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# GUAIAN-5.12-OLIDES FROM LEONTODON HISPIDUS

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Key Word Index—Leontodon hispidus; Asteraceae; Lactuceae; sesquiterpene lactones; guaianolides.

**Abstract**—The dichloromethane extract of the whole plant of *Leontodon hispidus* afforded one new and two known guaian-5,12-olides. The structure of the new compound has been established by extensive 2D NMR techniques as 14-hydroxyhypocretenolide- $\beta$ -D-glucopyranoside-4'-14"-hydroxyhypocretenoate. © 1998 Published by Elsevier Science Ltd. All rights reserved

#### INTRODUCTION

Leontodon hispidus L. distributed over most of Europe to Asia Minor, the Caucasus and Persia is one of about 50 species of the genus Leontodon [1]. Only for one species, L. autumnalis L., has the occurrence of sesquiterpene lactones (jacquilenin and 8-deoxylactucin) been reported [2]. Guaian-5,12-olides, a rare group of sesquiterpene lactones, have so far only been found in three other members of the Lactuceae tribe of the Asteraceae: Hypochoeris cretensis (L.) Bory & Chaub., Hedypnois cretica (L.) Dum.-Courset and Crepis aurea (L.) Cass. [3–5].

## RESULTS AND DISCUSSION

The CH<sub>2</sub>Cl<sub>2</sub> extract of the air dried whole plant of *L. hispidus* was repeatedly chromatographed on Sephadex LH-20, silica gel and reversed phase material to give one new hypocretenolide derivative (3) in addition to the known compounds 14-hydroxy-hypocretenolide (1) and 14-hydroxyhypocretenolide- $\beta$ -D-glucopyranoside (2).

Identification of 1 and 2 was based on a comparison of their MS (positive ESI m/z 261 [M+H]<sup>+</sup> and 423 [M+H]<sup>+</sup>, respectively) and <sup>1</sup>H NMR data with those given in literature [4]. Since the <sup>13</sup>C NMR data from 1 and 2 have not yet been published these are shown in Table 1.

The molecular formula of 3 was determined to be  $C_{36}H_{42}O_{12}$  from the HR FAB mass spectrum (m/z)

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683.2654  $[M+H]^+$ ). The ESI mass spectrum showed a quasimolecular ion peak at m/z 683  $[M+H]^+$  and major fragments at m/z 423  $[M-260+H]^+$  and 261  $[M-422+H]^+$  suggesting substance 3 to be a derivative of 2 esterified with the parent sesquiterpenoid acid of 1.

This was confirmed by the 'H NMR spectrum (Table 2) which showed signals for two olefinic methylene groups [ $\delta_{\rm H}$  6.65 (br d,  $J_{13,7} = 1.5$  Hz) and  $\delta_{\rm H}$  5.92 (br s) as well as  $\delta_{\rm H}$  6.30 (br d,  $J_{13.7}$  = 1.5 Hz) and  $\delta_{\rm H}$ 5.78 (br s)], two olefinic methyl groups ( $\delta_H$  2.16; d,  $J_{15,3} = 1.5$  Hz), two vinylic protons located in the  $\alpha$ position to a carbonyl function ( $\delta_{\rm H}$  6.31; q,  $J_{3,15} = 1.5$ Hz and 6.03; q,  $J_{3'',15''} = 1.5$  Hz), two oxygen-bearing methylene groups ( $\delta_H$  5.30 and 4.98; d, J = 13.5 Hz;  $\delta_{\rm H}$  4.97 and 4.65; d, J = 14.5 Hz) and one anomeric sugar proton ( $\delta_{\rm H}$  4.45; d, J = 8.0 Hz). The glucose moiety was identified by a combination of selective HSQC, HSQC-TOCSY and phase sensitive COSY experiments. The <sup>13</sup>C NMR spectrum showed the presence of 36 carbons: two methyls, 11 methylenes, nine methines and 14 quarternaries. Fifteen carbon signals assignable to the sesquiterpene lactone groups of 3 were almost superimposable on the corresponding signals of 2, whereas those of the sesquiterpene group with the opened y-lactone ring showed the following shift differences: the signals of C-4" ( $\delta$  177.7), C-6" ( $\delta$ 40.4) and C-11" ( $\delta$  146.5) were shifted downfield by 7.1, 7.8 and 8.2 ppm, respectively, and those of C-5"  $(\delta 79.9)$ , C-8"  $(\delta 28.8)$  and C-13"  $(\delta 124.9)$  shifted upfield by 10.0, 5.6 and 6.0 ppm (Table 2). In comparison with the  $^{13}$ C NMR data of 2 the C-4′ ( $\delta$  72.7) signal of the glucose moiety was shifted to highfield (+1.2 ppm) and the resonances of C-3' ( $\delta$  75.7) and C-5' ( $\delta$  75.3) were shifted downfield (-2.6 and -3.0ppm, respectively) indicating acylation of the

