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# Aromatic compounds from the liverwort *Plagiochila spinulosa*<sup>1</sup>

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#### Abstract

Two known and five novel 9,10-dihydrophenanthrene derivatives, two orsellinic acid derivatives and a lunularic acid derivative have been isolated from the liverwort *Plagiochila spinulosa* and identified following NMR studies. A further known 9,10-dihydrophenanthrene derivative has been identified using <sup>1</sup>H NMR fingerprinting of the extracts. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Plagiochila spinulosa; Hepaticae; Liverwort; NMR fingerprinting; GC-MS; <sup>1</sup>H and <sup>13</sup>C NMR parameters; Aromatic compounds; 9,10-Dihydrophenanthrenes

# 1. Introduction

We have carried out phytochemical studies of a variety of liverwort species that are native to Scotland and have isolated a range of terpenoids, e.g. sesquiterpenoids from Chiloscyphus pallescens (Connolly, Harrison, & Rycroft, 1982) and Saccogyna viticulosa (Connolly, Harrison, & Rycroft, 1994) and diterpenoids from Nardia scalaris (Connolly, Harrison, Phillips, & Rycroft, 1984) and Scapania undulata (Huneck et al., 1986). Liverworts belonging to the genus Plagiochila are known to be rich sources of not only terpenoids but also aromatic compounds (Asakawa, 1995), and we have previously reported the 1-(3,4-dihydroxy-5-methoxybenzyl)-3of methylbut-2-ene in P. rutilans from Cuba (Huneck, Connolly, Harrison, Joseph, & Pócs, 1984) and 3,4-dihydroxy-3'-methoxybibenzyl in P. exigua (Rycroft, Cole, & Aslam, 1998), as well as the sesquiterpenoids plagiochiline C and atlanticol in P. atlantica (Rycroft & Cole, 1998) and plagiochiline V in *P. porelloides* (Söderström et al., 1999). A recent report of a new and still unidentified *Plagiochila* species (Anton, Kraut, Mues, & Morales, 1997) that contains 9,10-dihydrophenanthrenes has prompted us to provide full details of our studies (Rycroft, 1990; Connolly, 1997) of the aromatic constituents of *P. spinulosa* (Dicks.) Dum., which is commonly encountered in the west of Britain (Watson, 1981). Several collections of *P. spinulosa* have been made and, using both GC–mass spectrometry and <sup>1</sup>H NMR fingerprinting of the crude extracts, it has been found that differences are minor and that all the compounds described herein are present in all the samples in similar proportions.

#### 2. Results and discussion

This report deals with the nonterpenoid constituents of three representative samples (A–C) of *P. spinulosa* that were collected in the west of Scotland and in the English Lake District (the structural elucidation of the diterpenoid — bibenzyl conjugates spinuloplagins A, B (Rycroft, 1990) and C will be reported elsewhere). The ether extract of sample A was subjected to normal phase vacuum liquid chromatography to afford seven compounds described here. The first compound was 9,10-

<sup>&</sup>lt;sup>1</sup> Part 7 in the series 'NMR Fingerprinting of Liverworts'. For part 6 see Söderström, Rycroft, Cole, and Wei (1999).

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$$H_{3}CO \stackrel{6}{\longleftrightarrow} R^{2}O \stackrel{1}{\longleftrightarrow} R^{2} = CH_{3}$$

$$R^{2}O \stackrel{1}{\longleftrightarrow} R^{2} = CH_{3}, R^{2} = H$$

$$(9) R^{1} = R^{2} = CH_{3}$$

$$(1) R^{1} = R^{2} = CH_{3}, R^{2} = H$$

$$(2) R^{1} = CH_{3}, R^{2} = H$$

$$(9) R^{1} = R^{2} = H$$

$$(13) R^{1} = H, R^{2} = CH_{3}$$

$$(14) R^{1} = R^{3} = H, R^{2} = R^{4} = CH_{3}, R^{3} = OCH_{3}$$

$$(14) R^{1} = R^{3} = H, R^{2} = R^{4} = CH_{3}, R^{3} = OCH_{3}$$

$$(14) R^{1} = R^{3} = H, R^{2} = R^{4} = CH_{3}, R^{3} = OCH_{3}$$

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$$(14) R^{1} = R^{2} = R^{4} = CH_{3}, R^{2} = CH_{3}$$

$$(15) R^{1} = R^{2} = R^{4} = CH_{3}, R^{2} = CH_{3}$$

$$(16) R^{1} = CO_{2}CH_{3}, R^{2} = CH_{3}$$

$$(17) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(19) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(19) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(11) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(12) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(13) R^{1} = R^{2} = CH_{3}$$

$$(14) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(15) R^{1} = CH_{3}, R^{2} = CH_{3}$$

$$(16) R^{1} = CH_{3}, R^{2} = CH_{3}$$

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dihydro-2,5,6-trimethoxyphenanthrene (1). Its  $^{13}$ C NMR spectrum (Table 1) contained 17 signals including three methoxyls, two methylenes, five aromatic methines and seven substituted aromatic carbons, three of which were oxygenated. These carbon signals and the high-resolution mass spectrum (m/z 270.1250) established the molecular formula as  $C_{17}H_{18}O_3$ , consistent with a dihydrophenanthrene. The  $^1$ H NMR spectrum (Table 2) showed the expected four proton multiplet signal at  $\delta_{\rm H}$  2.73 (H<sub>2</sub>-9 and H<sub>2</sub>-10) (Letcher & Nhamo, 1971), in addition to three methoxyl signals [ $\delta_{\rm H}$  3.68 (5-OCH<sub>3</sub>), 3.83 (2-OCH<sub>3</sub>)

