



Two serratane triterpenes from the stem bark of *Picea jezoensis* var. *hondoensis*[☆]

Reiko Tanaka*, Kazuhiro Tsujimoto, Shunyo Matsunaga

Osaka University of Pharmaceutical Sciences, 4-20-1 Nasahara, Takatsuki, Osaka, 569-1094, Japan

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Abstract

Two serratane triterpenoids were isolated from the stem bark of *Picea jezoensis* var. *hondoensis*, together with two known compounds, 3 β -methoxyserrat-14-en-21-one and 3 β -methoxyserrat-14-en-21 α -ol. The serratane triterpenoids were characterized as 14 β ,15 β -epoxy-3 α -methoxyserrat-21 β -ol and 3 α -methoxy-21 β -hydroxyserrat-14-en-16-one, on the basis of chemical and spectroscopic evidence. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: *Picea jezoensis* var. *hondoensis*; Pinaceae; Stem bark; Triterpenes; 14 β ,15 β -epoxy-3 α -methoxyserrat-21 β -ol; 3 α -methoxy-21 β -hydroxyserrat-14-en-16-one

1. Introduction

Previously we reported that the CHCl₃ extract of the stem bark of *Picea jezoensis* (Sieb. et Zucc.) Carr. var. *hondoensis* Rhed. (Japanese name: Touhi, Pinaceae), contained eight serratane triterpenoids including 21 β -methoxyserrat-14-en-3-one, 21 α -methoxyserrat-13-en-3-one and 21 β -hydroxyserrat-14-en-3-one (Tanaka, Mun, Usami & Matsunaga, 1994; Tanaka, Tsuboi & Matsunaga, 1994).

Recently, we reported that the stem bark of *P. jezoensis* var. *hondoensis* contained 21 α -hydroxy-3 β -methoxyserrat-14-en-29-al and 29-nor-3 α -methoxyserrat-14-en-21-one (Tanaka, Tsujimoto, Muraoka & Matsunaga, 1998).

Further careful examination of the stem bark of this extract has led to the isolation of two new triterpenoids, **1** and **2**, besides two known compounds, 3 β -methoxyserrat-14-en-21-one (**3**) (Tanaka, Ohmori, Minoura & Matsunaga, 1996) and 3 β -methoxy serrat-

14-en-21 α -ol (**4**) (Fang, Tsai & Cheng, 1991). This paper deals with the structures of **1** and **2**.

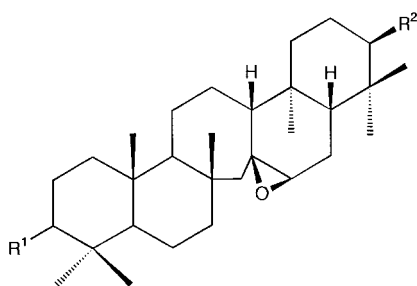
2. Results and discussion

The known compounds were confirmed to be 3 β -methoxyserrat-14-en-21-one (**3**) (Tanaka et al., 1996) and 3 β -methoxyserrat-14-en-21 α -ol (**4**) (Fang et al., 1991), respectively, as physical and spectral data were in good agreement with those already reported in the literature data.

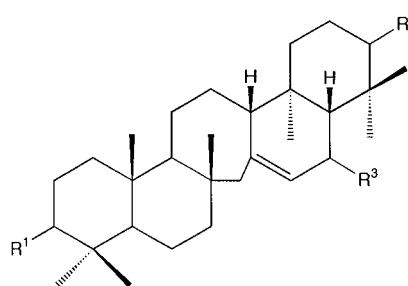
Compound **1** was assigned the molecular formula C₃₁H₅₂O₃, by HREIMS. The ¹H- and ¹³C-NMR spectral data (Tables 1 and 2) exhibited the presence of seven tertiary methyl groups, an equatorial methine proton [δ _H 2.78 (1H, *t*, *J* = 2.5 Hz); δ _C 85.8 (*d*)] geminal to a methoxy group [δ _H 3.31 (3H, *s*, OMe); δ _C 57.1 (*q*)], an equatorial methine proton [δ _H 3.40 (1H, *t*, *J* = 2.4 Hz); δ _C 75.7 (*d*)] geminal to a hydroxyl group (ν _{max} 3533 cm⁻¹), a trisubstituted epoxy ring [δ _H 2.80 (1H, *br s*); δ _C 59.3(*d*) and 61.4 (*s*)]. Acetylation gave a monoacetate (**1a**). The ¹H- and ¹³C-NMR spectra were similar to those of 14 β ,15 β -epoxy-3 β -methoxyserrat-21 β -ol (**5**) which was obtained from *Picea jezoensis* var. *jezoensis*

[☆] Part 4 in the series 'Serratanes from the stem bark of *Picea jezoensis* var. *hondoensis*'; for Part 3 see Tanaka et al. (1998).

