

Immunosuppressive diacetylenes, ceramides and cerebrosides from *Hydrocotyle leucocephala*

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Abstract

Three C-17 diacetylenic compounds (**1–3**), one monoterpenoid (**4**), seven ceramides (leucoceramides A–G, **5a–g**), six cerebrosides (leucocerebrosides A–F, **6a–f**) and nine known compounds were isolated from the methanolic extract of *Hydrocotyle leucocephala*. Their structures were established by spectroscopic methods. The isolated compounds **1–3**, **5a–g**, **6a–f** and **7** were shown to be active in the lipopolysaccharide (LPS) induced cytokine production assay for IL-10, IL-12, and TNF- α .

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1. Introduction

Hydrocotyle leucocephala Cham. & Schlecht. (Apiaceae), commonly known as “orejitas”, “sombrellita”, or Brazilian pennywort, is an aquatic plant originally from Brazil, and widespread worldwide due to its use as an ornamental aquarium plant. The plant is used as a diuretic, antihelminthic and antidiarrheal in Colombia, and is also used as a bioremediator to remove inorganic nutrients from fresh water in recirculating aquaculture systems in Korea (Ma et al., 2003). Previous chemical studies of members of the genus *Hydrocotyle* have resulted in isolation of quercetin-3-*O*-galactoside from *H. umbellata* L. (Adams et al., 1998), *trans*- β -farnesene, thymol methyl ether and

α -terpinenes from *H. sibthorpioides* Lam. and *H. maritima* Honda (Asakawa et al., 1982), monogalactosylmonoacylglycerols and other known compounds from *H. ramiflora* Maxim. (Kwon et al., 1998), oleanane and ursane type glycosides from *H. ranunculoides* Blume (Della Greca et al., 1994a,b), *H. sibthorpioides* (Matsushita et al., 2004), and *Centella asiatica* Urb. (Matsuda et al., 2001), as well as fourteen C-15 diacetylene compounds from *H. sibthorpioides* (Shulte et al., 1973).

As part of our study of medicinal plants collected in Colombia, we report the isolation, identification, and biological evaluation of three new diacetylenes, a monoterpenoid, a mixture of seven leucoceramides (**5a–g**), a mixture of six leucocerebrosides (**6a–f**), along with nine other known compounds (**7–15**) obtained from the methanolic extract of *H. leucocephala*. The mixtures were identified through chemical degradation analyses. The purified compounds

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were evaluated for their potential immunosuppressive activity, by testing in a LPS-induced cytokine induction assay.

2. Results and discussion

The methanolic extract of *Hydrocotyle leucocephala* was partitioned between EtOAc, BuOH, and H₂O. The EtOAc layer was fractionated by column chromatography and HPLC to give three new C-17 diacetylenic compounds (**1–3**), one monoterpenoid (**4**), mixtures of seven ceramides (**5a–g**) and six galactocerebrosides (**6a–f**) as well as nine known compounds (**7–15**).

Compound **1** was obtained as yellow oil. It showed IR absorptions for hydroxyl groups (broad, 3374 cm^{−1}), acetylenic carbons (2235 and 2150 cm^{−1}), and double bonds (1646 cm^{−1}). The UV spectra showed the absorption maxima at 287 (3.00), 269 (3.07), 253 (3.11) nm. These spectroscopic data are characteristic for acetylenic compounds (Shim et al., 1987). The ¹H NMR spectrum (Table 1) showed signals for a vinyl group at δ_{H} 5.94 (*ddd*, 1H, $J = 17.0, 10.2, 5.2$ Hz), 5.48 (*d*, 1H, $J = 17.0$ Hz), and 5.27 (*d*, 1H, $J = 10.2$ Hz), four hydroxymethines at δ_{H} 4.96 (*d*, 1H, $J = 5.2$ Hz), 4.55 (*d*, 1H, $J = 6.5$ Hz), 4.27 (*dt*, 1H, $J = 8.7, 4.0$ Hz), and 3.74 (*dd*, 1H, $J = 6.5, 4.0$ Hz), and a carbon chain with a terminal methyl. The ¹³C NMR spectrum (Table 1) and DEPT experiments exhibited resonances for a vinyl group (δ_{C} 135.9 and 117.8), four hydroxymethines (δ_{C} 63.7, 64.6 $\times 2$ and 75.9), four signals assignable to acetylenic carbons (δ_{C} 78.7, 78.2, 70.9, 70.3), in addition to an aliphatic chain of six methylenes and a terminal methyl. These data suggest a