Table 1 <sup>13</sup>C NMR data for compounds **1**, **2**, **8**, and **13** 

	1	2	8	13
1	113.1	113.3	110.0	114.4
2	158.4	158.3	147.5	154.5
3	111.6	111.3	139.0	113.4
4	129.5	129.7	150.5	129.9
4a	125.3	125.6	120.3	125.6
4b	128.0	120.1	125.3	127.9
5	146.5	142.8	128.3	147.0
6	152.0	145.7	111.6	152.1
7	110.0	108.2	157.5	110.2
8	122.8	118.2	113.0	122.9
8a	131.4	131.6	139.4	131.4
9	29.5	29.7	30.2	29.5
0	30.5	30.5	30.1	30.4
0a	140.5	140.1	134.9	140.9
-OCH <sub>3</sub>	55.1	55.2	_	_
-OCH <sub>3</sub>	_	_	61.1	-
-OCH <sub>3</sub>	_	_	60.0	_
-OCH <sub>3</sub>	60.0	_	_	60.0
-OCH <sub>3</sub>	56.0	56.3	_	56.0
-OCH <sub>3</sub>	_	_	55.1	_

and 3.88 (6-OCH<sub>3</sub>)]. Two sets of coupled aromatic protons were present, an ortho-coupled pair [ $\delta_{\rm H}$  6.74 (d,  $J_{AB}$  = 8.2 Hz, H-8) and 6.92 (d,  $J_{AB}$  = 8.2 Hz, H-7)] and the three spin system of a 1,2,4-trisubstituted aromatic ring  $[\delta_{\rm H} 6.83 \text{ (dd, } J = 2.8 \text{ and } 8.6 \text{ Hz, H-3}), 6.76 \text{ (d, } J = 2.$ Hz, H-1) and 8.36 (d, J=8.6 Hz, H-4)]. The most deshielded resonance [ $\delta_{\rm H}$  8.36] was characteristic of H-4 (or H-5) of a 9,10-dihydrophenanthrene where C-5 (or C-4) bears an oxygen (Majumder & Joardar, 1985). Two of the methoxyl groups could therefore be placed at C-2 and C-5, whilst the third must be attached to C-6, because the <sup>13</sup>C NMR spectrum revealed that there were substituents in both of the positions ortho to one of the methoxyls ( $\delta_C \approx 60$ ). Acquisition of a NOESY spectrum confirmed the substitution pattern. Thus, correlations were observed between the C-2 methoxyl group and both H-1 and H-3, as well as between the C-5 methoxyl and H-4 and the C-6 methoxyl and H-7. The <sup>13</sup>C NMR spectrum was assigned by comparison with similar compounds.

The second and third compounds were identified as 9,10-dihydro-2,6-dimethoxyphenanthren-5-ol (2) and 9,10-dihydro-2,3,7-trimethoxyphenanthrene (3), which have been reported recently from an unidentified Costa Rican *Plagiochila* species (Anton et al., 1997).

The fourth compound was identified as 9,10-dihydro-2,3,4,7-tetramethoxyphenanthrene (4),  $C_{18}H_{20}O_4$ . The presence of a four proton multiplet at  $\delta_H$  2.74 ( $H_2$ -9 and  $H_2$ -10) in the <sup>1</sup>H NMR spectrum (Table 2) revealed that it was a dihydrophenanthrene. The <sup>1</sup>H NMR spectrum also had four methoxyl resonances at  $\delta_H$  3.76 (s, 4-OCH<sub>3</sub>), 3.83 (s, 7-OCH<sub>3</sub>), 3.87 (s, 2-OCH<sub>3</sub>) and 3.90 (s, 3-OCH<sub>3</sub>), as well as an aromatic singlet at  $\delta_H$  6.57 (s, H-1) and a three spin aromatic system [ $\delta_H$  6.81 (dd, J=2.8 and 8.7 Hz, H-6), 6.76 (d, J=2.8 Hz, H-8) and 8.21 (d, J=8.7

Table 2 <sup>1</sup>H NMR parameters for 9,10-dihydrophenanthrenes **1–4, 8, 9** and **13** 

