

NERIIFOLIONE, A TRITERPENE FROM *EUPHORBIA NERIIFOLIA*

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Abstract—A novel triterpene, 9,19-cyclolanost-20(21)-en-24-ol-3-one, named neriifolione, and cycloartenol were isolated from the latex of *Euphorbia neriifolia*. Their structures were elucidated with the help of chemical and physical data (^1H NMR, ^{13}C NMR, IR and mass spectra). © 1998 Elsevier Science Ltd. All rights reserved

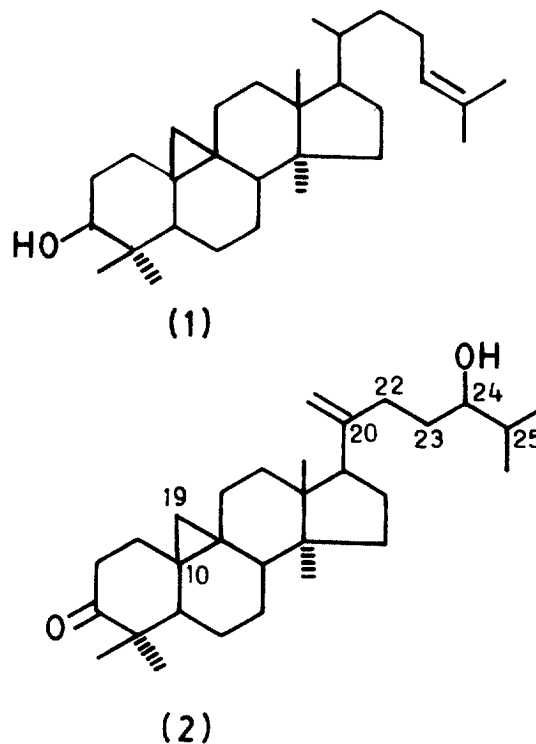
INTRODUCTION

The plant *Euphorbia neriifolia*, especially the latex, has been reported to possess a wide range of medicinal properties [1–11] which has prompted a comprehensive investigation of latex of *E. neriifolia*. A number of compounds have already been isolated from this plant [12–16]. We now report the isolation and characterization of a novel triterpene cyclolanost-20(21)-en-24-ol-3-one, named neriifolione (2), along with cycloartenol (1) from the latex of *E. neriifolia*.

RESULTS AND DISCUSSION

The dried latex of *E. neriifolia*, collected from A.M.U. Campus, Aligarh, India, was extracted successively with petrol, benzene and acetone. The petrol and benzene extract on concentration gave a very small amount of the residue. The acetone extract was concentrated and chromatographed on a silica gel column. Elution of the column with petrol–benzene (1:1) afforded granular white crystals mp. 86–90°C labelled as “compound A”. The ^1H NMR spectrum of compound A indicated it to be a mixture of two compounds. Repeated column chromatography followed by preparative TLC (benzene–chloroform 1:1) resolved it into two pure compounds EN-1 (major) and EN-2 (minor).

EN-1 (200 mg) was crystallized from chloroform–methanol as white shining crystals, mp 95°C, analyzed for $\text{C}_{30}\text{H}_{50}\text{O}$, the molecular ion peak at m/z 426. It was characterized as cycloartenol (1) by comparison of its IR, ^1H NMR and mass spectra with those of an authentic sample [17, 18].



