

Phytochemistry 50 (1999) 477-480

Triterpene saponins from Glinus lotoides var. dictamnoides¹

Arafa I. Hamed^a, Nasr A. El-Emary^{b, *}

^aFaculty of Science, South Valley University, Aswan 81528, Egypt ^bFaculty of Pharmacy, Assiut University, Assiut 71526, Egypt

Received 9 April 1998; revised 22 June 1998; accepted 23 June 1998

Abstract

Two 5 β -H-hopane triterpenoidal saponins (glinusides D and E) were isolated on reinvestigation of the *n*-butanol soluble fraction of *Glinus lotoides* var. *dictamnoides*, in contrast to the 5 α -H, 6-keto-hopane derivatives isolated from the same source. The structures of these saponins were established by extensive spectral analysis as 3-O- β -L-arabinopyranosyl-22- α -L-rhamnopyranosyl (1 \rightarrow 4)- β -D-glucopyranosyl-5 β -H-16 β -hydroxyhopane and 3-O- β -L-arabinopyranosyl-22-O- β -D-glucopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-5 β -H-16- β -hydroxyhopane, respectively. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: Glinus lotoides var. dictamnoides; Molluginaceae; 5β-H-hopane; Triterpenoidal saponins; Glinusides D and E

1. Introduction

In a previous publication, we have reported the isolation and structure elucidation of three new saponins (Hamed, Sprenguel, El-Emary, Mitome, Miyaoka, & Yamada, 1996) from the aerial parts of the n-butanol soluble fraction of *Glinus lotoides* L.=G. dictamnoides Burm. (=Mollugo glinus A. Rich) (Boules, 1995). All of these saponins have been shown to be 5α -H, 6-ketohopane glycosides carrying trisaccharide or disaccharide units. The present paper deals with the isolation and structure elucidation of two new saponins with the hopane skeleton, but with different substitution patterns, from the same source.

2. Results and discussion

Chromatographic separation of the *n*-butanol soluble fraction obtained from the methanol extract of the aerial parts of *G. lotoides* var. *dictamnoides* led to isolation of further new glycosidic compounds. The first new compound (1) showed its $[M+2Na]^+$ ion peak at m/z 944 in the FAB mass spectrum, which was consistent with $C_{47}H_{78}O_{16}$. The ¹H NMR spectrum displayed the presence of six methyl singlets at δ 0.71, 0.93, 0.98, 1.06, 1.55 and

1.61, and two methyl singlets at δ 1.40. There was also a broad singlet integrated for one proton at δ 2.03 for H-5, which showed a cross peak (HMBC) with C-3 (H-5/C-3, δ 2.03/88.99) and also a cross peak between H-3/C-5 (δ 3.43/31.19). The downfield shift of H-5 and its carbon $(\delta 31.19)$ indicates an A/B cis-ring fusion (Bernardo, Pinto, & Parente, 1996; Pomilio, Gonzalez, & Eceizabarrena, 1996). There was a signal for one proton overlapping with sugar signals at δ 4.25 for H-16, which showed a cross peak (HMBC) with C-13 and C-21 (δ 4.25/49.7/67.6). The ¹H NMR spectrum displayed signals for three anomeric protons at δ 6.56 (1H, br s), 5.17 (1H, d, J = 7.6 Hz) and 4.89 (1H, d, J = 7.4 Hz). A three proton doublet at δ 1.70 (J=6.13 Hz) was typical for the methyl group of a 6-deoxy sugar. The 13C NMR spectrum revealed the presence of six saturated quaternary carbons $(\delta 38.78, 40.34, 42.81, 43.50, 45.41 \text{ and } 83.11)$ and three anomeric carbons (δ 98.43, 101.85 and 106.40). The number and chemical shifts of the tertiary methyl functions, quaternary carbons and shift of C-5 (which resonated at δ 31.19) suggested that compound 1 was a hopane triterpene triglycoside with the 5β-H configuration. Upon hydrolysis of compound 1 with 5% HCl-methanol, it gave arabinose, glucose and rhamnose as sugar components, which were identified by TLC by direct comparison with authentic samples (Barua, Datta, Ray, & Venkateswaran, 1986). The anomeric centers of the glucosyl and arabinosyl moieties were each determined to have the β -configuration based on large $J_{1,2}$ values (7.4– 7.6 Hz). The linkage of the 22-O-disaccharide and 3-O-monosaccharide was deduced from the 2-D COSY,

¹ A preliminary report about this study has been presented in the International Symposium on Plant Glycosides (ISPG, August 12–15, 1997), Kunming, Yannan, China.

^{*} Corresponding author.