	1	2	3	4	8	9	13
H-1	6.76 d 2.8	6.78 d 2.8	6.73 s	6.57	6.62	6.71 d 2.8	6.75 m
H-2	_	_	_	_	_	_	_
H-3	6.83 dd 2.8, 8.6	6.86 dd 2.8, 8.6	_	_	_	6.76 dd 2.8, 8.5	6.75 m
H-4	8.36 d 8.6	8.36 d 8.6	7.18 s	_	_	8.31 d 8.5	8.31 d 8
H-5	_	_	7.54 d 8.4	8.21 d 8.7	8.18 d 8.5	_	_
H-6	_	_	6.83 dd 2.8, 8.4	6.81 dd 2.8, 8.7	6.81 dd 2.8, 8.5	_	_
H-7	6.92 d 8.2	6.68 d 8.1	_	_	_	6.69 d 8.1	6.75 m
H-8	6.74 d 8.2	6.73 d 8.1	6.77 d 2.8	6.77 d 2.8	6.76 d 2.8	6.72 d 8.1	6.93 d 8.0
H-9,10	2.73 br s	2.75 br s	2.80 m	2.74 br s	2.71 br s	2.73 br s	2.73 br s
2-OCH <sub>3</sub>	3.83 s	3.83 s	3.92 s	3.87 s	_	_	_
3-OCH <sub>3</sub>	_	_	3.89 s	3.90 s	3.96 s	_	_
4-OCH <sub>3</sub>	_	_	_	3.76 s	3.74 s	_	_
5-OCH <sub>3</sub>	3.68 s	_	_	_	_	_	3.69 s
6-OCH <sub>3</sub>	3.88 s	3.92 s	_	_	_	3.91 s	3.89 s
7-OCH <sub>3</sub>	_	_	3.83 s	3.83 s	3.83 s	_	_
2-OH	_	_	_	_	5.76 br s	4.68 br s	4.80 br s
5-OH	_	6.20	_	_	_	6.19 br s	_

Hz, H-5)] similar to that of 1. NOE difference experiments determined the positions of the methoxyl groups. On irradiation of the benzylic protons, NOEs were observed at H-1 ( $\delta_{\rm H}$  6.57, 12%) and H-8 ( $\delta_{\rm H}$  6.76, 8%) and thus established the presence of a proton ( $\delta_{\rm H}$  6.57) at C-1. NOEs were also observed at H-6 (7%) and at H-8 (7%) upon irradiation of the methoxyl protons at  $\delta_{\rm H}$  3.83 (7-OCH<sub>3</sub>), confirming the position of this methoxyl and the substitution pattern of this compound. Whilst this is the first report of this compound in nature, it has previously been reported as a synthetic derivative of 9,10-dihydro-4-methoxyphenanthrene-2,3,7-triol (5) from the orchid *Bulbophyllum vaginatum*; its physical properties were in good agreement with those reported in the literature (Leong, Kang, Harrison, & Powell, 1997).

The next compound proved not to be a 9,10-dihydrophenanthrene but was identified as methyl 6-methoxy-2-methyl-3,4-methylenedioxybenzoate (6). Its <sup>13</sup>C NMR spectrum had 11 signals including an aromatic methyl  $\delta_{\rm C}$ 12.4], a methyl ester  $[\delta_C 168.0 (C=O) \text{ and } 52.1 (OCH_3)]$ , a methoxyl [ $\delta_C$  56.8 (6-OCH<sub>3</sub>)], a methylenedioxy group  $[\delta_{\rm C} \ 101.2]$ , an aromatic methine  $[\delta_{\rm C} \ 92.7 \ (\text{C--}5)]$ , three oxygenated aromatic carbons [ $\delta_{\rm C}$  140.0 (C-3), 148.4 (C-4) and 152.9 (C-6)] and two quaternary carbons [ $\delta_{\rm C}$  117.7 (C-2) and 116.3 (C-1)]. These properties and the mass spectrum (m/z 224.0692) revealed the molecular formula to be C<sub>11</sub>H<sub>12</sub>O<sub>5</sub>. The <sup>1</sup>H NMR spectrum contained a single aromatic proton resonance [ $\delta_{\rm H}$  6.38 (s, H-5)], which indicated a pentasubstituted benzene ring structure. The strongly shielded nature of this proton suggested that it lay between two oxygen substituents. The <sup>1</sup>H spectrum also contained signals for an aromatic methyl [ $\delta_{\rm H}$  2.14 (d, J=0.5 Hz)], two methoxyls [ $\delta_H$  3.74 (s, 6-OCH<sub>3</sub>) and  $\delta_{\rm H}$  3.86 (s, CO<sub>2</sub>CH<sub>3</sub>)] and a methylenedioxy group [ $\delta_{\rm H}$  5.98 (s)]. These spectroscopic properties permitted only two possible structures, **6** and **7**. The crucial HMBC correlation (Fig. 1) from the aromatic methyl protons and the methylene protons to the same oxygenated aromatic carbon (C-3) was sufficient to preclude the latter structure. The other correlations were as expected and enabled the assignment of the <sup>1</sup>H and <sup>13</sup>C NMR spectra.

The  $^{1}H$  NMR spectrum (Table 2) of the next compound, 9,10-dihydro-3,4,7-trimethoxyphenanthren-2-ol (**8**),  $C_{17}H_{18}O_4$ , differed from that of **4** only in the presence of three methoxyls [ $\delta_{\rm H}$  3.74 (s, 4-OCH<sub>3</sub>), 3.83 (s, 7-OCH<sub>3</sub>) and 3.96 (s, 3-OCH<sub>3</sub>)], and one hydroxyl [ $\delta_{\rm H}$  5.76 (s, 2-OH) instead of four methoxyls. The  $^{13}C$  NMR spectrum revealed that two of the methoxyls were *ortho*-disubstituted [ $\delta_{\rm C}$  55.1 (7-OCH<sub>3</sub>), 60.0 and 61.0 (4-OCH<sub>3</sub> and 3-OCH<sub>3</sub>)]. This indicated that the two compounds possessed the same oxygenation pattern. The position of the free hydroxyl group was shown to be C-2, since saturation of one of the methoxyl groups [ $\delta_{\rm H}$  3.83 (7-

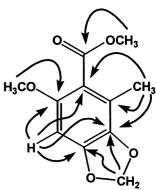


Fig. 1. HMBC correlations for 6.

