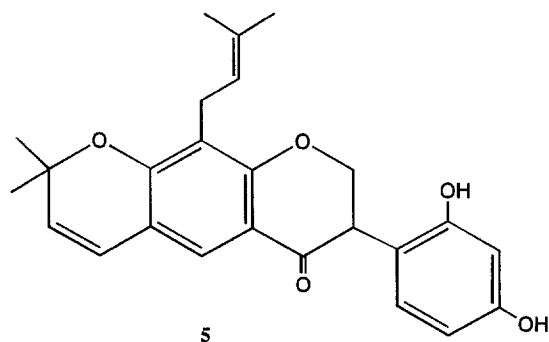


- 1 $\text{R}_1 = \text{R}_2 = \text{H}$
 2 $\text{R}_1 = \text{OH}, \text{R}_2 = \text{H}$
 3 $\text{R}_1 = \text{H}, \text{R}_2 = \text{OH}$



1). Comparison of the ^1H NMR and ^{13}C NMR spectra of 2 with those of 1 displayed identical positions of

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the four aromatic protons and the two prenyl groups, indicating the same substituent patterns for aromatic rings A and B and the assignment of the carbinol carbon at the C-3 position was decided from the HMBC spectrum which revealed correlation between C-3 (δ 74.6) and H-2 (δ 4.37 and 4.92) and/or H-6' (δ 7.29). Assignment of all the ^1H NMR and ^{13}C NMR signals of 1–4 was accomplished by analyses of the ^1H – ^1H COSY, HMQC and HMBC spectra. Therefore, the structure of orientanol D was represented as formula 2.

Orientanol E (3) was obtained as a colourless oil and the molecular formula was confirmed to be $\text{C}_{25}\text{H}_{28}\text{O}_6$ by HRMS (m/z 424.1894). The IR spectrum showed also the presence of a conjugated carbonyl and hydroxyl groups. In the ^1H NMR spectrum of 3, a chelated hydroxyl group at C-5 (δ 12.67) was observed and a further three aromatic protons (δ 6.33, 6.45 and 6.96), an oxomethylene group (δ 4.50 and 4.61), a methine proton (δ 4.24) and two prenyl groups (δ 1.65,

Table 2. ^{13}C NMR spectral data of 1–4 in acetone- d_6

C	1	2	3	4
2	71.6	74.5	71.1	71.9
3	47.5	74.6	47.3	47.7
4	192.4	191.5	199.1	191.8
5	125.9	126.2	160.6	128.0
6	123.0	123.6	108.7	124.1
7	159.7	160.2	162.1	157.3
8	116.3	116.3	107.7	109.9
9	160.4	159.8	159.1	157.2
10	115.4	113.3	103.7	115.6
1'	114.8	116.6	114.1	114.6
2'	157.2	159.6	157.0	157.5
3'	103.8	104.5	103.7	103.8
4'	158.6	158.0	158.9	158.7
5'	107.8	107.4	107.8	107.8
6'	131.1	128.5	131.7	131.3
1''	28.7	28.6	21.8	28.2
2''	122.7	122.5	123.4	123.3
3''	133.7	133.8	132.1	132.6
4''	17.8	17.8	17.9	17.9
5''	25.9	25.9	25.8	25.9
1'''	22.8	22.7	22.2	116.6
2'''	123.0	122.8	123.6	129.7
3'''	132.3	132.4	132.0	78.1
4'''	17.9	17.9	17.9	28.3*
5'''	25.9	25.9	25.8	28.4*

* Assignments in the same vertical column may be interchanged.

1.74, 1.75, 3.31, 3.33, 5.17 and 5.19) were assigned by comparison of the ^1H NMR and ^{13}C NMR spectra

