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Allanxanthone B, a polyisoprenylated xanthone from the stem bark of *Allanblackia monticola* Staner L.C

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Abstract

In addition to five known compounds including three xanthones, tovophyllin A, rubraxanthone and garciniafuran, one pentacyclic triterpene, lupeol and one phytosterol, stigmasterol, a polyisoprenylated xanthone named allanxanthone B was isolated from the stem bark of *Allanblackia monticola*. The structure of the new compound was assigned as 2-geranyl-1,3,6-trihydroxy-2',2'-dimethyl[5',6':7,8]xanthone by means of spectroscopic analysis. The antimicrobial activities of some of these compounds against a range of micro-organisms are also reported.

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1. Introduction

Allanblackia monticola Staner L. C., which belongs to the plant family Guttiferae is widely distributed in West Province of Cameroon, where it is used for the treatment of certain human ailments such as respiratory infections, diarrhoea and toothache (Raponda-Walker and Sillans, 1961). Further phytochemical studies of plants belonging to the genus Allanblackia have revealed the presence of xanthones, benzophenones, biflavonoids, phytosterols and saponins (Locksley and Murray, 1971; Blunt et al., 1999; Nkengfack et al., 2002). Some of these compounds exhibit a wide range of biological and pharmacological activities such as cytototoxic, anti-inflammatory, antimicrobial and antifungal (Nagem and Peres, 1997; Nagem et al., 2000); as well as HIV inhibi-

tory activity (Blunt et al., 1999). As part of our continuous search for biologically active compounds from *Allanblackia* species, we now report the phytochemical analysis of the stem bark of *A. monticola* as well as the in vitro antimicrobial activity of some of the isolated compounds.

2. Results and discussion

Extensive column chromatography of a methylene chloride–methanol (1:1) extract of the stem bark of *A. monticola* led to the isolation of a new polyisoprenylated xanthone, allanxanthone B (1), along with three known xanthones, tovophyllin A (3) (Graham et al., 1993); rubraxanthone (2) (Ampofo and Waterman, 1986); garciniafuran (6) (Nilar and Leslie, 2002); one pentacyclic triterpene, lupeol (5) (Wenkert et al., 1978) and one phytosterol, stigmasterol (4) (Diakow et al., 1978).

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Allanxanthone B (1), $C_{28}H_{30}O_6$, was obtained as yellow powder, m.p. 158-160 °C. The compound gave a dark green color with methanolic ferric chloride, indicating that it was phenolic. The UV spectrum of (1) was typical of a 1,3,6,7-oxygenated xanthone and showed bathochromic shift with AlCl₃ and NaOAc reagents indicative of the presence of free hydroxyl groups at C-1 and at C-3 or C-6 (Lockley et al., 1966). The ¹H and ¹³C NMR spectra (see experimental and Table 1) showed the presence of a chelated hydroxyl group $[\delta_H]$ 13.70 (1H, s, 1-OH)], a chelated carbonyl [$\delta_{\rm C}$ 181.5 (s, C-9)], two isolated aromatic protons [$\delta_{\rm H}$ 6.40 (1H, s, H-4) and 6.80 (1H, s, H-5); $\delta_{\rm C}$ 102.0 (d, C-5) and 93.0. (d, C-4), a dimethylchromene ring $[\delta_H 1.30 (6H, s, H_3-$ 5, H₃-6) 5.90 and 8.15 (each 1H, d, J = 10 Hz); $\delta_{\rm C}$ 30.1 (q, C-5, C-6'); 79.5 (s, C-2'); 131.8 (d, C-3'); 122.0 (d, C-4') and a geranyl moiety [$\delta_{\rm H}$ 1.60, 1.64, 1.78 (3H each, s, $3 \times \text{CH}_3$); 2.06 (4H, m, H-4", H-5"); 3.45 (2H, d, J = 7Hz, H-1"); 5.10 (14, brt, J = 7 Hz, H-6"); 5,25 (1H, brt, J = 7 Hz, H-2") (Asai et al., 1995; Pornpipat na Pattalung et al., 1994). The absence of the signal ($\delta_{\rm H}$ 7.75–7.80) due to the aromatic proton (H-8) located at a peri position to the carbonyl group suggested that