OCH<sub>3</sub>)] gave NOEs at H-6 (4%) and H-8 (4%) and the other two methoxyls were *ortho*-disubstituted. Further NOEs from H-1 to H<sub>2</sub>-10 (1%) and the phenolic hydroxyl proton (1%), from the phenolic hydroxyl proton to H-1 (2%) and to 3-OCH<sub>3</sub> (1%), from 3-OCH<sub>3</sub> to the phenolic hydroxyl (3%) and 4-OCH<sub>3</sub> (0.3%), from 4-OCH<sub>3</sub> to 3-OCH<sub>3</sub> (0.6%) and H-5 (5%), and from H-5 to H-6 (10%) and 4-OCH<sub>3</sub> (1%) were in accordance with the proposed structure. The carbon assignments are based on comparison with data reported for similar compounds, as well as HMQC and HMBC spectra.

9,10-Dihydro-6-methoxyphenanthrene-2,5-diol  $C_{15}H_{14}O_3$ , was isolated from sample A only, but has been observed in several NMR and GC-mass spectrometry fingerprints. It again showed the characteristic four proton multiplet at  $\delta_{\rm H}$  2.73 (H<sub>2</sub>-9 and H<sub>2</sub>-10) due to a 9,10dihydrophenanthrene system. The <sup>1</sup>H NMR spectrum (Table 2) was almost identical to those of 1 and 2, apart from the presence of only one methoxyl [ $\delta_H$  3.91 (s, 6-OCH<sub>3</sub>)], as well as two phenolic hydroxyls [ $\delta_{\rm H}$  4.68 (s, 2-OH) and  $\delta_{\rm H}$  6.19 (s, 5-OH); both exchangeable with D<sub>2</sub>O]. It seemed therefore that 9 possessed the same oxygenation pattern as 1 and 2. NOE difference experiments readily confirmed this and enabled the position of the methoxyl to be established. Irradiation of the four proton multiplet at  $\delta_{\rm H}$  2.73 (H<sub>2</sub>-9 and H<sub>2</sub>-10) showed NOEs at H-1 (15%) and at H-8 ( $\delta_{\rm H}$  6.71, 1%). Irradiation of H-8 enhanced the H-7 signal and gave a NOE at  $\delta_{\rm H}$  2.73 (H-9 and 10, 1%). Irradiation of the phenolic proton at  $\delta_{\rm H}$ 4.68 afforded NOEs at H-3 (2%) and H-1 (3%) indicating that the hydroxyl group at  $\delta_{\rm H}$  4.68 was attached to C-2. Irradiation of the remaining phenolic hydroxyl at  $\delta_{\rm H}$  6.19 gave NOEs at H-4 (2%) and at the methoxyl group (0.2%), revealing that the substitution pattern of ring A was shown as in structure (9).

The constituents of sample B were isolated by preparative reverse-phase HPLC of the MeOH extract. Three compounds were isolated in addition to 1-4, 6 and 8. The first of these was a tetrasubstituted methyl benzoate, C<sub>12</sub>H<sub>16</sub>O<sub>5</sub>. Its <sup>1</sup>H and <sup>13</sup>C NMR spectra (see Section 3) were similar to those of 6, except that the methylenedioxy group signals had been replaced by those of two methoxyl groups. This suggested that the compound was methyl 3,4,6-trimethoxy-2-methylbenzoate (10). Confirmation of the proposed structure was achieved using HMBC (Fig. 2). The methoxyl group at C-3 (identified because of the correlations involving the aromatic methyl protons) resonated at  $\delta_{\rm C}$  60.5 and therefore had to be ortho to another methoxyl at C-4. The signals for C-4 and C-6 (along with the attached methoxyls) could not be assigned unambiguously, but were made by comparison with those of 6.

The second compound, methyl 4-hydroxy-4'-O-methyllunularate (11), was also a methyl ester [ $\delta_{\rm H}$  4.00 (s, CO<sub>2</sub>CH<sub>3</sub>);  $\delta_{\rm C}$  172.4 ( $CO_2$ CH<sub>3</sub>) and 52.9 (CO<sub>2</sub>CH<sub>3</sub>)] but the occurrence of 12 aromatic carbon signals in the  $^{13}$ C

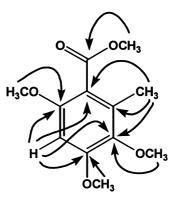


Fig. 2. HMBC correlations for 10.

