

CHROMONE GLYCOSIDES AND FLAVONOIDS FROM *HYPERICUM JAPONICUM*

QING-LI WU*, SHENG-PING WANG, LI-JUN DU, SHU-MING ZHANG, JUN-SHAN YANG and PEI-GEN XIAO

Institute of Medicinal Plant Development, Peking Union Medical College and Chinese Academy of Medical Sciences, Beijing 100094, P. R. China

(Received 12 January 1998; received in revised form 12 January 1998)

Key Word Index—*Hypericum japonicum*; Clusiaceae; chromone glycoside; flavonoid; coagulant activity; 5, 7-dihydroxy-2-(1-methylpropyl) chromone-8- β -D-glucoside; 5, 7-dihydroxy-2-isopropylchromone-8- β -D-glucoside; 7, 8-(2'', 2''-dimethylpyrano)-5, 3', 4'-trihydroxy-3-methoxyflavone; (2R,3R) dihydroquercetin-3, 7-O- α -L-dirhamnoside.

Abstract—From the aerial parts of *Hypericum japonicum*, two novel chromone glycosides, 5, 7-dihydroxy-2-(1-methylpropyl) chromone-8- β -D-glucoside and 5, 7-dihydroxy-2-isopropylchromone-8- β -D-glucoside, and two new flavonoids, 7, 8-(2'', 2''-dimethylpyrano)-5, 3', 4'-trihydroxy-3-methoxyflavone and (2R, 3R) dihydroquercetin-3, 7-O- α -L-dirhamnoside were isolated together with nine known flavonoids. Their structures were deduced from spectroscopic and chemical evidence. Some of the compounds were found to exert an interesting coagulant activity in an *in vitro* test. © 1998 Elsevier Science Ltd. All rights reserved

INTRODUCTION

Hypericum (Clusiaceae) is a large genus of herbs or shrubs, which occurs widely in temperate regions of the world. In China there are 55 species and 8 subspecies and half of them have been used in Chinese herbal medicine mainly for the treatment of infectious hepatitis. The recent surge of interest in chemistry of this genus has led to the isolation of more than 100 components with different biological activities. In particular, extracts of *H. perforatum* are now widely used in Europe as drugs for the treatment of depression [1–3].

H. japonicum Thunb. ex Murray is a Chinese medicinal plant [4]. Previous papers reported the isolation of thirteen phloroglucinol derivatives [4–8], five flavonoids [9, 10], a peptide [11], a lactone [12] and two xanthenes [10, 13]. As a chemical investigation of this plant, two new chromone glycosides, two novel flavonoids and nine known flavonoids were isolated; some of them showed good coagulant bioactivity *in vitro*.

RESULTS AND DISCUSSION

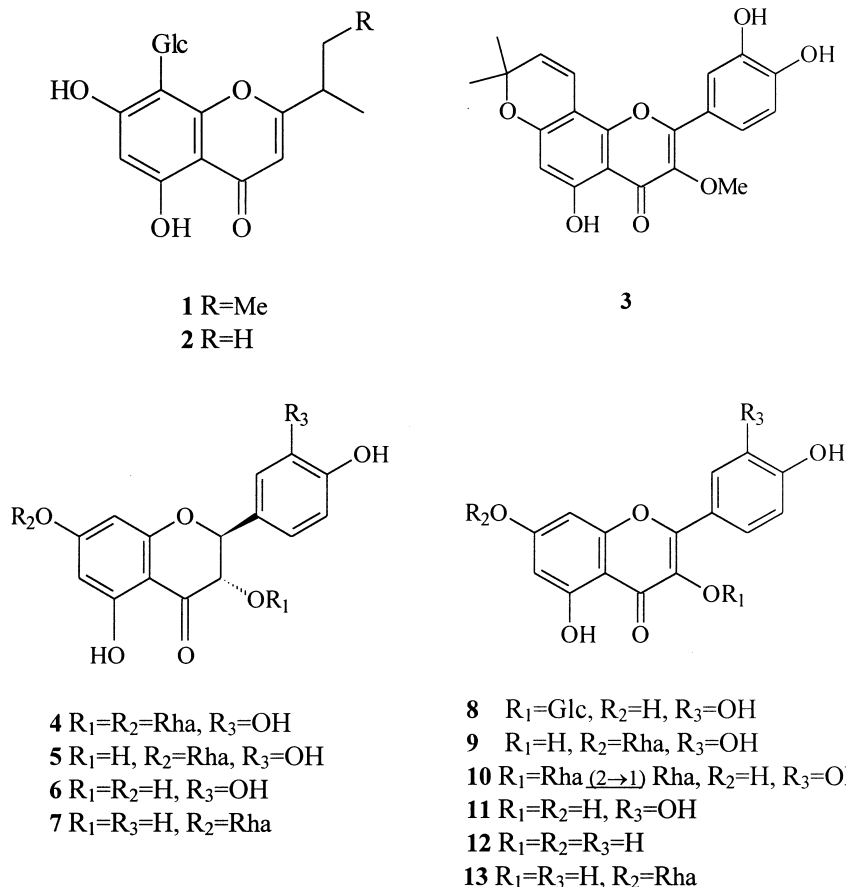
The ethanol extract of dried aerial parts was extracted with petrol, CH₂Cl₂, EtOAc, Me₂CO and