* Corresponding author.



	R ¹	R ²
1	α -OMe	OH
1a	α -OMe	OAc
5	β -OMe	OH
5a	β -OMe	OAc



	R ¹	R ²	R ³
2	α -OMe	β -OH	: O
2a	α -OMe	β -OAc	: O
3	β -OMe	: O	H ₂
4	β -OMe	α -OH	H ₂
4a	β -OMe	α -OAc	H ₂
6	α -OMe	β -OH	H ₂
6a	α -OMe	β -OAc	H ₂
7	β -OMe	β -OH	H ₂
7a	β -OMe	β -OAc	H ₂

(Ezomatsu) (Tanaka et al., 1996), except for C-3 configuration. The C-3 chemical shift values of **5** were extremely different from those of **1** which appeared at [δ_{H} 2.62 (1H, *dd*, $J = 12.2, 4.4$ Hz) and δ_{C} 88.5 (*d*)]. Hence, compound **1** was suggested to be the C-3 α epimer of **5**. The conclusive evidence for this structure including an epoxy configuration was confirmed by the NOESY experiment, in which H-3 β correlated with Me-23 and Me-24, and H-15 correlated with H-27 β and Me-28. The EIMS spectra of **1** and **1a** (see Section 3) exhibited the same fragment ion peaks (ions **a**, **d**, **e**, **f**, **g**, **h**, **j**, **k**, **l** and **m**) as **5** and 14 β ,15 β -epoxy-3 β -methoxyserrat-21 β -yl acetate (**5a**) (Tanaka et al., 1996). These data suggested that **1** should be 14 β ,15 β -epoxy-3 α -methoxyserrat-21 β -ol, and this assumption was proved by synthesis. Oxidation of 3 α -methoxyserrat-14-en-21 β -ol (**6**), the most abundant triterpene constituent of this plant, with *m*-chloroperbenzoic acid (*m*-CPBA) furnished an epoxy compound identical in all respects with compound **1**.

Compound **2** was determined the molecular formula as C₃₁H₅₀O₃, from HREIMS. The UV and IR spectra indicated absorption bands for a hydroxyl group (ν_{max} 3396 cm⁻¹) and an α,β -unsaturated six membered ring ketone [λ_{max} 272 nm (ϵ 8000); ν_{max} 1661 cm⁻¹]. The ¹H- and ¹³C-NMR spectra (Tables 1 and 2) exhibited signals for seven tertiary methyl groups, an equatorial methine proton [δ_{H} 2.78 (1H, *t*, $J = 2.5$ Hz); δ_{C} 85.7 (*d*)] geminal to a methoxy group [δ_{H} 3.31 (3H, *s*, OMe); δ_{C} 57.1 (*q*)], an equatorial methine proton [δ_{H} 3.34 (1H, *t*, $J = 2.7$ Hz); δ_{C} 76.8 (*d*)] geminal to a hydroxyl group, a trisubstituted double bond [δ_{H} 5.70 (1H, *br s*); δ_{C} 128.6 (*d*) and 163.7 (*s*)] and a conjugated ketone group [δ_{C} 201.2 (*s*)]. The DEPT spectrum