NMR spectrum (see Experimental), as well as <sup>1</sup>H NMR signals for a para-disubstituted aromatic ring  $[\delta_H 7.07]$ and 6.83 (AA'BB',  $J_{AB}$ =8.0 Hz) H-2', H-3', H-4', H-5'] excluded both 9,10-dihydrophenanthrene and simple benzoic acid structures. The molecular formula, C<sub>17</sub>H<sub>18</sub>O<sub>5</sub>, which was derived from EI-mass spectrometry and <sup>13</sup>C NMR, and the presence of two deshielded methylene groups [ $\delta_H$  3.10 (m,  $H_2$ - $\alpha$ ) and 2.75 (m,  $H_2$ - $\beta$ );  $\delta_C$ 38.7 (C- $\alpha$ ) and 38.2 (C- $\beta$ )] suggested a bibenzyl structure. The second aromatic ring was tetrasubstituted, since a pair of *ortho* aromatic protons was also present  $[\delta_{\rm H} 6.98$ (d, J=8.1 Hz, H-5) and 6.61 (d, J=8.1 Hz, H-6)]. One methoxyl [ $\delta_H$  3.80 (s, 4'-OCH<sub>3</sub>);  $\delta_C$  55.7 (4'-OCH<sub>3</sub>)] and two hydroxyl groups [ $\delta_{\rm H}$  11.50 (s, 2-OH) and 5.65 (br s, 3-OH)] were present in addition to the methoxycarbonyl. One of the hydroxyl groups was placed *ortho* to the ester; hydrogen bonding then accounted for the deshielded nature of the hydroxyl proton. The relative disposition of the substituents was readily determined using HMBC (Fig. 3). Thus, the second methoxyl group was attached to the disubstituted aromatic ring. The more shielded benzylic methylene and the nonchelated hydroxyl proton correlated to different aromatic methine carbons and this established that the groups were mutually para. The remaining correlations of the hydroxyl proton to two oxygenated aromatic carbons revealed that the hydroxyl groups were ortho and that the compound was methyl 4hydroxy-4'-O-methyllunularate (11). The other cor-

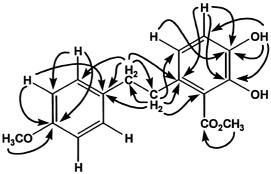


Fig. 3. HMBC correlations for 11.

relations observed were consistent with the proposed structure. Recently, Mues and coworkers reported (Anton et al., 1997) a compound with identical spectroscopic properties to 11 from an unidentified Costa Rican *Plagiochila* species but proposed a different structure (12). In consultation with the Saarbrücken group, we have concluded that structure 12 is untenable, because it would give a  $^{13}$ C NMR resonance at ca.  $\delta$  60 for the *ortho*-disubstituted methoxyl, rather than at  $\delta_{\rm C}$  52.4 as observed; identity of the compounds was confirmed by comparison with GC–mass spectrometry data of the acetylation product of a sample of the bibenzyl provided by Professor Mues.

The final compound isolated proved to be another 9,10-dihydrophenanthrene because it gave NMR signals (Tables 1–2) for a pair of benzylic methylenes [ $\delta_{\rm H}$  2.73 (br s,  $H_2$ -9 and  $H_2$ -10);  $\delta_C$  30.4 and 29.5 (C-9 and C-10)], as well as for the proton at H-4 of a 5-oxygenated 9,10dihydrophenanthrene [ $\delta_{\rm H}$  8.31 (d, J=8 Hz, H-4)]. The NMR shifts were very similar to those of 1, 2 and 9, again indicating a 2,5,6-trioxygenation pattern. The substituents were identified as two methoxyls [ $\delta_{\rm H}$  3.89 (s, 6-OCH<sub>3</sub>) and 3.69 (s, 5-OCH<sub>3</sub>);  $\delta_{\rm C}$  60.0 (5-OCH<sub>3</sub>) and 56.0 (6-OCH<sub>3</sub>)], as well as a hydroxyl group [ $\delta_{\rm H}$  4.80 (br s)]. Although the overlapped nature of the aromatic proton resonances reduced the usefulness of the HMBC spectrum, it was still possible to determine the compound's structure. Since H-4 showed a correlation to an oxygenated carbon which did not, in turn, correlate with any of the methoxyl protons, C-2 must be hydroxylated. The two methoxyl groups were therefore attached to C-5 and C-6 and the compound was 9,10-dihydro-5,6-dimethoxyphenanthren-2-ol (13). The <sup>13</sup>C NMR spectrum Table 1 was assigned using HMBC and by comparison with known compounds.

During the <sup>1</sup>H NMR and GC-mass spectrometric analysis of sample C, a further compound (which was not isolated) was identified as 9,10-dihydro-3,7-dimethoxyphenanthren-2-ol (**14**) by comparison of its <sup>1</sup>H NMR shifts with those of the same compound isolated by Mues and coworkers (Anton et al., 1997).

This is the first report of the isolation of the metabolites of P. spinulosa. Compound **8** is the major aromatic compound in all specimens of P. spinulosa that we have studied. An earlier paper (Inoue & Asakawa, 1988) detailed a GC-mass spectrometric study of Belgian P. spinulosa, in which a number of unidentified compounds were assigned as sesqui- and diterpenoids simply on the basis of their  $M_r$ s. The compounds in the Belgian sample that have  $[M]^+$  m/z 224 (peak 106), m/z 240 (peak 107) and m/z 286 (peak 3) are likely to be **6**, **10** and **8**, respectively, from the present study. Compounds **6** and **10** have also been found in P. killarniensis (Rycroft, Cole, Aslam, Lamont, & Gabriel, 1999).