MeOH in a Soxhlet. The EtOAc and Me₂CO fractions were purified and thirteen chromone glycosides and flavonoids were obtained, four of which are new compounds (**1**, **2**, **3** and **4**).

Compound **1**, white crystals, mp 80–82°, [α]_D²⁵ = +13.4° (MeOH), has the molecular formula of C₁₉H₂₄O₉ based on EI-MS (M⁺ *m/z* 396), ¹H NMR and ¹³C NMR. Both the UV (MeOH) and IR (KBr) absorption maxima (250, 270 nm and 1660, 1620, 1590, 1430 cm^{−1}, respectively) of the compound were typical of a 5, 7-dioxygenated chromone nucleus [14].

In its ¹H NMR (500 MHz, DMSO-d₆) spectrum, the upfield proton signals at δ 2.63 (1H, *m*, H-1'), 1.70, 1.56 (each 1H, *m*, H-2'^{a,b}), 1.21 (3H, *d*, *J* = 6.9 Hz, Me-1') and 0.86 ppm (3H, *t*, *J* = 7.4 Hz, H-3') indicated the presence of 1-methylpropyl group. The aromatic proton singlets at δ 6.20 (1H), 6.12 ppm (1H) were due to H-6 or H-8 and H-3 [14–16]. The anomeric proton signal at δ 4.63 (1H, *d*, *J* = 10.0 Hz, H-1'') and carbon signals at δ 81.5 (C-5''), 78.7 (C-3''), 73.3 (C-1''), 71.2 (C-2''), 70.9 (C-4''), 61.7 (C-6'') in its ¹³C NMR (125 MHz, DMSO-d₆) showed the presence of β -D-glucose moiety. In its HMBC, the proton signals at δ 4.63 (H-1'') and 6.12 ppm (H-3) corresponding to the carbon signals at δ 164.2 (C-7), 104.7 (C-8), 156.6 (C-8a) and 39.8 (C-1') respectively, suggested the location of glucose moiety and 1-methylpropyl to be at C-8 and C-2 (Figure 1). Thus **1** is 5, 7-dihydroxy-2-(1-methylpropyl)chromone-8- β -D-glucoside. From the ¹H-¹H COSY and ¹³C-¹H COSY, all of the carbon signals were assigned (see Experimental).

* Author to whom correspondence should be addressed.



Compound **2**, gave white powder, mp 152–154°, $[\alpha]_D^{25} = +9.8^\circ$ (MeOH), with molecular formula of C₁₈H₂₂O₉, based on EI-MS (M^+ m/z 382), ¹H NMR and ¹³C NMR. The bands at 250, 270 in its UV (MeOH) and 1655, 1620, 1430 cm⁻¹ in its IR (KBr) showed presence of 5, 7-dioxygenated chromone nucleus [14].

The ¹H NMR and ¹³C NMR spectra of **2** closely resembled those of **1** except for upfield proton and carbon signals. In its ¹H NMR (500 MHz, DMSO-*d*₆) spectrum, the upfield proton signals at δ 2.88 (1H, m, H') and 1.26, 1.25 (each 3H, *d*, $J=6.6$ Hz, 1'-Me's) indicated the presence of isopropyl group at C-2. The aromatic proton singlets at δ 6.25 (1H), 6.14 ppm (1H) were due to H-6 and H-3 [14–16], respectively. The anomeric proton signal at δ 4.64 (1H, *d*, $J=9.8$ Hz, H1'') and carbon signals at δ 81.5 (C-5''), 78.6 (C-3''), 73.1 (C-1''), 71.2 (C-2''), 70.8 (C-4''), 61.8 (C-6'') in its ¹³C NMR (125 MHz, DMSO-*d*₆) showed the presence of β -D-glucose moiety at C-8. Thus taking into consideration all mentioned above data and analyses, **2** was established as 5, 7-dihydroxy-2-isopropylchromone-8- β -D-glucoside, which is in agreement with its ¹³C NMR spectral data (see experimental).