of **2** revealed seven methyls, nine methylenes, four methines, a methoxy group, two oxymethines, a trisubstituted double bond, five quaternary carbons and a ketone group. Acetylation of compound **2** gave a monoacetate (**2a**), whose C-21 carbinolic methine proton resonance was shifted to δ 4.58 (1H, *t*, $J = 2.7$ Hz). The ¹³C-NMR chemical shifts of **2** related to C-14, C-15, C-16, C-17 and C-18 were considerably different from those of **6**, although the other signals of both compounds had very close chemical shifts. The ¹H-NMR signals of Me-28, Me-29 and Me-30 showed paramagnetic shift ($\Delta\delta_{\text{C}}$ 0.11, 0.24 and 0.33) when compared to those of **6**. These data indicated that **2** must be a serrat-14-en-16-one derivative bearing an axial methoxyl group at C-3 and an axial hydroxyl group at C-21. This assumption was supported by analyzing HMQC, HMBC, ¹H/¹H COSY and NOESY spectra. In the HMBC spectrum, C-16 was correlated with H-15 and H-17 β protons. Accordingly, **2** was proved as 3 α -methoxy-21 β -hydroxyserrat-14-en-16-one; this structure was confirmed by synthesis (Tanaka & Matsunaga, 1991). Treatment of 3 α -methoxyserrat-14-en-21 β -yl acetate (**6a**) with tertiary-butyl chromate in carbon tetrachloride furnished 3 α -methoxy-21 β -acetoxyserrat-14-en-16-one which was identical in all respects with **2a**.

This is the first report for the isolation of **2** in the literature, although 16-oxoserratenediol (3 β ,21 α -dihydroxyserrat-14-en-16-one) and its 3 α ,21 β - and 3 β ,21 β -dihydroxyl analogues (Tsuda, Fujimoto & Kimpara, 1975), 16-oxoclavanol (3 α ,24,30-trihydroxyserrat-14-en-16-one), 16-oxolycoclavanol (3 α ,21 β ,24-trihydroxyserrat-14-en-16-one) and 16-oxoserrat-14-en-16-one (Tsuda, Fujimoto &

Table 1
500 MHz ^1H -NMR spectral data of **1**, **2**, and **2a**^a

H	1	2	2a
1 α	1.18 <i>m</i>	1.22 <i>m</i>	1.24 <i>m</i>
1 β	1.50 <i>m</i>	1.46 <i>dt</i> (13.6, 3.9)	1.46 <i>m</i>
2 α	1.72 <i>m</i>	1.72 <i>m</i>	1.73 <i>m</i>
2 β	1.72 <i>m</i>	1.72 <i>m</i>	1.73 <i>m</i>
3 β	2.78 <i>t</i> (2.5)	2.78 <i>t</i> (2.5)	2.78 <i>t</i> (2.5)
5 α	1.23 <i>dd</i> (13.1, 2.3)	1.28 <i>dd</i> (10.5, 4.6)	1.28 <i>m</i>
6 α	1.43 <i>m</i>	1.43 <i>m</i>	1.38 <i>m</i>
6 β	1.37 <i>m</i>	1.43 <i>m</i>	1.44 <i>m</i>
7 α	1.21 <i>m</i>	1.33 <i>m</i>	1.34 <i>m</i>
7 β	1.34 <i>dt</i> (12.8, 3.1)	1.41 <i>m</i>	1.45 <i>m</i>
9 α	0.89 <i>dd</i> (12.2, 2.1)	1.07 <i>dd</i> (12.2, 2.2)	1.10 <i>dd</i> (12.2, 2.2)
11 α	1.98 <i>m</i>	2.10 <i>ddd</i> (12.2, 7.3, 3.4)	2.13 <i>ddd</i> (12.2, 7.2, 3.4)
11 β	1.28 <i>m</i>	1.24 <i>m</i>	1.28 <i>m</i>
12 α	1.04 <i>m</i>	1.20 <i>m</i>	1.20 <i>m</i>
12 β	1.88 <i>m</i>	1.86 <i>dd</i> (11.6, 7.3)	1.84 <i>dd</i> (11.6, 7.2)
13 β	1.47 <i>dd</i> (15.1, 2.1)	2.33 <i>dd</i> (11.6, 2.0)	2.33 <i>dd</i> (11.6, 2.0)
15	2.80 <i>br s</i>	5.70 <i>br s</i>	5.72 <i>br s</i>
16 α	1.69 <i>ddd</i> (14.6, 13.1, 2.0)	—	—
16 β	1.94 <i>ddd</i> (14.6, 4.3, 2.0)	—	—
17 β	1.46 <i>dd</i> (13.1, 4.3)	2.53 <i>s</i>	2.44 <i>s</i>
19 α	1.52 <i>m</i>	1.54 <i>dt</i> (13.3, 3.5)	1.59 <i>dt</i> (13.5, 3.5)
19 β	1.38 <i>m</i>	1.79 <i>ddd</i> (15.0, 13.3, 3.5)	1.59 <i>ddd</i> (15.0, 13.3, 3.5)
20 α	1.77 <i>ddd</i> (14.8, 4.5, 2.3)	1.90 <i>m</i>	1.88 <i>m</i>
20 β	1.55 <i>m</i>	1.65 <i>ddd</i> (13.8, 6.5, 2.7)	1.72 <i>m</i>
21 α	3.40 <i>t</i> (2.4)	3.34 <i>t</i> (2.7)	4.58 <i>t</i> (2.7)
23	0.91 <i>s</i>	0.93 <i>s</i>	0.93 <i>s</i>
24	0.82 <i>s</i>	0.832 <i>s</i>	0.84 <i>s</i>
25	0.83 <i>s</i>	0.825 <i>s</i>	0.85 <i>s</i>
26	1.07 <i>s</i>	0.86 <i>s</i>	0.90 <i>s</i>
27 α	1.91 <i>d</i> (14.4)	2.44 <i>d</i> (14.7)	2.45 <i>d</i> (15.1)
27 β	0.72 <i>d</i> (14.4)	1.90 <i>d</i> (14.7)	1.92 <i>d</i> (15.1)
28	0.73 <i>s</i>	0.79 <i>s</i>	0.81 <i>s</i>
29	0.89 <i>s</i>	1.12 <i>s</i>	1.18 <i>s</i>
30	0.93 <i>s</i>	1.26 <i>s</i>	1.16 <i>s</i>
OMe	3.31 <i>s</i>	3.31 <i>s</i>	3.31 <i>s</i>
OAc	—	—	2.10 <i>s</i>