EN-2 (70 mg) was crystallized from chloroform–methanol as white shining crystals m.p. 92°C, analyzed for $\text{C}_{30}\text{H}_{48}\text{O}_2$. The IR spectrum showed the absorption maxima at 3400 (OH), 3040 (cyclopropane ring), 1705 (C=O), 1380, 1360 (geminal dimethyl) and 880 cm^{-1} (terminal methylene group $=\text{CH}_2$). The ^1H NMR spectrum exhibited two doublets at δ 0.15 and δ 0.39 ($J = 5$ Hz) characteristic of a cyclopropane ring in 9,19-cyclotriterpenoids [18]. A positive Zimmerman

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Table 1. ^{13}C NMR chemical shift [δ c values (ppm)] of neriifolione

C no.	Chemical shift	C no.	Chemical shift
C-1	35.5	C-16	26.52
C-2	33.53	C-17	52.10
C-3	215.10	C-18	18.70
C-4	39.4	C-19	29.61
C-5	50.63	C-20	138.25
C-6	20.91	C-21	113.21
C-7	28.30	C-22	34.50
C-8	47.51	C-23	25.10
C-9	20.02	C-24	76.30
C-10	26.00	C-25	30.51
C-11	25.91	C-26	22.50
C-12	36.24	C-27	22.71
C-13	45.31	C-28	19.42
C-14	48.60	C-29	24.71
C-15	32.71	C-30	20.52

reaction [19] indicated the presence of carbonyl group at C-3. Six methyl protons appeared as a multiplet in the range of δ 0.89–1.10. A doublet at δ 4.65 ($J = 8$ Hz) was assigned to a terminal methylene group placed at $\Delta^{20(21)}$ which was supported by the peaks at m/z 339, 353 and 367 in the mass spectrum. The carbinyl hydrogen appeared as a multiplet at δ 3.25. The chemical shift of the carbinyl hydrogen ruled out the placement of OH at C-22 position (as it would be α to oxygen and allylic). The presence of the fragments at m/z 339, 353 and 367 further suggested that the hydroxyl was not at C-22. This left the positions C-24 and C-25. The presence of hydroxyl at C-25 would shift the terminal methyl to below δ 1.1, which was not observed. Furthermore, the absence of the fragment at m/z 59 due to $>\text{C}=\text{O}^+\text{H}$ indicated that the hydroxyl group was not at C-25. In the high resolution ^1H NMR spectrum, signals near δ 1.01 and 1.03 indicated two methyl doublets due to isoprenyl units. This suggests the placement of the hydroxyl group at C-24. The mass spectrum showed the molecular ion peak at m/z 440. The ^{13}C NMR (Table 1) showed one carbonyl carbon at δ 215.10. The terminal methylene carbons (C-20 and C-21) appeared at δ 138.25 and 113.21 and C-24 was observed at δ 76.30.

On the basis of the above results, the compound EN-2 was characterized as 9,19-cyclolanost-20(21)-en-24-ol-3-one and named as neriifolione (2).

EXPERIMENTAL

Mps were determined on a Reichert microscope hot stage apparatus and are uncorr. IR spectra were measured on Shimadzo IR-408. Mass spectra were obtained by electron impact a 70 eV on a JEOL JMS-300 spectrometer. ^1H NMR and ^{13}C NMR spectra were recorded on Bruker WM-400 using CDCl_3 as

solvent and TMS as internal standard. Chemical shifts are quoted in δ ppm.

Plant material

Latex of *E. neriifolia* was collected from A.M.U. campus and the plant was identified by Prof. Wazahat Hussain, Department of Botany, A.M.U., Aligarh-202002, India.

Extraction and isolation

The latex of *E. neriifolia* (500 g) was dried under red. pres., then extracted successively with petrol, benzene and Me_2CO . Petrol and benzene extracts showed similar behaviour on TLC examination. The Me_2CO extract was chromatographed over a silica gel column using petrol, petrol–benzene (1:1) as the eluting solvents. The fractions obtained from petrol–benzene (1:1) gave a solid mass, labelled as compound "A" (350 mg), crystallized from CHCl_3 – MeOH as granular crystals m.p. 86–90°. Spectral analysis of compound "A" showed it to be a mixture of two components which were separated into individual compounds EN-1 and EN-2 by repeated CC followed by prep. TLC (benzene–chloroform, 3:1).