Table 1. <sup>1</sup>H NMR data of compound 3 (500 MHz, δ-values in MeOH-d<sub>s</sub>)†

	Н	н	
3	6.31 1H, q (1.5)	3"	6.03 1H, q (1.5)
6	2.55 1H, ddd (14.5, 4.5, 1.0)	6"	2.04  1H, d (11.0)
	2.16 1H,†		2.01  1H, d(11.0)
7	3.39 1H, ddd (4.5, 3.0, 1.5)	7″	2.64 1H,†
8	2.64 1H, dddd (15.0, 13.0, 3.0, 1.0)	8"	2.25 1H, m
	2.10 1H,†		1.83 1H, m
9	2.90 1H, ddd (16.5, 5.5, 2.0)	9"	3.01 1H, ddd (16.0, 7.5, 7.5)
	2.49 1H, ddd (16.5, 13.0, 2.0)		2.76 1H, ddd (16.0, 7.0, 5.0)
13	6.65 1H, br d (1.5)	13"	6.30 1H, $br d(1.5)$
	5.92 1H, <i>br s</i>		5.78 1H, br s
14	5.30 1H, d (13.5)	14"	4.97 1H, d (14.5)
	4.98 1H, d (13.5)		4.65  1H, d(14.5)
15	2.16 3H, <i>d</i> (1.5)	15"	2.16  3H, d(1.5)
1'	4.45 1H, d (8.0)		
2'	3.36 1H, dd (9.5, 8.0)		
3′	3.64 1H, dd (9.5, 9.5)		
4′	4.90 1H, dd (9.5, 9.5)		
5′	3.52 1H, ddd (9.5, 5.5, 2.0)		
6′	3.64 1H, dd (12.0, 2.0)		
	3.56 1H, dd (12.0, 5.5)		

<sup>†</sup> Signals are overlapping.

hydroxyl group at C-4' [8]. This was verified by a HMBC experiment (Table 3) which showed long range couplings between H-4' of the sugar moiety and the carbonyl carbon 12" of the sesquiterpenoid acid part of the molecule. Correlations; between the geminal protons of C-14 and C-1' as well as the anomeric proton H-1' and C-14 confirmed the linkage between the glucose and the 14-hydroxyhypocretenolide moiety.  $\beta$ -Glucosidation was proven by the charac-

teristic coupling of the anomeric proton at C-1'  $(J_{1',2'} = 8.0 \text{ Hz})$ . According to Bohlmann and Singh [3] the 5,12-guaianolide ring requires an equatorial orientation of H-7, which is in agreement with the observed coupling between H-7  $(\delta_{\text{H}} \ 3.39; \ ddd)$  and H-13  $(\delta_{\text{H}} \ 6.65; \ d, \ J_{7,13} = 1.5 \ \text{Hz})$ . The configuration at C-7" has been proven to be identical with that at C-7, since 3 in a mixture of water and methanol (3:2) decomposes into equimolar amounts of 1 and 2. This

<sup>‡</sup> Coupling constants (J) in Hz are given in parentheses.

<sup>§</sup> All assignments are based on HMBC and HSQC experiments.

Table 2. <sup>13</sup>C NMR data of compounds 1–3 (75.1 MHz,  $\delta$  values in MeOH- $d_s$ )