^a Measured in CDCl_3 . Assignments were made by HMQC, HMBC, ^1H - ^1H COSY and NOESY experiments.

Kimpara, 1975), lycoclavanin (3 α ,20 β ,21 β ,24-tetrahydroxyserrat-14-en-16-one) (Tsuda, Fujimoto, Morimoto & Sano, 1975), and 16-oxolyclanitin (3 α ,20 β ,21 β ,24,29-pentahydroxyserrat-14-en-16-one) (Tsuda, Fujimoto, Isobe, Sano & Kobayashi, 1974) had been isolated from *Lycopodium clavatum* and *Lycopodium serratum*.

3. Experimental

3.1. General

Mps.: uncorr. Optical rotations: CHCl_3 at 23°; UV: EtOH; IR: KBr discs; ^1H -NMR (500 MHz) and ^{13}C -NMR (125 MHz): CDCl_3 with TMS as internal standard; EIMS: 70 eV (probe). CC: silica gel 60 and alumina 90 (each 70–230 mesh, Merck); TLC: silica gel HF₂₅₄ and PF₂₅₄ (Merck).

3.2. Isolation of compounds

Extraction, isolation of 21 α -hydroxy-3 β -methoxyserrat-14-en-29-al and 29-nor-3 α -methoxyserrat-14-en-21-one by residues A and B from the silica gel CC of the CHCl_3 extract of the stem bark of *P. jezoensis* var. *hondoensis* has been reported (Tanaka et al., 1998).

Repeated silica gel CC (1 kg) of the frs 41–56 (residue C, 35.73 g) of the CHCl_3 extract of *P. jezoensis* var. *hondoensis* gave a crystalline mass (207 mg) from the frs 16–19. Rechromatography with Al_2O_3 eluting with *n*-hexane: C_6H_6 5:1 gave 3 β -methoxyserrat-14-en-21-one (**3**), 111 mg, mp 268–270° (MeOH- CHCl_3), $[\alpha]_D^{25} -29$ (*c* 0.57) (lit. (Tanaka et al., 1996) mp 268.5–270°, $[\alpha]_D^{25} -29$), identical in all respects with an authentic sample. Subsequent CC of residue C with the same solvent afforded 3 α -methoxyserrat-14-en-21 β -ol (**6**) (21.76 g) from frs 21–37, 3 β -methoxyserrat-14-en-21 β -ol (**7**) (2.38 g) from frs 55–72, and a poorly-separ-