molecular formula of C₁₇H₂₆O₄ in agreement with the adduct ion LRFABMS at m/z 299 [M+Na−H₂O]⁺. The foregoing data indicated that **1** is a C-17 diacetylenic compound with a vinyl group and four hydroxyls in its structure. Several C-17 diacetylenic derivatives have been isolated from *Panax ginseng* roots, (Shim et al., 1987; Hirakura et al., 1992) and the ¹³C and ¹H NMR spectroscopic data of **1** were very similar to those of ginsenosides I–K by Hirakura et al. (1992). However, some differences were detected, in particular, the presence of four hydroxyl groups never before reported for acetylenic compounds. ¹H–¹H correlations between the resonance at δ_{H} 4.96 with the vinyl protons at δ_{H} 5.94, 5.48 and 5.27, and HMBC correlations for this same proton with the vinyl carbons (δ_{C} 135.9, 117.8) and one acetylenic carbon (δ_{C} 78.7), permitted the assignment of a hydroxyl at position C-3. In the same way, the COSY spectrum showed correlations between the signals at δ_{H} 4.55, 3.74 and 4.27, and the HMBC spectrum showed correlations for the hydroxymethine resonance at δ_{H} 4.55 with the acetylenic carbon at δ_{C} 78.2 leading to assignments of the hydroxyls at positions C-8, C-9 and C-10 as well as that of the acetylenic carbons. Additionally, the hydroxymethine resonance at δ_{H} 4.27 showed HMBC correlations with the methylene signal at δ_{C} 34.8. These correlations allowed us to establish that compound **1** was 1-heptadecen-4,6-diyne-3,8,9,10-tetraol. The stereochemistry for compound **1** at the triol system (C-8, C-9 and C-10) was estimated as α, α, β by the method proposed by Higashibayashi et al. (2003), where the value of the coupling constant for a triol system was characteristic for each diastereomer measured in CD₃OD, D₂O, DMSO or *d*₆-pyridine. From that, the coupling constant

Table 1
¹H- and ¹³C NMR spectroscopic data for compounds **1–3** (CDCl₃)

	1		2		3	
	$\delta_{\text{H}}^{\text{a}}$	$\delta_{\text{C}}^{\text{b}}$	$\delta_{\text{H}}^{\text{a}}$	$\delta_{\text{C}}^{\text{b}}$	$\delta_{\text{H}}^{\text{a}}$	$\delta_{\text{C}}^{\text{b}}$
1	5.48, <i>d</i> , 1H, (17.0)	117.8	5.50, <i>d</i> , 1H, (17)	117.8	5.48, <i>d</i> , 1H, (17.0)	117.5
	5.27, <i>d</i> , 1H, (10.2)	–	5.28, <i>d</i> , 1H, (10.0)	–	5.29, <i>d</i> , 1H, (10.2)	–
2	5.94, <i>ddd</i> , 1H, (17.0, 10.2, 5.2)	135.9	5.94, <i>ddd</i> , 1H, (17.0, 10.0, 5.2)	135.9	5.95, <i>ddd</i> , 1H, (17.0, 10.2, 5.2)	135.6
3	4.96, <i>d</i> , 1H, (5.2)	63.7	4.95, <i>d</i> , 1H, <i>d</i> , (5.2)	63.7	4.95, <i>d</i> , 1H, (5.2)	63.5
4	–	78.7	–	78.8	–	78.6
5	–	70.3	–	71.2	–	70.0
6	–	70.9	–	70.2	–	70.6
7	–	78.2	–	74.9	–	76.1
8	4.55, <i>d</i> , 1H, (6.5)	64.6	5.47, 1H, <i>d</i> , (6.6)	65.7	4.78, <i>d</i> , 1H, (7.3)	62.7
9	3.74, <i>dd</i> , 1H, (6.5, 4.0)	75.9	3.91, <i>dd</i> , 1H, (6.6, 3.1)	74.4	5.15, <i>dd</i> , 1H, (7.3, 2.9)	75.8
10	4.27, <i>dt</i> , 1H, (8.7, 4.0)	64.6	4.10, <i>ddd</i> , 1H, (8.5, 4.4, 3.1)	63.9	4.29, <i>ddd</i> , 1H, (8.3, 5.5, 2.9)	60.8
11	1.79, <i>m</i> , 2H	34.8	1.86, <i>m</i> , 2H	35.0	1.72, <i>m</i> , 2H	34.6
12	1.45, <i>brm</i> , 1H	26.7	1.52, <i>m</i> , 1H	26.8	1.55, <i>m</i> , 1H	26.4
	1.39, <i>brm</i> , 1H	–	1.42, <i>m</i> , 1H	–	1.41, <i>m</i> , 1H	–
13	1.29, <i>m</i> , 2H	29.4	1.29, <i>m</i> , 2H	29.4	1.29, <i>m</i> , 2H	29.0
14	1.29, <i>m</i> , 2H	29.3	1.29, <i>m</i> , 2H	29.3	1.29, <i>m</i> , 2H	28.9
15	1.29, <i>m</i> , 2H	32.0	1.29, <i>m</i> , 2H	32.0	1.29, <i>m</i> , 2H	31.7
16	1.29, <i>m</i> , 2H	22.9	1.29, <i>m</i> , 2H	22.9	1.29, <i>m</i> , 2H	22.6
17	0.88, <i>brt</i> , 3H, (7.0)	14.4	0.89, <i>brt</i> , 3H, (7.0)	14.4	0.89, <i>brt</i> , 3H, (7.0)	14.0
COO	–	–	–	169.4	–	170.3
CH ₃	–	–	2.14, 3H, <i>s</i>	21.0	2.19, 3H, <i>s</i>	20.7

^a δ_{H} (400 MHz) in ppm, (J in Hz).