Table 1 ¹H (300.135 MHz) and ¹³C (75.469 MHZ) assignments for allanxanthone B (1) in CD₃ COCD₃

Attribution	¹³ C	Multiplicity	¹ H [<i>m</i> , J (Hz)]	HMBC	
1	161.5	S	_	_	
2	111.0	S	_	_	
3	163.0	S	_	_	
4	93.0	d	6.40(s)	4a, 3, 2, 9a	
4a	156.2	S	_	_	
5a	153.5	S	_	_	
5	102.1	d	6.80(s)	4b, 8a, 6	
6	158.2	S	_	_	
7	144.1	S	_	_	
8	109.1	S	_	_	
8a	119.2	S	_	_	
9	181.5	S	_	_	
9a	103.0	S	_	_	
2'	79.5	S	_	_	
3'	131.8	d	5.90 (<i>d</i> , 10.33)	2', 4', 6', 5'	
4′	122.0	d	8.15 (<i>d</i> , 10.33)	3′, 8, 8a	
5'	30.1	q	1.30 (s)	2', 3', 6'	
6′	30.1	q	1.30 (s)	2', 3', 5'	
1"	22.0	t	3.45 (d, 7.20)	2, 1, 3	
2"	123.1	d	5.25 (t, 7.20)	1", 2, 3", 4", 10"	
3"	135.2	S	_	_	
4"	39.7	t	1.66 (m)	3", 5"	
5"	26.5	t	1.64 (m)	4", 5", 3", 7"	
6"	124.1	d	5.10(t)	_	
7"	131.3	S	_	_	
8"	25.6	q	1.53 (s)	_	
9"	17.7	q	1.43 (s)	_	
10"	16.4	q	1.66 (s)	_	
1-OH	_	_	13.4 (s)	1, 2, 9a	
3-OH	-	_	1.68 (brs)	3, 2, 4	
6-OH	-	_	2.68 (brs)	6, 5	

the methyl chromene was fused in an angular manner at C-8 through an oxygen atom at C-7. This was confirmed, on one hand, by the chemical shift of one cisolefinic proton of the chromene ring which appeared in lower field ($\delta_{\rm H}$ 8.15) caused by the anisotropic effect of the carbonyl group and, on the other hand, by the HMBC (Fig. 1) spectrum in which the same olefinic proton showed cross-peaks with C-8 ($\delta_{\rm C}$ 109.1) and C-8a ($\delta_{\rm C}$ 119.2). The cross-peaks between a chelated hydroxyl group at $\delta_{\rm H}$ 13.70 and allylic proton at $\delta_{\rm H}$ 3,45 in the NOESY spectrum (Fig. 2) proved that the geranyl moiety was located at the C-2-position while the remaining proton was at C-4 position. The ortho-position of the oxygenated carbons of ring B was supported by the value of their ¹³C NMR chemical shifts (Nilar and Leslie, 2002). From the above spectroscopic data, structure of allanxanthone B was unambiguously assigned as, {2-geranyl-1,3,6-trihydroxy-2',2'-dimethylpyrano [5',6': 7,8]-xanthone. Further confirmation of the structure of allanxanthone B came from the comparison of its ¹H NMR data with those reported in the literature for toxyloxanthone B (Chen and Chen, 1985) and garcinone B (Ishiguro et al., 1995) isolated, respectively, from Hypericum sampsonii and Hypericum patulum and which have the same substitution pattern as compound 1.

Crude extracts and compounds (1, 2, 3 and 6) were tested for their antimicrobial potency against Gram-pos-

$$H_3$$
C H H O H

Fig. 1. Significant HMBC correlations of compound 1.

Fig. 2. Selected NOESY correlations of compound 1.