Compound **3**, pale yellow crystals, mp 190–192°, has the molecular formula, C₂₁H₁₈O₇, which was established by its EI-MS (M^+ m/z 382), ¹H NMR and

¹³C NMR spectra. The UV (MeOH) bands at 248, 282sh, 324 nm suggested a flavonol skeleton. The ¹H NMR (300 MHz, DMSO-*d*₆) spectrum of **3** showed the aromatic proton signals at δ 7.54 (1H, *d*, $J=2.0$ Hz), 7.42 (1H, *dd*, $J=2.0, 8.5$ Hz), 6.90 (1H, *d*, $J=8.5$ Hz) due to H-2', H-6' and H-5' of B-ring. The AB-type doublets at δ 6.60 and 5.78 (each 1H, *d*, $J=10.0$ Hz, H-4'' and H-3'') together with a six-proton singlet at δ 1.43 (2''-Me's) indicated a 2, 2-dimethylpyran attached to A-ring. In addition, its ¹H NMR spectrum revealed a methoxyl singlet at δ 3.79. To confirm the substitution pattern, a NOE difference experiment was conducted. Irradiation of the methoxyl signal resulted in significant enhancement of two aromatic proton signals at δ 7.54 (H-2') and 7.42 (H-6') indicating the methoxyl group at C-3. The aromatic singlet at δ 6.46 was due to C-6 or C-8. In its ¹³C NMR (75 MHz, DMSO-*d*₆) spectrum, the downfield shift (*ca.* 10.3 ppm) for C-8 suggested the prenylated substitution at C-8 [17, 18] and its HMBC (Figure 2) confirmed the structure elucidation as **3** which is in agreement with its ¹³C NMR spectral data (see experimental). This is only a second report of a prenylated flavonoid from the family [10].

Compound **4**, a white crystals, mp 247–249°, $[\alpha]_D^{25} = -44.8^\circ$ (MeOH), gave a molecular ion peak m/z 595 [M-H] in the negative FAB mass spectrum and a

molecular ion peak of aglycone m/z 304 in its EI-MS. The UV (MeOH) bands at 274sh, 286 and 340sh nm suggested a dihydroflavonol structure. In the ^1H NMR (500 MHz, $\text{DMSO}-d_6$) spectrum of **4**, the aromatic proton signals at δ 6.88 (1H, *brs*), 6.74 (2H, *m*), 6.17, 6.13 (each 1H, *d*, $J=2.1$ Hz) and aliphatic signals at δ 5.30 and 4.71 (each 1H, *d*, $J=10.3$ Hz) suggested the presence of dihydroquercetin as the aglycone of **4**, and should be described to H-2', H-5', 6', H-8, H-6, H-2 and H-3, respectively [9]. A *trans*-diaxial relationship between the protons at C-2 and C-3 in **4** was evident from the 10.3 Hz coupling constant. Acidic hydrolysis of **4** yielded L-rhamnose and an aglycone, which was spectroscopically (^{13}C NMR) identical with the known (2*R*, 3*R*) dihydroquercetin (**6**). In its ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$), the characteristic glycosylation shifts +4.0 and -2.2 ppm were observed for C-3 and C-7, indicating the locations of the rhamnose moieties to be the C-3 and C-7 [19]. The anomeric protons at δ 5.45 (1H, *brs*, 7 Rha H-1) and 4.00 (1H, *brs*, 3 Rha H-1) in the ^1H NMR established the α -configuration of rhamnose. Thus **4** is (2*R*, 3*R*) dihydroquercetin-3, 7-*O*- α -L-dirhamnoside. From the ^1H - ^1H COSY and ^{13}C - ^1H COSY of **4**, all of the carbon signals were assigned (see Experimental).