HMBC and NOESY spectra, where there were cross peaks between H-1"/C-22 (δ 5.17/83.11) H-1"'/C-4" (δ 6.56/66.9) and H-1'/C-3 (δ 4.89/88.99). The ¹H NMR and ¹³C NMR spectra revealed that the aglycone was a triterpenoid with a hopane skeleton (Wilkins, Elix, Gaual, & Moberg, 1989; Tsuda & Isobe, 1965; Beierbeck, Saunders, & Apsimon, 1977; Patra, Meitra, Tapan, & Bar, 1981) and contained three hydroxyl groups (two secondary at C-3, C-16 and one tertiary at C-22).

All the data above suggested that compound 1 was a glycoside with a new aglycone, 3β , 16β , 22β -trihydroxy- 5β -H-hopane, which is isolated from a natural source for the first time. Hence, the isolated compound has the structure 3-O- β -L-arabinopyranosyl-22- α -L-rhamnopyranosyl($1\rightarrow 4$)- β -D-glucopyranosyl- 5β -H-16- Π -hydroxyhopane (1); this compound has been given the name glinuside D.

The second new compound (2) exhibited its $[M+K+H_2O]^+$ ion peak at m/z 971 in the FAB mass spectrum and was consistent with C₄₇H₇₈O₁₇. The ¹H NMR revealed the presence of six methyl singlets at δ 0.70, 0.95, 1.00, 1.17, 1.52 and 1.58, and two methyl singlets at δ 1.40; there was also a broad singlet integrated for one proton at δ 2.03 for H-5 of similar configuration with compound (1). The ¹H NMR spectrum contained signals for three anomeric protons at δ 5.15 (1H, d, J = 7.65 Hz), $\delta 4.99 \text{ (1H, } d, J = 7.15 \text{ Hz})$ and $\delta 4.92 \text{ over-}$ lapping with water signal. Acid hydrolysis of compound 2 afforded glucose and arabinose, which were identified by TLC (direct comparison with authentic samples). Comparison of the ¹H NMR and ¹³C NMR data (Tables 1 and 2) of 2 with those of 1 showed the same sugar moieties as 1 except that rhamnose was replaced by another glucose unit, which showed a downfield shift of its carbon at δ 107.69. From the above evidence, the structure of compound 2 was proved to be 3-O-β-L-arabinopyranosyl-22-O- β -D-glucopyranosyl(1 \rightarrow 4)- β -D-glucopyranosyl- 5β -H,16- β -hydroxyhopane, which is isolated from a natural source for the first time and this compound has been given the name glinuside **E**.

3. Experimental

Optical rotations were measured on a JASCO-360 digital polarimeter; UV spectra were obtained on a Hitachi 200-10 spectrophotometer; IR spectra were taken on a JASCO IR-A-2 spectrometer; ¹H NMR, ¹³C NMR, ¹H¹H COSY, NOESY, HMQC and HMBC spectra were taken on a Bruker AM-400, Bruker AM-500; MS were obtained on a Hitachi Rmu-7M spectrometer.

3.1. Extraction and isolation of the compounds

As described earlier in Hamed et al. (1996).

3.2. Acid hydrolysis of the saponin

The saponin (5 mg) was refluxed with 5 ml, 5% HCl–MeOH on a steam bath for 6 h. The product was diluted with H₂O and extracted with CHCl₃ in a separatory funnel and the respective aglycones were separated. The aq. filtrate was neutralized with AgCO₃, filtered and evapd. The resulting syrup was subjected to TLC on silica gel using EtOA–MeOH–HOAc–H₂O (13:3:4:3) as developing solvent against reference sugars, and the dried chromatograms was sprayed with *p*-anisaldehyde–H₂SO₄ reagent.

3.3. Glinuside D (1)

White powder, $[\alpha]_D$ 9.73 (c = 0.26 MeOH); IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3245. FAB–MS m/z (rel. int.): 944 [M + 2Na]⁺ (25) (C₄₇H₇₈O₁₆), 736 [M–2Na–Glc]⁺ (13) along with frag-

Table 1 1 H NMR spectral data for the isolated saponins (400 MHz, pyridine- d_{5})