^b δ_{C} (100 MHz) in ppm.

for a $8\alpha, 9\alpha, 10\alpha$ system in methanol should have a coupling constant value $J_{8,9} = J_{9,10} = 4.0$ Hz, for $8\alpha, 9\beta, 10\alpha$, $J_{8,9} = J_{9,10} = 7.0$ Hz; for $8\alpha, 9\alpha, 10\beta$ $J_{8,9} = 8.0$ Hz, $J_{9,10} = 2.5$; and for $8\alpha, 9\beta, 10\beta$ $J_{8,9} = 2.5$ Hz; $J_{9,10} = 8.0$ Hz. These values are reported in the supporting information of the cited reference, and C-10 was adopted as the terminal extreme to apply the model. The obtained values for compound **1** coupling constant in CD_3OD at positions 8,9,10 were $J_{8,9} = 8.8$ Hz, $J_{9,10} = 2.0$ in agreement with an α, α, β system. From this analysis the structure of compound **1** was established as is shown in Fig. 1.

Compound **2**, isolated as a pale yellow oil, showed very similar UV data to that of **1** and the IR spectrum had the same absorption bands, with the exception of an absorption for a carbonyl group at 1749 cm^{-1} . The ^1H - and ^{13}C NMR spectroscopic data for **2** were very similar to those obtained for **1** (Table 1), except for the presence of an acetyl group with methyl signals at δ_{H} 2.14 (s, 3H), δ_{C} 21.0 and a carbonyl at δ_{C} 169.4. This information, and the LRFABMS data, suggested that the molecular formula of **2** is $\text{C}_{19}\text{H}_{28}\text{O}_5$ (m/z 341 $[\text{M}+\text{Na}-\text{H}_2\text{O}]^+$). From the COSY spectrum, the hydroxymethines at δ_{H} 5.47 (d, 1H, $J = 6.6$ Hz), 3.91 (dd, 1H, $J = 6.6, 3.1$ Hz) and 4.10 (ddd, 1H, $J = 8.5, 4.4, 3.1$ Hz) were assigned at the C-8, C-9 and C-10 positions, respectively. The chemical shift for

the hydroxymethine at δ_{H} 5.47, and its correlations in the HMBC spectrum with the hydroxymethine resonance at C-10 at δ_{C} 63.9 and the carbonyl at δ_{C} 169.4 allowed us to establish that compound **2** is 1-heptadecen-4,6-diyne-8-acetyl-3,9,10-triol. From comparison of the coupling constants for compounds **1** and **2** in the triol system (see Table 1), an $8\alpha, 9\alpha, 10\beta$ stereochemistry was adopted.

Compound **3** was assigned a molecular formula $\text{C}_{19}\text{H}_{28}\text{O}_5$ from the LRFABMS data, and gave UV and IR data closely related to **2**. This suggested that compound **3** is a C-17 diacetylenic compound with a vinyl terminal, three hydroxyl groups, and an acetyl group in its structure. However, from these data a different position of the acetyl group in **3** was detected according to the chemical shifts observed by ^1H and ^{13}C NMR spectroscopy for hydroxymethines at C-8, C-9 and C-10 (Table 1). The COSY spectrum allowed the assignment of the hydroxymethine signals at δ_{H} 4.78 (d, 1H, $J = 7.3$ Hz), 5.15 (dd, 1H, $J = 7.3, 2.9$ Hz), and 4.29 (ddd, 1H, $J = 8.3, 5.5, 2.9$ Hz) to the C-8, C-9 and C-10 positions, respectively. In the HMBC spectrum of **3**, correlations between the hydroxymethine signal at δ_{H} 5.15 (dd, 1H, $J = 7.3, 2.9$ Hz), assigned to H-9 from the COSY spectrum and the carbonyl signal at δ_{C} 170.3 were detected. Accordingly, **3** was identified as 1-heptadecen-4,6-diyne-9-acetyl-3,8,10-triol. From the