Compounds **5**-**13** were identified as (2*R*, 3*R*) dihydroquercetin-7-*O*- α -L rhamnoside (**5**) [9], (2*R*, 3*R*) dihydroquercetin (**6**) [18], 2, 3-*trans*-dihydro-3, 5, 4'-trihydroxyflavonol 7-*O*- α -L-rhamnoside (**7**) [20], quercetin 3-*O*- β -D-glucoside (**8**) [8, 21], quercetin-7-*O*- α -L-rhamnoside (**9**) [9], quercetin-3-*O*- α -L-rhamnosyl(1 \rightarrow 2)-*O*- α -L-rhamnoside (**10**) [22], quercetin (**11**), Kaempferol (**12**) [23] and Kaempferol-7-*O*- α -L-rhamnoside (**13**) [24], respectively, by spectroscopic analysis.

H. japonicum has been used for treatment of internal haemorrhage as a folk medicine in some regions of China [25]. So the isolated compounds were tested for their coagulant activity *in vitro* system. The results indicated that **8**, and **10** showed anticoagulation of APTT (Activated Partial Thromboplastin Time Reagent, purchased from Pacific Homostasis, USA), and **5**, **9** and **11** showed promoting coagulation of APTT. Whereas **9**, **10** and **11** were found to exert promoting coagulation of PT (Prothrombin Time Reagent, purchased from Pacific Homostasis, USA). Notably **9**, a flavonol glycoside with high concentration in *H. japonicum*, exhibited an excellent bioactivity. It can promote the coagulation of APTT and PT with dose-effect relationship and the effective concentration is to 10^{-5} g/L.

EXPERIMENTAL

General

All mps are uncorr. ^1H , ^{13}C NMR and 2D NMR spectra were recorded with TMS as int. standard. Chromatography separation were carried out on silica

gel, polyamide and Sephadex LH-20, TLC on silica gel G and polyamide film.

Plant material

Aerial parts of *Hypericum japonicum* Thunb. ex Murray were collected from Anhui province, China, in Sept., 1994. Voucher specimens are deposited in Institute of Medicinal Plant Development, Chinese Academy of Medical Sciences, Beijing, P. R. China.

Extraction and isolation

Air-dried aerial part (12 Kg) were extracted with 95% EtOH three times. The ethanol extract (1.1 Kg) was re-extracted successively with petrol, CH_2Cl_2 , EtOAc, Me_2CO and MeOH in soxhlet extractor. The EtOAc fr. (200 g) were chromatographed on silica gel column using a step-gradient CHCl_3 -MeOH and 1000 ml fr. was collected. Those frs. containing similar component as judged by TLC were combined and got ten frs. 1-10. Fr. 2 was repeatedly rechromatographed on polyamide (MeOH- H_2O and petrol- CHCl_3 -MeOH) followed by purification on Sephadex LH-20 (MeOH) to give **3** (60 mg). Fr. 4 was rechromatographed repeatedly (silica gel and polyamide, CHCl_3 -MeOH) and **6** (120 mg) and **12** (80 mg) were obtained. After recrystallization, **11** (2000 mg) was obtained from fr. 6. Fr. 8 was rechromatographed on polyamide (CHCl_3 -MeOH) and followed by gel filtration on Sephadex LH-20 (MeOH) to give **7** (120 mg) and **13** (60 mg). The acetone fr. (300 g) were chromatographed on silica gel column using a step-gradient CHCl_3 -MeOH and got fifteen frs. 1-15. Rechromatography of fr. 7 on polyamide (CHCl_3 -MeOH) and Sephadex LH-20 (MeOH) gave **5** (150 mg) and **9** (500 mg). Fr. 9 was repeatedly rechromatographed on polyamide (CHCl_3 -MeOH) and Sephadex LH-20 (MeOH) to give **4** (90 mg), **8** (200 mg), **10** (130 mg) and a mixture **1** and **2**. The mixture was further chromatographed on polyamide (*n*-BuOH-MeOH- H_2O -AcOH) and **1** (30 mg) and **2** (20 mg) were obtained.

5, 7-dihydroxy-2-(1-methylpropyl) chromone-8- β -D-glucoside (**1**), ^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 13.0 (1H, *br*, chelated H-5), 6.20 (1H, *s*, H-6), 6.12 (1H, *s*, H-3), 4.63 (1H, *d*, $J=10.0$ Hz, H-1''), 2.63 (1H, *m*, H-1'), 1.70, 1.56 (each 1H, *m*, 2'-H's), 1.21 (3H, *d*, $J=6.9$ Hz, Me-1'), 0.86 (3H, *t*, $J=7.4$ Hz, H-3'); ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$): δ 181.9 (C-4), 173.0 (C-2), 164.2 (C-7), 160.4 (C-5), 156.6 (C-8a), 106.2 (C-3), 104.7 (C-8), 103.3 (C-4a), 98.7 (C-6), 81.5 (C-5''), 78.7 (C-3''), 73.3 (C-1''), 71.2 (C-2''), 70.9 (C-4''), 61.7 (C-6''), 39.1 (C-1'), 27.0 (C-2'), 17.4 (Me-1'), 11.4 (C-3').