able mixt. (1.66 g) from frs 73–81. Acetylation of 1 g of the mixt. with Ac₂O–pyridine (1:1, 10 ml) at room temp. for 24 h and subsequent usual workup gave a residual solid (1.01 g), which was subjected to a 10% AgNO₃ impregnated silica gel (150 g) CC using *n*-hexane–C₆H₆ (5:1) to afford the acetate **7a** (813 mg) from frs 22–78 and 3β-methoxyserrat-14-en-21α-yl acetate (**4a**) (79 mg) from frs 94–102. Hydrolysis of compound **4a** (50 mg) with N/30 KOH/EtOH gave 3β-methoxyserrat-14-en-21α-ol (**4**), (48 mg), mp 318–321.5° (MeOH–CHCl₃), [α]_D –5 (*c* 0.44), which was identified by literature data (Fang et al., 1991).

Repeated silica gel CC (1 kg) of frs 104–122 (residue D, 14.30 g) of the extract yielded a crystalline solid (38 mg), from frs 23–38. Purification of the solid by prep. TLC [plate: 0.5 mm thick, 20 × 20 cm, solvent: CHCl₃–MeOH, 50:1] afforded compound **2** (22 mg). Subsequent CC with the same solvent yielded a crystalline solid (44 mg), from frs 44–49, which was purified by prep. TLC [plate: 0.5 mm thick, 20 × 20 cm, solvent: CHCl₃–MeOH, 50:1] to give compound **1** (39 mg).

3.3. 14β,15β-epoxy-3α-methoxyserratan-21β-ol (**1**)

Prisms, mp 279–281° (MeOH–CHCl₃), [α]_D –36 (*c* 0.12, CHCl₃), HREIMS *m/z* 472.3913 [M]⁺ (C₃₁H₅₂O₃ requires 472.3913), IR *v*_{max} cm^{–1}: 3533 (OH), 2968, 2892, 1457, 1388 and 1360 (gem-dimethyl), 1106, 1067 and 1000; ¹H- and ¹³C-NMR: see Tables 1 and 2; EIMS *m/z* (rel. int) (Tanaka et al., 1996): 472 [M]⁺ (21), 457 [M–Me]⁺ (9), 454.3798 [M–H₂O]⁺ (11), 440.3654 [ion **a**, calc for 440.3652] (36), 425 [a–Me] (12), 422 [a–H₂O] (7), 287 [ion **b**] (12), 257 [ion **d**] (18), 248 [ion **e**] (33), 237.1864 [ion **f**, calc for 237.1853] (16), 224.1766 [ion **g**, calc for 224.1775] (56), 221 [ion **h**] (29), 209.1521 [ion **j**, calc for 209.1540] (52), 203 (ion **k**) (25), 201 [ion **l**] (35), 191 [ion **m**] (23), 189 [ion **n**] (61) and 136 (100).

3.4. Acetylation of **1**

Compound **1** (13 mg) was dissolved in a mixt. of Ac₂O and C₅H₅N (1:1, 1 ml) and the mixt. was kept at room temp. overnight. Usual workup yielded a crude solid (13 mg), which was purified by prep. TLC to afford the corresponding acetate **1a**, 12 mg, mp 238–240° (MeOH–CHCl₃), [α]_D –47 (*c* 0.67, CHCl₃), IR *v*_{max} cm^{–1}: 1738 and 1245 (OAc), 2935, 2872, 1457, 1387 and 1363 (gem-dimethyl), 1165, and 1099; ¹H-NMR (C₅D₅N) δ: 0.73 (3H, *s*, Me-28), 0.81 (3H, *s*, Me-24), 0.83 (3H, *s*, Me-25), 0.84 (3H, *s*, Me-30), 0.88 (3H, *s*, Me-29), 1.03 (3H, *s*, Me-23), 1.21 (3H, *s*, Me-26), 2.06 (3H, *s*, OAc) 2.76 (1H, *t*, *J* = 2.5 Hz, H-3β), 2.78 (1H, *br s*, H-15), 3.32 (3H, *s*, OMe), 4.82 (1H, *t*, *J* = 2.7 Hz, H-21α); ¹³C-NMR: see Table 2; EIMS *m/z*