Table 1. ^1H NMR spectral data of 1–4 in acetone- d_6

H	1	2	3	4
2	4.59 <i>dd</i> (11.0, 5.1) 4.68 <i>dd</i> (11.0, 9.5)	4.37 <i>d</i> (11.7) 4.92 <i>d</i> (11.7)	4.50 <i>dd</i> (11.0, 5.1) 4.61 <i>t</i> -like (11.0)	4.59 <i>dd</i> (11.0, 5.1) 4.70 <i>dd</i> (11.0, 10.3)
3	4.10 <i>dd</i> (9.5, 5.1)		4.24 <i>dd</i> (11.0, 5.1)	4.13 <i>dd</i> (10.3, 5.1)
5	7.54 <i>s</i>	7.51 <i>s</i>		7.56 <i>s</i>
3'	6.42 <i>d</i> (2.9)	6.32 <i>d</i> (2.2)	6.45 <i>d</i> (2.2)	6.43 <i>d</i> (2.2)
5'	6.31 <i>dd</i> (8.1, 2.9)	6.29 <i>dd</i> (8.8, 2.2)	6.33 <i>dd</i> (8.1, 2.2)	6.32 <i>dd</i> (8.1, 2.2)
6'	6.97 <i>d</i> (8.1)	7.29 <i>d</i> (8.8)	6.96 <i>d</i> (8.1)	6.94 <i>d</i> (8.1)
1''	3.33 <i>d</i> (7.3)	3.32 <i>d</i> (7.3)	3.33 <i>d</i> (7.3)	3.24 <i>d</i> (7.3)
2''	5.32 <i>t</i> (7.3)	5.31 <i>t</i> (7.3)	5.17 <i>t</i> (7.3)*	5.26 <i>t</i> (7.3)
4''	1.70 <i>s</i>	1.69 <i>s</i>	1.74 <i>s</i>	1.73 <i>s</i>
5''	1.74 <i>s</i>	1.74 <i>s</i>	1.65 <i>s</i>	1.71 <i>s</i>
1'''	3.41 <i>d</i> (6.6)	3.40 <i>d</i> (7.3)	3.31 <i>d</i> (7.3)	6.64 <i>d</i> (10.3)
2'''	5.19 <i>t</i> (6.6)	5.18 <i>t</i> (7.3)	5.19 <i>t</i> (7.3)*	5.74 <i>d</i> (10.3)
4'''	1.77 <i>s</i>	1.77 <i>s</i>	1.75 <i>s</i>	1.47 <i>s</i> *
5'''	1.66 <i>s</i>	1.66 <i>s</i>	1.65 <i>s</i>	1.46 <i>s</i> *
3-OH		2.15 <i>br s</i> *		
5-OH			12.67 <i>br s</i>	
7-OH	8.03 <i>br s</i> *	2.67 <i>br s</i> *	2.85 <i>br s</i> †	
2'-OH	8.18 <i>br s</i> *	2.95 <i>br s</i> *	8.20 <i>br s</i> †	8.19 <i>br s</i> †
4'-OH	8.51 <i>br s</i> *	8.43 <i>br s</i> *	8.61 <i>br s</i> †	8.51 <i>br s</i> †

*, † Assignments in the same vertical column may be interchanged.

with those of **1**. In the ^{13}C NMR spectrum, the signals of the same partial structures were also in good agreement with those of **1** except for the signal of C-1" (δ 21.8). The higher shift value at C-1" in the ^{13}C NMR spectrum of **3** suggested that the prenyl group at the C-6 position is in the vicinity of diortho-oxygenated substituents [5] and further the assignment of the C-1" position was confirmed from the HMBC spectrum which showed correlation between C-6 (δ 108.7) and H-1" (δ 3.33) and/or 5-OH (δ 12.67), indicating the obvious location of the prenyl group at C-6. Therefore, the structure of orientanol E was represented as formula **3**.

Orientanol F (**4**) was obtained as a colourless oil and the IR spectrum indicated also the presence of a conjugated carbonyl and hydroxyl groups. The mass spectrum revealed the molecular ion ($\text{C}_{25}\text{H}_{26}\text{O}_5$) at m/z 406 and the first fragment at m/z 391 [$\text{M}-15$] $^+$, typical of 2,2-dimethylchromenes [6,7]. The ^1H NMR spectrum showed, in addition to signals of the characteristic dimethylchromene part (δ 1.46, 1.47, 5.74 and 6.64), signals of an oxymethylene group (δ 4.59 and 4.70), a methine proton (δ 4.13), four aromatic protons (δ 6.32, 6.43, 6.94 and 7.56) and a prenyl group (δ 1.71, 1.73, 3.24 and 5.26) which were virtually superimposable on those of **1** and, in the ^{13}C NMR spectrum, the signals of the aromatic ring B, ring C and the prenyl group displayed the same substitution patterns as those of **1**. Further, the placement of the prenyl group at the C-6 position was confirmed by the DIFNOE experiment which displayed NOE interaction between an aromatic proton at C-5 (δ 7.56) and an aliphatic proton at C-1" (δ 3.24) and/or an olefinic proton at C-2" (δ 5.26). Therefore, the structure of orientanol F was represented as formula **4** with angular chromene arrangement and the linear isomer (**5**) was simultaneously isolated from this plant.