EN-2 (70 mg). Crystallized from CHCl_3 – MeOH , shining crystals mp 92°C. It gave +ve Zimmerman test (violet colour with meta-dinitrobenzene in KOH), M^+ m/z 440 (Calc. for $\text{C}_{30}\text{H}_{48}\text{O}_2$), $[\alpha]_D^{27} -17.5^\circ$, IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} , 3400 (OH), 3040 (cyclopropane ring), 1750 ($>\text{C}=\text{O}$), 1380, 1360 (gem. dimethyl), 880 ($=\text{CH}_2$); ^1H NMR: δ 0.19 and 0.39 (2H, ABq, $J = 5$ Hz, cyclopropane ring), δ 0.89–1.1 (18 H, m , $6 \times \text{Me}$), 3.25 (1H, m , carbinyl hydrogen), 4.65 (2H, d , $J = 8$ Hz, $=\text{CH}_2$); ^{13}C NMR (Table 1). MS, m/z (rel. int.): 440 [M^+] (15), 425 [$\text{M}^+ - \text{Me}$] (12), 422 [$\text{M}^+ - \text{H}_2\text{O}$] (9), 367 (8), 353 (8.5), 339 (15), 313 [$\text{M}^+ - \text{side chain}$] (27.6), 302 (11), 175 (36).

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REFERENCES

1. Azam, K. M., Muheet-e-Azam, Vol. III. Nizami Press, Kanpur, 1892, pp. 103–105.
2. IBN-BAITAR, Z. M. A. (1197–1248), *Al-Jamey Lamufradatil Advia wal Aghzia*, Vol. I, Part-III, 114. Amara Press, Egypt, 1971, pp. 165–166.
3. Iqbal, N. K., Chandra, O. M., Gupta, K. P., Singhal, K. C. and Saxena, P. N., *Ind. J. Pharmacol.*, 1969, **1**, 98.
4. Lahon, L. C., Lhanikor, H. N. and Ahmad, N., *Ind. J. Pharmacol.*, 1979, **11**, 239–249.
5. Marium, G., Venkataraman, P. R. and Pandalai, K. M., *J. Scient. Inds. Res.*, 1947, **66**, 42–47.
6. Singh, R. and Singh, R., *Technol. Sindri*, 1972, **9**, 415–416.

7. Babbar, O. P., Joshi, M. N. and Madan, A. R., *Ind. J. Med. Res.*, 1982, **76**, 54–55.
8. Gangasatyam, N., *Sachitra Ayurveda*, 1981, **34**, 295–298.
9. Varshney, S. C. and Tyagi, N. K., *J. NIMA*, 1991, **33**, 7–9.
10. Tiwari, K. C., Majumdar, R., Bhattacharjee, S. and Chandra, S., *J. Res. Ind. Med. Yoga and Homeo*, 1978, **13**, 98–101.
11. Amin, K. M. Y., Faridi, M. A., Asif, M. and Khan, N. A., *Ind. J. Pharmacol.*, 1995, **27**, 60.
12. Ng, A. S., *Phytochemistry*, 1990, **29**, 662–664.
13. Baslas, R. K. and Agarwal, R., *Ind. J. Pharm. Sci.*, 1980, **42**, 66–67.
14. Baslas, R. K. and Agarwal, R., *Ind. J. Chem.*, 1980, **19B**, 717–718.
15. Chatterjee, A., Sarkar, K. S. and Mukhopadhyay, S., *Ind. J. Chem.*, 1978, **16B**, 1038–1039.
16. Ponsinet, G. and Ourisson, G., *Phytochemistry*, 1968, **7**, 89–98.
17. Aplin, R. T. and Hornby, G. M., *J. Chem. Soc. B*, 1966, 1078.
18. Goverdhan, C. H., Prasad, R. R. and Sundarmaniah, T., *Phytochemistry*, 1984, **23**, 411–413.
19. Zimmerman, W. Z., *Physiol. Chem.*, 1935, 237–257.