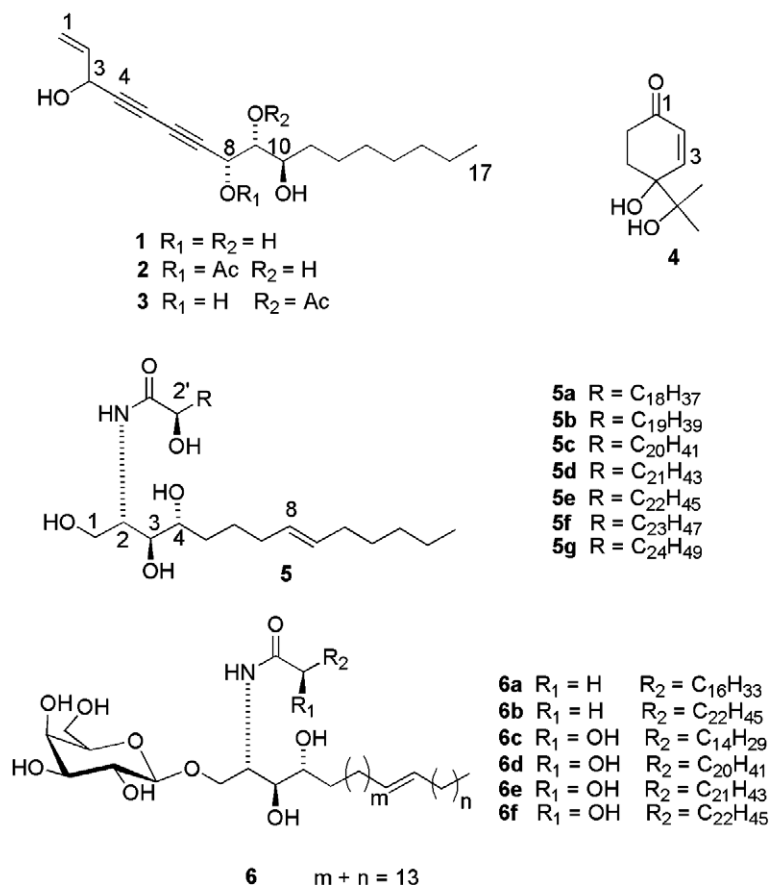


Fig. 1. Structure of the new compounds isolated from *H. leucephala*.

comparison of the coupling constants for compounds **1**, **2** and **3** in the triol system (see Table 1), an $8\alpha,9\alpha,10\beta$ stereochemistry was adopted.

Compounds **1–3** have the same stereochemistry at C-9 and C-10 to that of panaxytriol, a related compound isolated from *Ginseng radix rubra*, but without an hydroxyl at C-8. The stereochemistry of C-9 and C-10 for panaxytriol results from the opening of the epoxide ring of its precursor panaxydol (Kobayashi et al., 1997). From that, the new acetylenic compounds probably derive from the 1-heptadecen-4,6-diyne-9,10-epoxy-3,8-diol, previously isolated from *Panax quinquefolium* (Fujimoto et al., 1991). However, this compound could not be detected in the extract.

Compound **4** has a molecular formula of $C_9H_{14}O_3$ according to the HRFABMS data. The IR spectrum gave absorption bands for OH at 3405 cm^{-1} and an α,β -unsaturated ketone with an absorption at 1683 cm^{-1} . The UV spectrum gave an absorption for an α,β -unsaturated cyclic ketone at 215 nm. The ^1H and ^{13}C NMR spectroscopic data for this compound confirmed the presence of a *cis* α,β -unsaturated ketone with signals at δ_{H} 7.06 (*d*, 1H, $J = 10.3$ Hz), δ_{C} 148.3; 6.06 (*d*, 1H, $J = 10.3$ Hz), δ_{C} 130.3, and a carbonyl at δ_{C} 199.6. The NMR spectroscopic data also showed the presence of a methylene group with

resonances at δ_{H} 2.79 (*ddd*, 1H, $J = 16.8, 10.6, 10.6$ Hz) and 2.46 (*dt*, 1H, $J = 16.8, 4.2$ Hz) and δ_{C} 33.7; a second methylene with resonances at δ_{H} 2.12 (*ddd*, 1H, $J = 14.8, 10.6, 4.2$ Hz), 2.10 (*ddd*, 1H, $J = 14.8, 10.6, 4.2$ Hz) and δ_{C} 29.7; two quaternary carbons, each one bonded to oxygen at δ_{C} 74.4 and 73.1, and two methyls at δ_{H} 1.33 (*s*, 3H) δ_{C} 24.1 and δ_{H} 1.23 (*s*, 3H) and δ_{C} 25.0. From the HMBC spectrum, a cyclohexenone substructure was identified for the correlations between the methylene protons at δ_{H} 2.79 and 2.46 and the carbon resonances for the methylene, the carbonyl and the quaternary carbon at δ_{C} 29.7, 199.6 and 73.1, respectively. The methyl signals at δ_{H} 1.33 (*s*, 3H) and δ_{H} 1.23 (*s*, 3H) were shown to be geminal and correlated with the quaternary carbons at δ_{C} 74.4 and 73.1. These correlations allowed us to identify compound **4** as 4-hydroxy-4-(1-hydroxy-1-methylethyl)-cyclohex-2-en-1-one.