Table 2
Antibacterial activity of crude extract and compounds 1, 2, 3 and 6

Micro-organisms	Inhibition zone (mm)						
	Crude extract	1	2	3	6	Oxacillin	
V. anguillaium	_	_	_	_	_		
S. aureus	15	_	12	_	_	30	
C. tropicalis	_	_	-	_	_		

^{-,} not active against the tested micro-organism.

itive (Staphylococcus aureus, Vibrio anguillarium and Candida tropicalis) bacteria in an agar well diffusion assay. As shown in Table 2, crude extract and the four xanthones were found to be inactive against V. angillarium and C. tropicalis. While against S. aureus, crude extracts and compound 2 displayed moderate activity, compounds (1, 3 and 6) were found to be inactive.

3. Experimental

3.1. General experimental procedures

Melting points were determined on a Buchi apparatus and are uncorrected. UV spectra were obtained on a Shimadzu-265 spectrophotometer. NMR spectra were run on a Bruker instrument equipped with a 5mm ¹H and ¹³C probe operating at 300.135 and 75.469 MHz, respectively, with TMS as internal standard. ¹H assignments were made using 2D-COSY and NOESY (mixing time 800 ms experiments while ¹³C assignments were made using 2D - HSQC and HMBC experiments. Silica gel, 230-400 Mesh (Merk) and silica gel 70-230 Mesh (Merck) were used for flash and column chromatography, respectively, while precoated aluminium sheets silica gel 60 F₂₅₄ (Merck) were used for TLC with a mixture of cyclohexane-ethyl acetate as eluents; spot were visualised by UV (254 nm)and (365 nm) or by MeOH-H₂SO₄.

3.2. Plant material

The stem bark of *A. monticola* was collected in July, 2003 at Bagangte in West Province of Cameroon. This sample was identified by Dr. L. ZAPFACK, of Botanic Department, University of Yaoundé I, where a Voucher specimen in on deposit.

3.3. Extraction and isolation

Air dried, powdered stem bark of *A. monticola* (3 kg) was extracted at room temperature with a mixture of CH₂Cl₂–MeOH (1:1) and concentrated to dryness to afford a viscous residue (250 g). This residue was then fractionated by flash column chromatography using silica gel (230–400 mesh) eluted with the mixture of

cyclohexane–EtOA on time (7.5:2.5), cyclohexane–EtOAc (1:1); EtOAc and EtOAc–MeOH (7.5:2.5) to give four main fractions labelled A, B, C and D, respectively. Only fraction A and B showed a significant in vitro antimicrobial activity against a Gram-positive bacteria, *S. aureus* (diameter of inhibition = 13 and 9 mm, respectively).

Fraction A was then column chromatographed over *Si-gel* packed in cyclohexane; gradient elution was effected with cyclohexane–EtOAc mixture. A total of 90 fraction of ca. 300 ml each were collected and combined on the basis of TLC. The pure compounds were obtained either by direct crystallisation or after further purification by column chromatographies. Crystallisation of combined fraction 6–7 eluted with cyclohexane - EtOAc (19:1), gave lupeol (5) (10 g); the combined fractions (18–23) eluted with cyclohexane–EtOAc (18:2) gave tovophyllin A (3) (15 mg); the combined fraction (24–46) eluted with cyclohexane–EtOAc (17:3) afforded stigmasterol (4) (1.2 g) and garciniafuran (6) (1.5 mg).

Fraction B, treated in the same manner as A, yielded allanxanthone B(1) (14 mg) and rubraxanthone (2) (300 mg).

3.4. Antimicrobial assay

The crude extract and purified compounds from the stem bark of *A. monticola* were tested against micro-organisms, *S. aureus* (ATCC 6538), *C. tropicalis* (ATCC 66029) and *V. anguillarum* (ATCC 19264). The qualitative antimicrobial assay employed, was a classical discs diffusion procedure using Mueller Hinton agar (DIFCO). Paper discs were impregnated with 20 μL of DMSO solution of each sample (1 mg/mL) and allowed to evaporated at room temperature. Oxacillin (20 μL of a 1 mg/mL solution) was used as standard positive control. The plates with micro-organisms was incubated at 37 °C for 24 h for *S. aureus* and at 27 °C during 48 h for *V. anguillarum* and *C. tropicalis*. The diameter of inhibition zone around each disc were measured and recorded at the end of the incubating period.