Although the presence of aromatic compounds in the Hepaticae is well established (Asakawa, 1995), only a

few phenanthrene derivatives have been recorded. 3,7-Dimethoxyphenanthren-2-ol has been found in Indian *Marchantia polymorpha* (Asakawa, Tori, Takikawa, Krishnamurty, & Kar, 1987), 3-methoxyphenanthrene-2,7-diol, 2,3-dimethoxyphenanthren-7-ol and three biphenanthrenes in cultured *M. polymorpha* (Adam & Becker, 1994), jakinenol (sic; 9,10-dihydro-3,4-dimethoxyphenanthren-5-ol) in *Riccardia jackii* (Matsuo, Nozaki, Suzuki, & Nakayama, 1985), eight 9,10-dihydrophenanthrenes (including 2, 3 and 14) and one phenanthrene in an unidentified *Plagiochila* species from Costa Rica (Anton et al., 1997); in addition, two of the 9,10-dihydrophenanthrenes from the Costa Rican *Plagiochila* (but not including 2, 3 or 14 from *P. spinulosa*) have been found in *P. killarniensis* (Rycroft et al., 1999).

## 3. Experimental

# 3.1. General

Mps: uncorr. IR: CHCl<sub>3</sub> soln. EIMS: 70 eV. <sup>1</sup>H (<sup>13</sup>C) NMR: 200 (50) and 360 MHz in CDCl<sub>3</sub> relative to CHCl<sub>3</sub> and CDCl<sub>3</sub> at  $\delta_{\rm H}$  7.25 and  $\delta_{\rm C}$  77.0, respectively or 400 (100) MHz in CDCl<sub>3</sub> relative to TMS.

#### 3.2. Plant material

Voucher samples are deposited in the University of Glasgow herbarium (GL). Sample A: collected by JDC and DSR, Loch Ard Forest, Aberfoyle, Stirlingshire, 7 January 1995, det. DSR (confirmed by D. G. Long, Royal Botanic Garden Edinburgh); sample B: collected by JDC and DSR, Loch Katrine, Stirlingshire, 15 August 1992, det. DSR; sample C: collected by DSR, Rydal Beck, English Lake District, 11 September 1997, det. DSR, voucher No. DSR97137.

## 3.3. Isolation

VLC was performed using a column of silica gel  $G_{254}$  with a step gradient of  $Et_2O$ -petrol as eluent and 50 ml frs were collected. Prep. TLC was carried out on silica gel  $GF_{254}$  with an appropriate concn of  $Et_2O$  in petrol as eluent. Prep. HPLC was performed using a Hypersil 5  $\mu$ m ODS column ( $20 \times 250$  mm, 20 ml min<sup>-1</sup>) and a solvent gradient of 30–60% MeCN– $H_2O$  (both components contained 0.1% TFA) over 65 min. The detection wavelength was 215 nm.

## 3.4. GC-MS

Sample C (29 mg) was extracted with CDCl<sub>3</sub> and the NMR fingerprint (Rycroft, 1996) was recorded. The same solution was used for the GC study, which gave the RIs reported for the individual compounds. GC–MS con-

ditions were generally as reported previously (Rycroft, Cole, & Rong, 1998).

# 3.5. Compound characterization

## 3.5.1. Sample A

Dried and powdered plant material (221 g) gave a crude extract (4.4 g) on extraction with Et<sub>2</sub>O. VLC of the crude extract followed by prep. TLC of the crude compounds gave, in order of increasing polarity, **1** (14 mg), **2** (12 mg), **3** (2 mg), **4** (2 mg), **6** (70 mg), **8** (250 mg) and **9** (4 mg).