Table 2

125 MHz ¹³C-NMR spectral data of compounds **1**, **1a**, **2**, and **2a**^a

C	1	1a	2	2a
1	33.6 <i>t</i>	33.7 <i>t</i>	33.4 <i>t</i>	33.5 <i>t</i>
2	20.2 <i>t</i>	20.3 <i>t</i>	20.2 <i>t</i>	20.2 <i>t</i>
3	85.8 <i>d</i>	85.5 <i>d</i>	85.7 <i>d</i>	85.6 <i>d</i>
4	38.0 <i>s</i>	38.2 <i>s</i>	38.0 <i>s</i> ^b	38.0 <i>s</i> ^b
5	50.0 <i>d</i>	50.0 <i>d</i>	50.0 <i>d</i>	50.0 <i>d</i>
6	18.3 <i>t</i>	18.6 <i>t</i>	18.6 <i>t</i>	18.6 <i>t</i>
7	44.4 <i>t</i>	44.9 <i>t</i>	44.7 <i>t</i>	44.7 <i>t</i>
8	39.3 <i>s</i>	39.4 <i>s</i>	38.2 <i>s</i> ^b	38.2 <i>s</i> ^b
9	62.8 <i>d</i>	62.9 <i>d</i>	62.2 <i>d</i>	62.2 <i>d</i>
10	38.0 <i>s</i>	38.3 <i>s</i>	38.0 <i>s</i> ^b	38.1 <i>s</i> ^b
11	25.2 <i>t</i>	25.4 <i>t</i>	26.5 <i>t</i>	26.5 <i>t</i>
12	27.1 <i>t</i>	27.2 <i>t</i>	25.0 <i>t</i>	25.0 <i>t</i>
13	56.8 <i>d</i>	57.3 <i>d</i>	58.7 <i>d</i> ^c	58.8 <i>d</i>
14	61.4 <i>s</i>	60.8 <i>s</i>	163.7 <i>s</i>	163.6 <i>s</i>
15	59.3 <i>d</i>	59.0 <i>d</i>	128.6 <i>d</i>	128.6 <i>d</i>
16	22.8 <i>t</i>	23.0 <i>t</i>	201.2 <i>s</i>	200.4 <i>s</i>
17	38.0 <i>d</i>	39.6 <i>d</i>	58.8 <i>d</i> ^c	59.7 <i>d</i>
18	35.2 <i>s</i>	35.5 <i>s</i>	44.3 <i>s</i>	44.3 <i>s</i>
19	31.8 <i>t</i>	32.8 <i>t</i>	31.4 <i>t</i>	32.2 <i>t</i>
20	25.1 <i>t</i>	23.1 <i>t</i>	24.5 <i>t</i>	22.3 <i>t</i>
21	75.7 <i>d</i>	77.7 <i>d</i>	76.8 <i>d</i>	78.8 <i>d</i>
22	37.1 <i>s</i>	36.4 <i>s</i>	36.7 <i>s</i>	35.9 <i>s</i>
23	28.4 <i>q</i>	28.8 <i>q</i>	28.4 <i>q</i>	28.4 <i>q</i>
24	22.4 <i>q</i>	22.4 <i>q</i>	22.5 <i>q</i>	22.5 <i>q</i>
25	16.3 <i>q</i>	16.5 <i>q</i>	15.8 <i>q</i>	15.8 <i>q</i>
26	20.5 <i>q</i>	20.8 <i>q</i>	20.0 <i>q</i>	20.2 <i>q</i>
27	55.4 <i>t</i>	55.7 <i>t</i>	55.8 <i>t</i>	55.9 <i>t</i>
28	14.7 <i>q</i>	14.8 <i>q</i>	14.8 <i>q</i>	14.7 <i>q</i>
29	22.9 <i>q</i>	22.5 <i>q</i>	21.5 <i>q</i>	21.3 <i>q</i>
30	27.8 <i>q</i>	27.6 <i>q</i>	27.8 <i>q</i>	27.6 <i>q</i>
OMe	57.1 <i>q</i>	56.7 <i>q</i>	57.1 <i>q</i>	57.1 <i>q</i>
OCOMe	–	21.0 <i>q</i>	–	21.3 <i>q</i>
OCOMe	–	170.8 <i>s</i>	–	170.5 <i>s</i>