5, 7-dihydroxy-2-isopropyl chromone-8- β -D-glucoside (**2**), ^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 13.0 (1H, *br*, chelated H-5), 6.25 (1H, *s*, H-6), 6.14 (1H, *s*, H-3), 4.64 (1H, *d*, $J=9.8$ Hz, H-1''), 2.88 (1H, *m*, H-1'), 1.26, 1.25 (each 3H, *d*, $J=6.6$ Hz, 1'-Me's); ^1H NMR (300 MHz, CD_3OD): δ 6.24 (1H, *s*, H-6), 6.08

(1H, s, H-3), 4.90 (buried in solvent, H-1''), 2.93 (1H, m, H-1'), 1.35, 1.34 (each 3H, d, $J=6.6$ Hz, 1'-Me's); ^{13}C NMR (125 MHz, DMSO- d_6): δ 182.2 (C-4), 174.4 (C-2), 162.8 (C-7), 160.4 (C-5), 156.5 (C-8a), 105.0 (C-3), 104.1 (C-8), 102.0 (C-4a), 98.5 (C-6), 81.5 (C-5''), 78.6 (C-3''), 73.1 (C-1''), 71.2 (C-2''), 70.8 (C-4''), 61.8 (C-6''), 32.7 ((C-1'), 20.0, 19.5 (1'-Me's).

7, 8-(2'', 2''-dimethylpyrano)-5, 3', 4'-trihydroxy-3-methoxyflavone (**3**), ^1H NMR (300 MHz, DMSO- d_6): δ 13.07 (chelated OH-5), 7.54 (1H, d, $J=2.0$ Hz, H-2''), 7.42 (1H, dd, $J=8.5, 2.0$ Hz, H-6'), 6.90 (1H, d, $J=8.5$ Hz, H-5'), 6.60 (1H, d, $J=10.0$ Hz, H-4''), 6.46 (1H, s, H-6), 5.78 (1H, d, $J=10.0$ Hz, H-3''), 3.79 (3H, s, 3-OMe), 1.43 (6H, s, 2''-Me's); ^{13}C NMR (75 MHz, DMSO- d_6): δ 177.5 (C-4), 158.0 (C-7), 155.4 (C-2), 154.8 (C-5), 154.7 (C-9), 148.3 (C-4'), 144.6 (C-3'), 137.2 (C-3), 128.5 (C-3''), 120.0 (C-6', 1'), 115.1 (C-2'), 115.0 (C-5'), 113.9 (C-4''), 104.6 (C-10), 103.8 (C-8), 94.1 (C-6), 77.4 (C-2''), 59.1 (3-OMe), 27.2 (2''-Me's); EI-MS m/z (%): 382 [M^+] (31.2), 267 [M-Me^+] (100), 352 (12.4), 324 (6.0), 251 (4.2), 184 (9.8), 167 (9.2), 153 (7.9), 139 (8.8).

(2R, 3R) dihydroquercetin-3, 7-O- α -L-dirhamnoside (**4**), ^1H NMR (500 MHz, DMSO- d_6): δ 6.88 (1H, brs, H-2'), center at 6.74 (2H, m, H-5', 6'), 6.17 (1H, d, $J=2.1$ Hz, H-8), 6.13 (1H, d, $J=2.1$ Hz, H-6), 5.30 (1H, d, $J=10.3$ Hz, H-2), 4.71 (1H, d, $J=10.3$ Hz, H-3), 5.45 (1H, brs, 7Rha H-1), 4.00 (1H, brs, 3Rha H-1), 1.10 (3H, d, $J=6.2$ Hz, 3Rha-Me), 1.04 (3H, d, $J=6.4$ Hz, 7Rha-Me); ^{13}C NMR (125 MHz, DMSO- d_6): δ 195.1 (C-4), 164.2 (C-7), 162.8 (C-5), 161.9 (C-9), 145.9 (C-4'), 145.1 (C-3'), 126.7 (C-1'), 118.6 (C-6'), 115.3 (C-2'), 114.7 (C-5'), 102.5 (C-10), 100.0 (3 Rha C-1), 98.3 (7 Rha C-1), 96.5 (C-6), 95.4 (C-8), 81.6 (C-2), 75.7 (C-3), 70.2 (3 Rha C-2), 70.4 (3 Rha C-3), 71.6 (3 Rha C-4), 69.7 (3 Rha C-5), 70.0 (7 Rha C-2), 70.2 (7 Rha C-3), 71.5 (7 Rha C-4), 68.9 (7 Rha C-5), 17.7 (3 Rha C-6), 17.5 (3 Rha C-6); EI-MS m/z (%): 304 [M-rha-rha^+] (16.9), 286 (21.5), 153 (59.1), 123 (100); FAB-MS (m/z): 595 [M-H^+], 449 [M-Rha-H^+], 303 [M-Rha-rha-H^+].