Allanxanthone B (1). Yellow powder, m.p. 158–160 °C, +ESI-TOF-MS m/z 464,2135 ($C_{28}H_{32}06$ require m/z 464,2190) msms m/z (rel. int.): 407(68); 285 (84); 257 (95); 229 (28). UV $\lambda^{\rm Et0H}$ max nm: 242, 253, 310, 348; (+NaOH): 265, 296, 357; (+AlCl₃): 260, 341, 390; (+NaOAc): 290, 354. For ¹H and ¹³C NMR, see Table 1.

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References

- Ampofo, S.A., Waterman, P.G., 1986. Xanthone from Garcinia species. Phytochemistry 25, 2351–2355.
- Asai, F., Tosa, H., Tanaka, T., Iinuma, M., 1995. A xanthone from pericarps of *Garcinia mangosta*. Phytochemistry 39, 943–944.
- Blunt, J.W., Boswell, J.L., Poyd, M., Cardellina II, J.H., Fuller, R.W., 1999. Guttiferone F, the first prenylated benzophenone from Allanblackia stuhlmannii. J. Nat. Prod. 62, 130–132.
- Chen, M., Chen, C., 1985. Xanthones from Hypericum sampsonii. Heterocycles 23, 2543–2548.
- Diakow, P.R.P., Holland, H.L., Taylor, G.L., 1978. ¹³C nuclear magnetic resonance spectra of some C-19-hydroxy, C-5,6-epoxy, C-24-ethyl and C-19-norsteroids. Can. J. Chem. 56, 3121–3127.
- Graham, J.B., Leslie, J.H., Sia, G.-L., Sim, K.-Y., 1993. Triterpenoids, tocotrienols and xanthones from the bark of *Cratoxylum cochinchinense*. Phytochemistry 32 (5), 1245–1251.
- Ishiguro, K., Nakajima, M., Fukumoto, H., Isoi, K., 1995. Cooccurrence of prenylated xanthones and their cyclisation products

- in cell suspension cultures of *Hypericum patulum*. Phytochemistry 38, 867–869.
- Lockley, H.D., Moore, I., Scheinmann, F., 1966. Extractives from Guttiferae part III. The isolation and structure of symphoxanthone and globuxanthone from *Symphonia globulifera* L. J. Chem. Soc. C, 2186–2190.
- Locksley, H.D., Murray, I.G., 1971. Extractives from Guttiferae. Part XIX. The isolation of two benzophenones, six xanthones and two biflavonoids from the heartwood of *Allanblackia floribunda* oliver. J. Chem. Soc. C, 1332–1340.
- Nagem, T.J., Peres, V., 1997. Trioxygenated naturally occurring xanthones. Phytochemistry 44, 199–214.
- Nagem, T.J., de Oliveira, F., Peres, V., 2000. Tetraoxygenated naturally occurring xanthones. Phytochemistry 55, 683–710.
- Nilar, Leslie, J.H., 2002. Xanthones from the heartwood of *Garcinia mangosta* carbon-13 nuclear magnetic resonance spectrocopy of naturally. Occurring substances. Org. Mag. Resonance 11 (7), 337–343.
- Nkengfack, A.E., Azebaze, G.A., Vardamides, J.C., Fomum, Z.T., Van Heerden, F.R., 2002. A prenylated xanthone from Allanblackia floribunda. Phytochemistry 60, 381–384.
- Pornpipat na Pattalung, Thongtheeraparp, W., Wiriyachitra, P., Taylor, W.C., 1994. Xanthones of *Garcinia cowa*. Planta Med. 60, 365–368.
- Raponda-Walker, A., Sillans, R., 1961. Les plantes utiles du GABON. Paul LECHEVALIER, Paris VI.
- Wenkert, E., Baddeley, G.V., Burfitt, I.R., Moreno, L.N., 1978. Carbon-13 nuclear magnetic resonance spectroscopy of naturallyoccurring substances. Org. Magn. Reason. 11, 337–342.