^a Measured in CDCl₃; **1**, **2**, **2a**, and C₅D₅N: **1a**.^{b,c} May be interchanged within the same column.

z (rel. int) (Tanaka et al., 1996): 514.4020 [M]⁺, calc for 514.4020] (35), 499 [M–Me]⁺ (9), 482 [ion **a**] (8), 467 [a–Me] (4), 454 [M–HOAc]⁺ (10), 439 [M–HOAc–Me]⁺ (12), 287 [ion **b**] (7), 266.1862 [ion **g**, calc for 266.1881] (80), 257 [ion **d**] (20), 251.1626 [ion **j**, calc for 251.1646] (67), 248.2137 [ion **e**, calc for 248.2139] (44), 221 [ion **h**] (27), 203 (ion **k**) (25), 201.1643 [ion **l**, calc for 201.1643] (37), 191 [ion **m**] (30), 189 [ion **n**] (50) and 136 (100).

3.5. Synthesis of **1** from **6**

A solution of *m*-CPBA (30 mg) in CHCl₃ (3 ml) was gradually added to a solution of compound **6** (30 mg) in CHCl₃ (2 ml) with stirring at room temperature for 4 h, when the reaction mixture was washed with 5% aqueous Na₂CO₃ and H₂O. Evaporation of the solvent under reduced pressure afforded a residue which was purified by prep. TLC [plate: 0.5 mm thick, 20 × 20 cm, solvent: CHCl₃–MeOH, 50:1] to give

14 β ,15 β -epoxy-3 α -methoxyserrat-21 β -ol, 26 mg, mp 280–281° (MeOH–CHCl₃), [α]_D –36 (c 0.25, CHCl₃). The resulting product was identified by direct comparison with data for compound **1**.

3.6. 3 α -methoxy-21 β -hydroxyserrat-14-en-16-one (**2**)

Needles, mp 320–322° (MeOH–CHCl₃), [α]_D –83 (c 0.13, CHCl₃), HREIMS: m/z 470.3757 (C₃₁H₅₀O₃ requires 470.3756); UV λ_{\max} (ϵ) nm: 230 sh, 272 (3500, 8000); IR ν_{\max} cm^{–1}: 3396 (OH), 2933, 2861, 1661 (C=C–C=O), 1458, 1387 and 1361 (gem-dimethyl), 1245, 1184, 1132, 1107, 1090, 1103, 986, 935, 876 and 793 (HC=C<); ¹H- and ¹³C-NMR: see Tables 1 and 2; EIMS m/z (rel. int.): 470 [M]⁺ (100), 452 [M–H₂O]⁺ (6), 438 [M–MeOH]⁺ (21), 405 (18), 371 (4), 330 (6), 261 (24), 221 (46), 203 (26), 189 (65).

3.7. Acetylation of **2**

Treatment of compound **2** (10 mg) as described for **1** yielded a crude solid (10 mg), which was purified by prep. TLC to afford the corresponding acetate (**2a**), 10 mg, amorphous solid, EIMS: m/z 512 [M]⁺; ¹H- and ¹³C-NMR: see Tables 1 and 2.

3.8. Synthesis of **2a** from **6a**

A soln of freshly prepd. CrO₂(*O*-*t*-Bu)₂ (1 ml) in CCl₄ (7.5 ml) was dropwise added to a soln of 3 α -methoxyserrat-14-en-21 β -yl acetate (**6a**) (102 mg) in CCl₄ (20 ml) and the mixt. was heated at 80° for 12 h. After cooling, 10 ml of 5% aqueous NaHSO₃ was added to the mixt. to destroy any excess oxidant. The organic layer was washed with H₂O and dried over Na₂SO₄; removal of the solvent under reduced pres-

sure yielded a residual solid (99 mg), which was purified by prep. TLC (plate: 0.5 mm thick, 20 × 20 cm, solvent: CHCl₃–MeOH, 50:1) to give 3 α -methoxy-21 β -acetoxyserrat-14-en-16-one, [M]⁺ m/z : 512. It was identified by direct comparison with the data for compound **2a**.

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