- 3.5.1.1. 9,10-Dihydro-2,5,6-trimethoxyphenanthrene (1). Mp 74–76°C (ex petrol-CHCl<sub>3</sub>).  $R_i$  2204. EI–MS (rel. int.): 270 [M]<sup>+</sup> (100), 255 (60), 227 (33), 212 (21), 197 (23), 196 (17), 195 (20), 181 (13), 169 (20), 165 (19), 153 (17), 152 (24), 141 (22), 139 (21), 115 (24); HREI–MS: m/z 270.1250 ( $C_{17}H_{18}O_3$  requires m/z 270.1256).  $^1H$  (200 MHz) and  $^{13}C$  (50 MHz) NMR: Tables 1 and 2.
- 3.5.1.2. 9,10-Dihydro-2,6-dimethoxyphenanthren-5-ol (2). Mp 118–120°C (ex petrol). EI–MS (rel. int.): 256 [M]<sup>+</sup> (100), 241 (28), 223 (16), 213 (17), 197 (13), 195 (20), 181 (34), 169 (17), 165 (12), 153 (33), 152 (36), 151 (12), 141 (22), 139 (18), 115 (29); HREI–MS: m/z 256.1113 (C<sub>16</sub>H<sub>16</sub>O<sub>3</sub> requires m/z 256.1099). <sup>1</sup>H (200 MHz) and <sup>13</sup>C (50 MHz) NMR: Tabs. 1–2. *Acetate* (prepared in situ):  $R_i$  2355.
- 3.5.1.3. 9,10-Dihydro-2,3,7-trimethoxyphenanthrene (3). Gum.  $R_i$  2359. EI–MS: m/z 270 [M]<sup>+</sup> (100), 255 (55), 227 (29), 212 (16), 195 (20), 165 (22), 152 (27), 115 (19). <sup>1</sup>H (360 MHz) NMR: Table 2.
- 3.5.1.4. 9,10-Dihydro-2,3,4,7-tetramethoxyphenanthrene (4). Mp 134–138°C.  $R_i$  2363. EI–MS (rel. int.): 300 [M]<sup>+</sup> (100), 285 (29), 271 (15), 257 (23), 242 (39), 232 (18), 227 (33), 226 (19), 225 (20), 199 (23), 198 (26), 197 (10), 174 (11), 171 (56), 165 (12), 156 (14), 153 (10), 152 (15), 139 (26), 128 (47), 115 (12); HREI–MS: m/z 300.1352 ( $C_{18}H_{20}O_4$  requires m/z 300.1362).  $^1H$  (360 MHz) NMR: Table 2.
- 3.5.1.5. Methyl 6-methoxy-2-methyl-3,4-methylenedioxybenzoate (6). Amorphous, mp 79–80°C. IR  $v_{\rm max}$  cm<sup>-1</sup>: 1733.  $R_i$  1642. EI–MS (rel. int.): 224 [M]<sup>+</sup> (62), 193 (100), 178 (23), 135 (14), 123 (10), 119 (12), 107 (10), 79 (25), 77 (33); HREI–MS: m/z 224.0692 (C<sub>11</sub>H<sub>12</sub>O<sub>5</sub> requires m/z 224.0685). <sup>1</sup>H (200 MHz) and <sup>13</sup>C (50 MHz) NMR: see text.
- 3.5.1.6. 9,10-Dihydro-3,4,7-trimethoxyphenanthren-2-ol (8). Mp 139–142°C (ex Et<sub>2</sub>O–petrol). R<sub>i</sub> 2313 (2455 after acetylation). EI–MS (rel. int.): 286 [M]<sup>+</sup> (88), 271 (32), 256 (25), 253 (16), 243 (14), 239 (50), 228 (54), 227 (12),

- 225 (29), 213 (34), 211 (67), 199 (38), 197 (21), 185 (37), 183 (30), 181 (21), 171 (39), 168 (22), 165 (15), 157 (34), 155 (17), 153 (24), 152 (36), 141 (29), 140 (30), 139 (58), 128 (100), 127 (58), 115 (47); HREI–MS: m/z 286.1181 ( $C_{17}H_{18}O_4$  requires m/z 286.1205).  $^1H$  (200 MHz) and  $^{13}C$  (50 MHz) NMR: Tabs. 1–2.
- 3.5.1.7. 9,10-Dihydro-6-methoxyphenanthrene-2,5-diol (9). Gum. EI–MS (rel. int.): 242 [M]+ (100), 227 (20), 209 (25), 199 (19), 181 (55), 169 (14), 153 (22), 152 (25), 151 (11), 141 (15), 139 (14), 115 (25); HREI–MS: m/z 242.0950 ( $C_{15}H_{14}O_3$  requires m/z 242.0943).  $^1H$  (360 MHz) NMR: Table 2. Diacetate (prepared in situ):  $R_i$  2476.

## 3.5.2. Sample B

Extraction of dried and powdered plant material (161 g) with MeOH gave a crude extract (13.3 g). Prep. HPLC gave, in order of elution, 10, 6, 2, 1, 8, 11, 4, 3 and 13.

- 3.5.2.1. Methyl 3,4,6-trimethoxy-2-methylbenzoate (10).  $R_i$  1706. EI–MS (rel. int.): 240 [M]<sup>+</sup> (100), 225 (50), 209 (45), 193 (52), 165 (46). <sup>1</sup>H NMR (400 MHz): δ 6.37 (s, H-5), 3.89 (s, CO<sub>2</sub>CH<sub>3</sub>), 3.88 (s, 4-OCH<sub>3</sub>), 3.81 (s, 6-OCH<sub>3</sub>), 3.72 (s, 3-OCH<sub>3</sub>), 2.21 (s, H<sub>3</sub>-8). <sup>13</sup>C NMR (100 MHz): δ 168.4 (C-7), 154.2 (C-4), 153.3 (C-6), 141.2 (C-3), 130.6 (C-2), 116.5 (C-1), 95.1 (C-5), 60.5 (3-OCH<sub>3</sub>), 56.5 (6-OCH<sub>3</sub>), 55.9 (4-OCH<sub>3</sub>), 52.2 (CO<sub>2</sub>CH<sub>3</sub>), 12.9 (C-8).
- 3.5.2.2. Methyl 4-hydroxy-4'-O-methyllunularate (11). 
  <sup>1</sup>H NMR (400 MHz): see text. <sup>13</sup>C NMR (100 MHz):  $\delta$  172.4 (C=O), 158.3 (C-4'), 150.2 (C-3), 144.0 (C-4), 135.6 (C-1), 134.5 (C-1'), 129.6 (C-2' and C-6'), 122.6 (C-6), 119.0 (C-5), 114.2 (C-3' and C-5'), 111.8 (C-2), 55.7 (4'-OCH<sub>3</sub>), 52.9 (CO<sub>2</sub>CH<sub>3</sub>), 38.7 (C-α), 38.2 (C-β). Diacetate (prepared in situ): GC–MS R<sub>i</sub> 2619: 386 [M]<sup>+</sup> (15), 302 (5), 284 (2), 270 (4), 269 (2), 181 (2), 149 (3), 121 (100), 91 (2), 78 (3), 77 (3), 43 (17).
- 3.5.2.3. 9,10-Dihydro-5,6-dimethoxyphenanthren-2-ol (13).  $^{1}$ H (400 MHz) and  $^{13}$ C NMR (100 MHz): Tables 1–2. Acetate (prepared in situ): GC–MS  $R_i$  2344: 298 [M] $^{+}$  (39), 256 (100), 241 (40), 225 (2), 198 (4), 197 (5), 181 (8), 169 (3), 165 (5), 152 (4), 141 (4), 139 (2), 137 (6), 115 (6), 43 (21).