С	1	2	3	С	3
ı	137.1 <i>s</i>	136.4	137.5	1"	136.2
2	195.2 s	194.7	194.7	2"	197.6
3	136.0 d	134.7	135.6	3"	131.8
4	171.5 s	170.6	171.1	4"	177.7
5	91.0 s	89.9	90.7	5"	79.9
6	33.4 t	32.6	32.7	6"	40.4
7	39.0 d	38.2	38.6	7"	37.3
8	35.1 t	34.5	34.4	8"	28.8*
9	$26.0 \ t$	25.7	25.8	9"	28.9*
10	157.9 s	154.1	154.6	10"	153.4
11	138.9 s	138.3	138.7	11"	146.5
12	167.3 s	165.9	166.9	12"	167.8
13	$131.0 \ t$	130.9	130.5	13"	124.9
14	61.5 t	68.3	68.3	14"	61.9
15	13.0 q	11.9	12.7	15"	12.9
l'		104.5 d	104.7		
2′		74.5 d	76.0		
3′		78.3 d	75.7		
4′		71.5 d	72.7		
5′		78.3 d	75.3		
6′		61.8 t	62.4		

<sup>\*</sup> Signals may be interchangeable.

Table 3. HMBC data for compound 3

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Н	С	Н	С
3	1, 2, 4, 5, 15	3"	1", 2", 4", 5", 15"
6	1, 4, 5	6"	1', 5", 8"
	4, 5		1', 5", 8"
7	-	7"	_
8	10	8"	7", 9", 10"
	7		7", 9"
9	1, 7, 8, 10	9"	1", 7", 8", 10", 14"
	1, 7, 8, 10, 14		1", 7", 8", 10", 14"
13	7, 11, 12	13"	7", 11", 12"
	7, 12		7", 12"
14	1, 9, 10, 1'	14"	1", 9", 10"
	1, 9, 10, 1'	14*"	
15	3, 4, 5	15"	3", 4", 5"
1′	14		
2′	1', 4'		
3′	2', 5'		
4′	2', 6'		
5′	3', 4'		
6′	5′		
	5'		

reaction could be inhibited by using pure MeOH as solvent or by acidifying aqueous MeOH with TFA. Thus, 3 is 14-hydroxyhypocretenolide-β-D-glucopyranoside-4'-14"-hydroxyhypocretenoate.

### EXPERIMENTAL

#### Plant material

Leontodon hispidus L. was collected in June 1996 in the vicinity of Innsbruck. A voucher specimen is deposited at the Institute of Pharmacognosy.

## Extraction and isolation of compounds 1-3

Whole, air dried plants (3.1 kg) were ground and extracted exhaustively at room temp. with CH<sub>2</sub>Cl<sub>2</sub> yielding 50 g of residue after evaporation of the solvent *in vacuo*. Fifteen grams of the residue were chromatographed on a Sephadex LH-20 column using MeOH as solvent. Frs containing compounds 1–3 were combined and repeatedly chromatographed on silica gel columns using gradients of EtOAc-MeOH. Final purification of compounds 1–3 was carried out by MPLC on a RP-18-column (Merck) using a gradient of TFA (0.01%)—MeOH-MeCN (3:1) to give 1 (8 mg), 2 (5 mg) and 3 (35 mg).

## Decomposition of 3

One mg amounts of 3 were dissolved respectively in 1.0 ml  $\rm H_2O-MeOH$  (3:2), pure MeOH and  $\rm H_2O-MeOH$  (3:2) acidified with TFA (pH 1.5) and kept at room temp. for 24 h. The solns were analyzed by HPLC using a Zorbax Rx-C18 4.6 mm  $\times$  25 cm column (P.N. 880967.902), a gradient of  $\rm H_2O-MeOH$  starting from 4:1 to 39:61 in 24 min; flow rate: 1.0 ml min<sup>-1</sup>; detection: 245 nm.  $R_i$ s of 1–3: 13.4, 10.6 and 19.4 min, respectively.

Compound 3. Amorphous wax-like substance melting between 96° and 102°. FAB-MS (positive ion), (rel. int.): m/z 683.2654 [M+H]<sup>+</sup> (100); ESIMS (positive ion), (rel. int.): m/z 705.5 [M+Na]<sup>+</sup> (14), 683.5 [M+Na]<sup>+</sup> (18), 665.5 [M-H<sub>2</sub>O+H]<sup>+</sup> (23), 423.3 [M-sesquiterpenic acid+H]<sup>+</sup> (27), 405.2 [M-sesquiterpenid acid-H<sub>2</sub>O+H]<sup>+</sup> (62), 261.2 [M-sesquiterpenid acid-glucose+H]<sup>+</sup> (100), 243.2 [M-sesquiterpenid acid-glucose-H<sub>2</sub>O+H]<sup>+</sup>] (39).

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<sup>†</sup> Assignments are based on DEPT, HMBC and HSQC experiments.

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