REFERENCES

- Hobbs, C., *Herbal Gram.*, 1989, **18/19**, 24.
- Leuschner, J., *2nd International Congress on Phytomedicine and 7th Congress of the German Society of Phytotherapy*, September 11–14 1996, Munich, Germany, SL-80.
- Wheatly, D., *2nd International Congress on Phytomedicine and 7th Congress of the German Society of Phytotherapy*, September 11–14 1996, Munich, Germany, SL-86.
- Ishiguro, K., Yamaki, M., Kashihara, M. and Takagi, S., *Planta Med.*, 1986, **52**, 288.
- Ishiguro, K., Yamaki, M., Kashihara, M. and Takagi, S., *Planta Med.*, 1987, **53**, 415.
- Gu, G. M., Feng, S. Z. and Wang, X. Y., *Huaxue Xuebao*, 1988, **46**, 246.
- Ishiguro, K., Yamaki, M., Takagi, S., Yamagata, T. and Tomita, K., *J. Chem. Soc., Chem. Commun.*, 1985, 26.
- Ishiguro, K., Yamaki, M., Kashihara, M. and Takagi, S., Isoi, K., *Planta Med.*, 1990, **56**, 274.
- Ishiguro, K., Nagata, S., Fukumoto, H., Yamaki, T. S. and Isoi, K., *Phytochemistry*, 1991, **30**, 3152.
- Ishiguro, K., Nagata, S., Fukumoto, H., Yamaki, M., Takagi, S., Isoi, K. and Yoshiaki, O., *Phytochemistry*, 1993, **32**, 1585.
- Ishiguro, K., Nagata, S., Fukumoto, H. and Yamaki, M., Isoi, K., *Phytochemistry*, 1991, **30**, 3639.
- Ishiguro, K., Nagata, S., Fukumoto, H., Yamaki, T. S. and Isoi, K., *Phytochemistry*, 1990, **29**, 1010.
- Ishiguro, K., Nagareya, N., Suitani, A. and Fukumoto, H., *Phytochemistry*, 1997, **44**, 1065.
- Jimenez, C., Marcos, M., Villaverde, M. C., Riguera, R., Castedo, L. and Stermitz, F., *Phytochemistry*, 1989, **28**, 1992.
- Bohlmann, F., Misra, L. N. and Jakupovic, J., *Planta Med.*, 1984, **50**, 174.
- Parmar, V. S., Jha, H. N., Gupta, A. K., Prasad, A. K. and Gupta, S., *Tetrahedron*, 1992, **48**, 1281.
- Fukai, T. and Nomuar, T., *Phytochemistry*, 1988, **27**, 259.
- Wagner, H., Chari, M. V. and Sonnenbichlen, J., *Tetrahedron Letters*, 1976, **0**, 1799.
- Markham, K. R. and Ternai, B., *Tetrahedron*, 1976, **32**, 2607.
- Cooke, R. G. and Haynes, H. F., *Aust. J. Chem.*, 1960, **13**, 150.
- Lim, Y. A., Mei, M. C., Kusumoto, I. T., Miyashiro, H., Hattori, M., Gupta, M. P. and Correa, M., *Phytotherapy Research*, 1997, **11**, 22.
- Li, F. and Liu, Y. L., *Yaoxue Xuebao*, 1988, **23**, 672.
- Markham, K. R., Ternai, B., Stanley, R., Geiger, H. and Mabry, T. J., *Tetrahedron*, 1978, **34**, 1389.
- Zaitsev, V. G., Fursa, N. S. and Belyaeva, L. G., *Khim. Prir. Soedin.*, 1983, **19**, 527.
- Pan, Y. H. and Guo, B. L., *Journal of Chinese Medicinal Materials* 1993, **16**(3), 14.