EXPERIMENTAL

General

Mps: uncorr.; CC: Merck silica gel 60 (230–400 mesh); TLC: glass plates precoated with Kieselgel 60 F₂₅₄ (Merck); the spots were detected by spraying with 50% H_2SO_4 and by UV light; ^1H NMR (500 and 600 MHz) and ^{13}C NMR (67.5): TMS int. standard; UV: MeOH.

Plant material

See Ref. [1].

Extraction and isolation

The CH_2Cl_2 soluble fraction was chromatographed on silica gel with solns of varying polarity of CHCl_3

as reported in ref. [1]. Frs 43–46 were purified by repeated CC (*n*-hexane– Me_2CO , 2:1) to give **3** (4.1 mg). Frs 47–50 were chromatographed on silica gel with benzene–EtOAc (5:1) and benzene–EtOAc (1:1) and subsequently by *n*-hexane– Me_2CO (2:1) and *n*-hexane– Me_2CO (1:1) to afford **1** (192 mg), **2** (20 mg), **4** (6.2 mg) and **5** (4.2 mg). The known compounds **1** [2, 3] and **5** [2] were identified by comparison of their spectral and physical data with those reported in the literature.

Bidwillon A (1). Colourless oil, $[\alpha]_D^{20} \pm 0^\circ$ (MeOH, *c* 0.1). IR (CHCl_3) ν_{max} cm^{-1} : 3600, 1650, 1600; UV λ_{max} nm: 206, 222, 284, 318; MS m/z : 408 [M] $^+$, 390, 373, 363, 335, 319, 291, 279, 273 (100%), 257, 244, 229, 217, 188, 173, 161; HRMS m/z : 408.1924 (M^+ , calcd for $\text{C}_{25}\text{H}_{28}\text{O}_5$: 408.1935); ^1H NMR: Table 1; ^{13}C NMR: Table 2.

Orientanol D (2). Colourless oil, $[\alpha]_D^{20} \pm 0^\circ$ (MeOH, *c* 0.1). IR (KBr) ν_{max} cm^{-1} : 3400, 1650, 1610; UV λ_{max} nm: 207, 221, 286, 318; MS m/z : 424 [M] $^+$, 406, 389, 363, 335, 330, 314, 295, 289, 273 (100%), 257, 243, 229, 217, 203, 201, 188, 161; HRMS m/z : 424.1875 (M^+ , calcd for $\text{C}_{25}\text{H}_{28}\text{O}_6$: 424.1884); ^1H NMR: Table 1; ^{13}C NMR: Table 2.

Orientanol E (3). Colourless oil, $[\alpha]_D^{20} \pm 0^\circ$ (MeOH, *c* 0.1). IR (KBr) ν_{max} cm^{-1} : 3400, 1640, 1610; UV λ_{max} nm: 208, 232 (sh), 296, 336; MS m/z : (424 [M] $^+$, 100%), 406, 391, 369, 363, 353, 335, 295, 289, 273, 260, 245, 233, 217, 204, 189, 177; HRMS m/z : 424.1894 (M^+ , calcd for $\text{C}_{25}\text{H}_{28}\text{O}_6$: 424.1884); ^1H NMR: Table 1; ^{13}C NMR: Table 2.

Orientanol F (4). Colourless oil, $[\alpha]_D^{20} \pm 0^\circ$ (MeOH, *c* 0.1). IR (KBr), ν_{max} cm^{-1} : 3400, 1640, 1600; UV λ_{max} nm: 206, 256, 264, 317; MS m/z : 406 [M] $^+$, 391, 388, 373, 363, 340, 337, 333, 309, 271, 255 (100%), 242, 227, 215, 187, 173; HRMS m/z : 406.1791 (M^+ , calcd for $\text{C}_{25}\text{H}_{26}\text{O}_5$: 406.1779); ^1H NMR: Table 1; ^{13}C NMR: Table 2.

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