#### 3.5.3. Sample C

In addition to the compounds described above, 9,10-dihydro-3,7-dimethoxyphenanthren-2-ol (14) was observed in the <sup>1</sup>H NMR spectrum of the CDCl<sub>3</sub> extract (concn approximately the same as that of 3) and in the GC–MS of the acetylated extract. The compound was identified by comparison with <sup>1</sup>H NMR data in the literature (Anton et al., 1997). *Acetate* of 14 (prepared in situ):

GC–MS *R<sub>i</sub>* 2479: 298 [M]<sup>+</sup> (22), 256 (100), 241 (26), 181 (1), 165 (1), 152 (4), 139 (2), 115 (1), 43 (14).

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#### References

- Adam, K. P., & Becker, H. (1994). Phytochemistry, 35, 139.
- Anton, H., Kraut, L., Mues, R., & Morales Z., M. I. (1997). Phytochemistry, 46, 1069.
- Asakawa, Y. (1995). In W. Hertz, G. W. Kirby, R. E. Moore, W. Steglich, & Ch. Tamm (Eds.), Progress in the Chemistry of Organic Natural Products (Vol. 65, p. 490). Vienna: Springer.
- Asakawa, Y., Tori, M., Takikawa, K., Krishnamurty, H. G., & Kar, S. K. (1987). *Phytochemistry*, 26, 1811.
- Connolly, J. D. (1997). Revista Latinoamericana de Química, 25, 77.
- Connolly, J. D., Harrison, L. J., Phillips, W. R. and Rycroft, D. S. (1984). *Journal of Chemical Research (S)*, 94.

- Connolly, J. D., Harrison, L. J., & Rycroft, D. S. (1982). Journal of the Chemical Society. Chemical Communications. 1236.
- Connolly, J. D., Harrison, L. J., & Rycroft, D. S. (1994). *Journal of Chemical Research* (S), 284.
- Inoue, H., & Asakawa, Y. (1988). Bulletin of the National Science Museum, Series B: Botany (Tokyo), 14, 143.
- Huneck, S., Connolly, J. D., Harrison, L. J., Joseph, R., Phillips, W. R., Rycroft, D. S., Ferguson, G., & Parvez, M. (1986). *Journal of Chemical Research* (M), 1601.
- Huneck, S., Connolly, J. D., Harrison, L. J., Joseph, R. S. I., & Pócs, T. (1984). *Phytochemistry*, 23, 2396.
- Leong, Y.-W., Kang, C.-C., Harrison, L. J., & Powell, A. D. (1997).
  Phytochemistry, 44, 157.
- Letcher, R. M., & Nhamo, L. R. M. (1971). Journal of the Chemical Society (C), 3070.
- Majumder, P. L., & Joardar, M. (1985). *Indian Journal of Chemistry*, Section B, 24, 1192.
- Matsuo, A., Nozaki, H., Suzuki, M., & Nakayama, M. (1985). *Journal of Chemical Research* (M), 1916.
- Rycroft, D. S. (1990). In H. D. Zinsmeister, & R. Mues (Eds.), Bry-ophytes: Their Chemistry and Chemical Taxonomy (p. 109). Oxford: Oxford University Press.
- Rycroft, D. S. (1996). Chemical Communications, 2187.
- Rycroft, D. S., & Cole, W. J. (1998). Phytochemistry, 49, 1641.
- Rycroft, D. S., Cole, W. J., & Aslam, N. (1998). *Phytochemistry*, 49, 145.
- Rycroft, D. S., Cole, W. J., Aslam, N., Lamont, Y. M. and Gabriel, R. (1999). *Phytochemistry*, 50, 1167.
- Rycroft, D. S., Cole, W. J., & Rong, S. (1998). *Phytochemistry*, 48, 1351.
- Söderström, L., Rycroft, D. S., Cole, W. J., & Wei, S. (1999). *Bryobrothera*, in press.
- Watson, E. V. (1981). *British Mosses and Liverworts* (3rd ed., p. 467). Cambridge: Cambridge University Press.