From fraction E-14.5, was obtained a white amorphous solid of a mixture of compounds **5a–g** by crystallization in MeOH. Analysis of the ^1H NMR spectroscopic data (Table 2) showed characteristic signals for ceramides at δ_{H} 8.57 (*d*, $J = 8.7$ Hz) for the amide protons; resonances at δ_{H} 5.10 (*m*, 1H) for the methines bonded to nitrogen, and signals for a hydroxymethylene at δ_{H} 4.49 (*dd*, 1H, $J = 10.3, 4.2$ Hz) and 4.42 (*dd*, 1H, $J = 10.3, 4.1$ Hz). Additionally, signals for three hydroxymethines at δ_{H}

Table 2
 ^1H - and ^{13}C NMR spectroscopic data for compounds **5a–g** and **6a–f** (d_5 -pyridine)

5a–g			6a–f		
	$\delta_{\text{H}}^{\text{a}}$	$\delta_{\text{C}}^{\text{b}}$		$\delta_{\text{H}}^{\text{a}}$	$\delta_{\text{C}}^{\text{b}}$
1	4.49, <i>dd</i> , 1H, (10.3, 4.2)	62.0	1	4.70, <i>dd</i> , 1H, (10.6, 6.6)	70.5
	4.42, <i>dd</i> , 1H, (10.3, 4.1)	–		4.51, <i>d</i> , 1H, (10.6, 4.3)	–
2	5.10, <i>m</i> , 1H	52.9	2	5.27, <i>m</i> , 1H	51.8
3	4.33, <i>dd</i> , 1H, (7.9, 4.7)	76.8	3	4.28, <i>d</i> , 1H, (4.7)	75.9
4	4.28, <i>brt</i> , 1H, (7.9)	72.8	4	4.57, <i>brs</i> , 1H	75.2
5	2.26, <i>m</i> , 2H	33.8	5	2.25, <i>m</i> , 2H	33.9
6	1.70, <i>m</i> , 2H	26.7	6	1.68, <i>m</i> , 2H	26.7
7	2.21, <i>m</i> , 2H,	33.3	7–18 ^c	2.21, <i>m</i> , 2H	33.4
8	5.52, <i>ddd</i> , 1H, (16.7, 6.0, 5.6)	130.8	7–18 ^c	5.52, <i>ddd</i> , 1H, (16.7, 6.0, 5.6)	130.9
9	5.48, <i>ddd</i> , 1H, (16.7, 6.0, 5.6)	130.7	7–18 ^c	5.49, <i>ddd</i> , 1H, (16.7, 6.0, 5.6)	130.7
10	2.00, <i>m</i> , 2H	33.0	7–18 ^c	1.98, <i>m</i> , 2H	33.0
11	1.25, <i>s</i>	25.8	7–18 ^c	1.24, <i>s</i>	30.0
12	1.25, <i>s</i>	32.1	19	1.24, <i>s</i>	32.2
13	1.25, <i>s</i>	22.9	20	1.24, <i>s</i>	22.9
14	1.25, <i>s</i>	14.3	21	1.24, <i>s</i>	14.3
1'	–	175.3	1'	–	175.8
2'	4.61, <i>brs</i> , 1H	72.4	2'	4.57, <i>brt</i> , 1H, (4.0)	72.5
3'	2.02, <i>m</i> , 2H	35.7	3'	2.02, <i>m</i> , 2H	35.6
4'	1.73, <i>m</i> , 2H	25.8	4'	1.73, <i>m</i> , 2H	25.7
1''	–	–	1''	4.90, <i>d</i> , 1H, (7.9)	105.6
2''	–	–	2''	4.00, <i>dd</i> , 1H, (7.9, 8.2)	75.2
3''	–	–	3''	4.18, <i>d</i> , (8.2)	78.6
4''	–	–	4''	4.19, <i>brs</i> , 1H	71.5
5''	–	–	5''	3.86, <i>brs</i> , 1H	78.5
6''	–	–	6''	4.48, <i>d</i> , 1H, (12)	62.8
	–	–		4.34, <i>dd</i> , (12, 5.4)	–
N–H	8.57, <i>d</i> , (8.7)	–	N–H	8.56, <i>d</i> , (9.2)	–

^a δ_{H} (400 MHz) in ppm, (J in Hz).

^b δ_{C} (100 MHz) in ppm.

^c Position of the double bond could not be determined.