Compound	Glinuside D	Glinuside E
Aglycone mo	pieties	
C-1	1.65, 1.00 (each 1H, m)	1.65, 1.00 (each 1H, m)
C-2	2.24, 1.93 (each 1H, m)	2.45, 1.93 (each 1H, m)
C-3	3.43 (1H, dd , $J=4.09$, 9.14 Hz)	3.54 (1H, dd, J = 5.16, 10.84 Hz)
C-4	=	=
C-5	2.03 (1H, s)	2.03 (1H, s)
C-6	1.07, 1.38 (each 1H, <i>m</i>)	1.07, 1.38 (each 1H, <i>m</i>)
C-0 C-7	1.95 (2H, <i>m</i>)	1.95 (2H, <i>m</i>)
C-8	1.93 (211, m)	1.93 (211, m)
C-9	1 27 (111) ^a	1 10 (111) ^a
	1.27 (1H) ^a	1.18 (1H) ^a
C-10	- 1.52 1.20 (1.1II)	- 1.52 1.20 (
C-11	1.53, 1.29 (each 1H, <i>m</i>)	1.53, 1.29 (each 1H, m)
C-12	1.34, 1.29 (each 1H, <i>m</i>)	1.34, 1.29 (each 1H, m)
C-13	1.53 (1H, <i>m</i>)	1.53 (1H, <i>m</i>)
C-14	_	_
C-15	1.60, 1.15 (each 1H, <i>m</i>)	1.60, 1.15 (each 1H, <i>m</i>)
C-16	4.25 (1H) ^a	$4.25 (1H)^a$
C-17	1.89 (1H) ^a	1.90 (1H) ^a
C-18	_	_
C-19	1.40, 0.90 (each 1H, m)	1.40, 0.90 (each 1H, m)
C-20	1.60, 1.35 (each 1H, m)	1.70, 1.40 (each 1H, m)
C-21	2.73 (1H, <i>m</i>)	2.72 (1H, <i>m</i>)
C-22	_	_
C-23	0.98 (3H, s)	1.00 (3H, s)
C-24	1.06 (3H, s)	1.17 (3H, s)
C-25	0.71 (3H, s)	0.70 (3H, s)
C-26	0.93 (3H, s)	0.95 (3H, s)
C-27	1.61 (3H, s)	1.58 (3H, s)
C-27 C-28		
	1.55 (3H, s)	1.52 (3H, s)
C-29	1.40 (3H, s)	1.40 (3H, s)
C-30	1.40 (3H, s)	1.40 (3H, s)
Sugar moieti	es	
Arabinose		
1′	4.89 (1H, d, J=7.4 Hz)	4.92 (1H) ^a
2′	4.02 (1H) ^a	4.05 (1H) ^a
3'	4.00 (1H) ^a	4.01 (1H) ^a
4′	4.17 (1H) ^a	4.17 (1H) ^a
5′	3.72, 4.16 (each 1H) ^a	3.72, 4.16 (each 1H) ^a
J	3.72, 4.10 (cacii 111)	3.72, 4.10 (cach 111)
Clusses		
Glucose	5 17 (1H J J 7 (H)	£16(1H J I 7(5H)
1"	5.17 (1H, d, J=7.6 Hz)	5.16 (1H, d, J=7.65 Hz)
2"	3.99 (1H, d, J = 8.4 Hz)	3.99 (1H, d, J=8.4 Hz)
3"	4.13 (1H, d, J = 8.4 Hz)	4.15 (1H, d, J = 8.4 Hz)
4"	4.86 (1H, <i>m</i>)	4.92 (1H, <i>m</i>)
5"	3.70 (1H, <i>m</i>)	3.78 (1H, <i>m</i>)
6"	4.30, 4.40 (each 1H) ^a	4.30, 4.40 (each 1H) ^a
	nd glucose, respectively	
1‴	6.56 (1H, <i>br s</i>)	4.99 (1H, d, J = 7.20 Hz)
2""	4.87 (1H) ^a	4.92 (1H) ^a
3‴	4.37 (1H) ^a	4.14 (1H) ^a
4‴	4.27 (1H) ^a	4.86 (1H) ^a
5‴	4.23 (1H, <i>m</i>)	3.80 (1H, t, J=4.89 Hz)
6‴	1.72 (3H, d , J =6.13 Hz)	4.32 (1H, <i>m</i>), 4.56
	(- ,,)	, , ,,,

Multiplicity was detected by DEPT experiment. ^a Overlapping with other signals.