4.33 (*dd*, 1H, $J = 7.9, 4.7$ Hz), δ_{H} 4.28 (*brt*, 1H, $J = 7.9$ Hz), δ_{H} 4.61 (*brs*, 1H), a double bond at δ_{H} 5.52 (*ddd*, 1H, $J = 16.7, 6.0, 5.6$ Hz), 5.48 (*ddd*, 1H, $J = 16.7, 6.0, 5.6$ Hz) and two aliphatic chains with resonances at δ_{H} 2.26, 2.02, 1.73, 1.70 and 1.25 were observed. The ^{13}C NMR spectrum (Table 2) showed the resonances for a hydroxymethylene at δ_{C} 62.0, three hydroxymethines at δ_{C} 76.8, 72.8, 72.4, a methine bonded to nitrogen at δ_{C} 52.9, the olefinic carbons at 130.8 and 130.7, a carbonyl at δ_{C} 175.3, and resonances for two aliphatic chains. The methine at δ_{H} 5.10 exhibited correlations in the COSY NMR spectrum with the amide proton (δ_{H} 8.57), the hydroxymethylene (δ_{H} 4.49, 4.42) and the hydroxymethine at δ_{H} 4.33. The same methine proton (δ_{H} 5.10) showed correlations in the HMBC spectrum with the hydroxymethylene (δ_{C} 62.0), the carbonyl (δ_{C} 175.3), and the hydroxymethines at δ_{C} 76.8 and 72.8. The hydroxymethine at δ_{H} 4.61 was correlated in the HMBC spectrum with the carbonyl at δ_{C} 175.3. Therefore, compounds **5a–g** are suggested to have a 2-hydroxy fatty acid moieties with a normal chain and a 1,3,4-trihydroxy substitution in the sphingosine moiety (Yamada et al., 1998). In order to determine the length of each chain, compounds **5a–g** were methanolized with methanolic hydrochloric acid, obtaining a mixture of fatty acid methyl esters and the sphingosine moieties, and each fraction was analyzed by GC-MS. The mixture of methyl esters revealed the presence of 2-hydroxyeicosanoic acid, 2-hydroxyeneicosanoic acid, 2-hydroxydocosanoic acid, 2-hydroxytricosanoic acid, 2-hydroxytetracosanoic acid, 2-hydroxypentacosanoic acid, and 2-hydroxyhexacosanoic acid methyl esters in ratios 1:2:25:16:43:5:3, respectively. The sphingosine fraction was analyzed by GC-MS as a TMSi derivative. In the gas chromatogram for this fraction was observed three peaks with ions at m/z 331 $[\text{M}+\text{TMSi}-\text{H}]^+$, 405 $[\text{M}+2\text{TMSi}]^+$, and 479 $[\text{M}+3\text{TMSi}+\text{H}]^+$. These ions correspond to adducts of a sphingosine moiety with a molecular formula $\text{C}_{14}\text{H}_{29}\text{NO}_3$ bearing one, two and three TMSi derivative groups, respectively. From these data, and the NMR spectroscopic correlations, the presence of 2-amino-1,3,4-trihydroxy-8-tetradecene was found as the only sphingosine. The position of the double bond was proposed on the basis of the strong ion at m/z 417 $[\text{M}+(\text{TMSi})_3-\text{H}-57]^+$, observed in the mass spectrum of the tri-TMSi derivative, probably due to allylic fragmentation at position C-10. The double bond *trans* geometry was established by the coupling constant values and the chemical shift of the methylene carbon adjacent to the olefinic carbon at δ_{C} 33.0 and 33.4 (Yamada et al., 1998). Chemical shifts in the ^1H and ^{13}C NMR spectra and the optical rotation value for the mixture of compounds **5a–g** were shown to be nearly identical with those reported for the ceramide *N*-(2'-*R*-hydroxynonacosanoyl)-*D*-erythro-2,3,4-trihydroxy-2-amino-octadecane (Lourenço et al., 1996). As a result of the comparison, the stereochemistry is presumed to be 2*S*, 3*S*, 4*R* for the ceramide moiety and 2'*R* for the fatty acid moiety. From the above analysis, the

structures of the ceramide mixture **5a–g** were determined as shown in Fig. 1.

From fraction E-14.11, compounds **6a–f** were obtained as white crystals by sequential passages over Sephadex LH-20 and CC. Analysis of the NMR spectra (^1H and ^{13}C , Table 2) showed characteristic signals for cerebrosides (Loukaci et al., 2000) at δ_{H} 8.56 (*d*, $J = 9.2$ Hz) for the amide proton; a galactose moiety, a 2-hydroxy fatty acid with a normal chain, and a 1,3,4-trihydroxy substituted sphingosine, the same as the sphingosine moiety of compounds **5a–g**. Positive FAB MS showed a series of ions at m/z 754–908. To identify the length of each chain (fatty acid and sphingosine), compounds **6a–f** were treated with methanolic hydrochloric acid, to give a mixture of fatty acid methyl esters and the sphingosine moieties with methyl galactoside; both fractions were analyzed by GC-MS in the same way as compounds **5a–g**. The mixture of methyl esters revealed the presence of octadecanoic acid, tetracosanoic acid, 2-hydroxyhexadecanoic acid, 2-hydroxydocosanoic acid, 2-hydroxytricosanoic acid, and 2-hydroxytetracosanoic acid methyl esters in ratios 3:2:2:4:2:3, respectively. The sphingosine fraction was analyzed by GC-MS as a TMSi derivative. In this fraction the presence of 2-amino-1,3,4-trihydroxyeneicosene was found as the only sphingosine due to the only one peak present at R_t 4.00 min and the adduct ion at m/z 429, corresponding to $[\text{M}+\text{TMSi}-\text{H}]^+$. Stereochemistry was adopted as 2*S*, 3*S*, 4*R* and 2'*R* by comparison of the chemical shifts reported for homologous galactocerebrosides (Loukaci et al., 2000). The *trans* geometry of the double bond was established by the coupling constant values and the chemical shift of the methylene carbon adjacent to the olefinic carbon at δ_{C} 33.0 and 33.4 (Yamada et al., 1998). The position of the double bond could not be assigned because of the small amount of sphingosine obtained. Accordingly, the structures of the cerebrosides mixture **6a–f** were determined as shown in Fig. 1.