Table 2 13 C NMR spectral data for the isolated saponins (100 MHz, pyridine- d_5)

<i>d</i> ₅)				
Compound	Glinuside D	Glinuside E		
Aglycone moieties				
C-1	27.5 (t)	27.2(t)		
C-2	26.9 (t)	26.5 (t)		
C-3	88.9 (d)	88.8 (d)		
C-4	45.4 (s)	46.0 (s)		
C-5	31.1 (<i>d</i>)	31.1 (<i>d</i>)		
C-6	39.3 (t)	39.2 (t)		
C-7	39.2 (t)	39.2 (t)		
C-8	42.8 (s)	42.8 (s)		
C-9	48.9 (d)	48.9 (d)		
C-10	38.8 (s)	38.8 (s)		
C-11	21.2 (t)	21.1 (t)		
C-12	23.9 (t)	23.8 (t)		
C-13	49.7 (d)	49.7 (d)		
C-14	43.5 (s)	43.5 (s)		
C-15	44.1 (t)	44.1 (t)		
C-16	77.9(d)	77.8 (d)		
C-17	50.6 (d)	50.5 (s)		
C-18	40.3 (s)	40.3 (s)		
C-19	43.7 (t)	43.7(t)		
C-20	45.0 (t)	45.0(t)		
C-21	67.6 (<i>d</i>)	67.6 (d)		
C-22	83.1 (s)	83.0 (s)		
C-23	17.3 (q)	17.3(q)		
C-24	17.3 (q)	17.3 (q)		
C-25	17.1 (q)	17.1 (q)		
C-26	17.1 (q)	17.1 (q)		
C-27	18.7 (q)	18.7 (q)		
C-28	18.4 (q)	18.4 (q)		
C-29	23.8 (q)	23.8 (q)		
C-30	24.9 (q)	25.0 (q)		
Sugar moieties				
Arabinose	10414	404 7 4 7		
1'	106.4 (d)	106.5 (d)		
2'	74.1 (d)	74.1 (d)		
3'	78.6 (d)	78.4 (<i>d</i>)		
4′	71.7 (d)	71.0 (d)		
5′	66.9 (t)	66.8 (t)		
Glucose				
1"	98.4 (d)	98.4 (d)		
2"	79.0 (d)	79.0 (d)		
3"	75.3 (d)	75.6 (d)		
4"	66.9 (d)	66.9 (d)		
5"	78.6 (d)	78.6 (d)		
6"	62.8 (t)	62.8 (t)		
Rhamnose and glucose, respectively				
1'''	101.8 (d)	107.8 (d)		
2""	72.5 (d)	79.5 (d)		
3‴	71.5 (d)	75.0 (d)		
4‴	77.9(d)	66.8 (d)		
5‴	69.6 (d)	78.4 (d)		
6'''	18.4 (q)	62.9 (t)		

Multiplicity was detected by DEPT experiment.

ment ions at 625 (19), 340 (50), 325(100), 183 (58) and 165 (54); ¹H NMR (400 MHz, pyridine- d_5) and ¹³C NMR (100 MHz, pyridine- d_5): (Tables 1–2).

3.4. *Glinuside E* (2)

White powder, $[\alpha]_D$ 4.34 (c=0.05 MeOH); IR $v_{\rm max}^{\rm KBr}$ cm⁻¹: 3245. Negative FAB–MS m/z (rel. int.): 971 $[M+K+H_2O]^+$ (45) $(C_{47}H_{80}O_{17})$, 752 $[M-K-Glc-H_2O]^+$ (32) along with fragment ions at 672 (52), 610 (56), 555 (65), 497 (59), 325 (65), 233 (71) and 115 (100); ¹H NMR (400 MHz, pyridine- d_5) and ¹³C NMR (100 MHz, pyridine- d_5): (Tables 1 and 2).

Acknowledgements

The authors are grateful to the staff of the analytical laboratory, Tokyo University of Pharmacy and Life Science, for spectral analysis.

References

Barua, A. K., Datta, P. K., Ray, S., & Venkateswaran, R. V. (1986). *Phytochemistry*, 25, 2577.

Beierbeck, H., Saunders, J. K., & Apsimon, J. W. (1977). Canadian Journal of Chemistry, 55, 2813.

Bernardo, R. R., Pinto, A. V., & Parente, J. P. (1996). *Phytochemistry*, 43, 465.

Boules, L. (1995). Flora of Egypt: checklist (p. 9). Cairo: Al Hadara Publishing.

Hamed, A. I., Sprenguel, I., El-Emary, N. A., Mitome, H., Miyaoka, H., & Yamada, Y. (1996). *Phytochemistry*, 43, 183.

Patra, A., Meitra, A., Tapan, K., & Bar, A. K. (1981). Organic Magnetic Resonance, 17, 148.

Pomilio, A. B., Gonzalez, M. D., & Eceizabarrena, C. C. (1996). *Phytochemistry*, 41, 1393.

Tsuda, Y., & Isobe, K. (1965). Tetrahedron Letters, 3337.

Wilkins, A. L., Elix, J. A., Gaual, K. L., & Moberg, R. (1989). Australian Journal of Chemistry, 42, 1415.