Additionally, nine known compounds were identified by comparison of their spectroscopic data with literature values as follows: (–)-hinokinin **7** (da Silva et al., 2005), spinasterol **8** (Kojima et al., 1990), lolilide **9** (Okada et al., 1994), dehidrovomifoliol **10** (Kato et al., 1977), 3-hydroxy-5 α ,6 α -epoxy-7-megastigmen-9-one **11** (Duan et al., 2002), kaempferol-3-*O*- α -L-arabinopyranoside **12** (Väsänge et al., 1997), quercetin-3-*O*- α -L-arabinopyranoside **13** (Almeida et al., 1998), quercetin-3-*O*- β -D-galactopyranoside **14** (Arisawa et al., 1984), and kaempferol-3-*O*- β -D-galactopyranoside **15** (Scharbert et al., 2004).

The effects of the test compounds in LPS-induced cytokine production are shown in Table 3, with data expressed as ratios to cytokine production induced by LPS. Compound **1** inhibited LPS-induced IL-10 production but not production of TNF- α and IL-12. Compounds **2**, **3** and **7** inhibited production of all cytokines tested. Although LPS-induced IL-10 was inhibited by compounds **4**, **5a–g**, **6a–f**, **12–15**, these compounds potentiated rather than inhibited the production of IL-12 induced by LPS. No significant effect of these compounds on TNF- α production

Table 3
Inhibitory effects of compounds **1–7**, **9**, **12–15** and prednisolone on cytokine production

Treatment	Cytokine production ratio ^a		
	TNF- α	IL12	IL10
LPS alone	1	1	1
1	0.79	0.98	ND ^b
2	0.72	0.57	ND ^b
3	0.4	ND ^b	ND ^b
4	0.91	2.56	0.34
5a–g	1	2.02	0.34
6a–f	1.13	2.92	0.23
7	0.36	0.44	0.37
9	1.05	3.44	0.35
12	0.89	2.48	0.35
13	0.99	2.73	0.37
14	1.06	2.86	0.36
15	0.88	3.01	0.42
Prednisolone	0.6	0.2	0.41

^a Cytokine production ratios were expressed as ratios to cytokine production induced by LPS. For protocols used, see Section 3. Concentrations used: test compounds 10 μ g/ml; prednisolone, 0.3 μ g/ml.

^b ND, not detected.

was observed. Inhibition of the cytokine production rate for TNF- α and IL-12 was markedly more potent for the acetylenic compounds **2** and **3** compared with **1**, and suggests some importance of the acetyl group in the activity of these compounds against Th1 cells. For compounds **5a–g** and **6a–f**, inhibition of Th2 cells was increased by the presence of the galactose moiety, according to the data for IL-10. For the TNF- α and IL-12 data, no structural relationships could be observed for these compounds. Compound **7**, like other related γ -valerolactone type lignans showed activity as a regulator of TNF- α production (Scharbert et al., 2004). Activity of this type of lignans is dose-dependent (Yeşilada et al., 2001). These findings indicated that compounds **2**, **3**, and **7** are inhibitors against the whole immune response, while compound **1** is a strong Th2 inhibitor, and compounds **4**, **5a–g**, **6a–f**, **8** and **12–15** act as Th1 inducers as well as Th2 inhibitors. This suggests that these compounds, from *H. leucocephala* may be effective therapeutic agents for Th2-type diseases.

3. Experimental

3.1. General experimental procedures

Optical rotations were measured with a JASCO DIP-370 digital polarimeter. UV spectra were recorded on a UV2100 UV–vis recording spectrometer (Shimadzu). IR spectra were recorded on a JASCO Fourier transform infrared spectrometer (FT/IR-420). NMR spectra (400 MHz for ¹H NMR, 100 MHz for ¹³C, both use TMS as internal standard) were measured on a Bruker AVANCE 400 spectrometer. EIMS were obtained on a JEOL JMSDX-303 and FABMS run on a SX-102A instrument. The solid supports for column chromatography and gel permeation chroma-

tography (GPC) were: silica gel 60N (Spherical Neutrial, Kanto Kagaku), Toyopearl HW-40 (Tosoh Bioscience), and Sephadex LH-20 (Pharmacia). The packed columns used were: Shodex H-2001, 2002 for GPC (CHCl₃); Shodex Asahipack GS310-2G for GPC (MeOH); Mightysil Si60 (250 \times 20 mm; Merck) for HPLC normal phase and Mightysil RP-18 (250 \times 20 mm; Merck) for HPLC-ODS. For HPLC JASCO equipment was used with detectors UV at 256 nm and RI. For GC-MS an HP-5870 series instrument was used, coupled to a mass analyzer Automass JMS-AM150 (JEOL). A DB-1 (30 m \times 0.25 mm \times 25 μ m film thickness) column was utilized. The injector temperature was maintained at 250 $^{\circ}$ C. The column oven was programmed from 50 to 300 $^{\circ}$ C, at 30 $^{\circ}$ /min. The carrier gas flow (He) was at 1.0 mL/min. The ratios of compounds were determined by area peaks in the chromatogram. All other solvents used for this work were analytical grade.

3.2. Plant material

Aerial parts of *Hydrocotyle leucocephala* Cham. & Schlecht. were collected in October 2002 from Manizales City in the central part of Colombia. A voucher specimen was identified by Dr. Julio Betancourt at the Instituto de Ciencias Naturales at Universidad Nacional de Colombia and registered as COL495157 specimen number 15. The plant material was dried in air.

3.3. Extraction and isolation

The dried aerial parts from *Hydrocotyle leucocephala* (662 g) were crushed and extracted three times with MeOH at 60 $^{\circ}$ C. Then, the methanolic extract was concentrated in vacuo to give a residue (102.6 g), which was partitioned between EtOAc, BuOH, and H₂O. The EtOAc layer was concentrated to obtain a residue (21 g), which was fractionated by silica gel CC eluted with different solvents of increasing polarity (hexane – EtOAc, EtOAc – MeOH) to obtain 15 fractions (E1–E15).

Fraction E7 (0.9 g) was further fractionated by silica gel CC with a discontinuous gradient of CHCl₃–MeOH (100:0–90:10) as eluent to give six fractions (E7.1–E7.6). In turn, fraction E7.1 (606 mg) was purified over silica gel column with a discontinuous gradient of CHCl₃–MeOH (100:0–98:2) to obtain 10 fractions (E7.1.1–E7.1.10), from that, fractions E7.1.1–E7.1.3 (84 mg) were combined and then separated by GPC (CHCl₃) to afford **7** (43 mg) and **8** (4 mg). Fraction E7.1.6 was subjected to GPC (CHCl₃) to yield **2** (3 mg) and **3** (2 mg). Fraction E9 (2.1 g) was fractionated by silica gel CC with a discontinuous gradient of CHCl₃–MeOH (100:0–90:10) as eluent to give 14 fractions (E9.1–E9.14). Compound **1** (12 mg) was isolated by GPC (CHCl₃) from E9.12.

The mixture of E11–E13 (2.2 g) was fractionated by silica gel CC with a discontinuous gradient of CHCl₃–MeOH (100:0–90:10) as eluent to give nine fractions (E11.1–E11.9). From these, E11.2 (510 mg) was fractionated by GPC using Toyopearl HW-40 with CHCl₃–MeOH (1:1)

as eluent to give six fractions (E11.2.1–E11.2.6). E11.2.5 (80 mg) was selected for further fractionation by GPC (CHCl_3), and led to seven fractions (E11.2.5.1–E11.2.5.7). From that, Fraction E11.5.7 corresponds to **9** (7 mg). Fraction E11.2.4 (100 mg) was isolated by GPC (CHCl_3) to produce eight additional fractions (E11.2.4.1–E11.2.4.8). From these, fractions E11.2.4.4 and E11.2.5.4 (18 mg) were recombined and fractionated by HPLC on a silica gel column with CHCl_3 –MeOH 95:5 to obtain **10** (2 mg) and **11** (3 mg). From E11.9 was obtained **4** (5 mg) by two successive GPC purifications with CHCl_3 as solvent.

Fractions E14 and E15 (5.1 g) were recombined and then fractionated by silica gel CC with a discontinuous gradient of CHCl_3 –MeOH (90:10–7:3) to give 16 further fractions (E14.1–E14.16). From that, fractions E14.11–E14.12 (1.6 g) were recombined and then fractionated by Sephadex LH-20 with MeOH as eluent to give **12** (18 mg) and **13** (12 mg). Fractions E14.14 and E14.15 (1.3 g), were combined and then fractionated by Sephadex LH-20 with MeOH as eluent to give **14** (378 mg) and **15** (90 mg). From E14.11.5 was obtained the mixture of ceramides **5a–g** (7 mg) by crystallization in MeOH. Finally silica gel CC of E14.11.6–E14.11.7 using a discontinuous gradient of CHCl_3 –MeOH (95:5–7:3) led to purification of **6a–f** (5 mg).

3.4. Methanolysis of **5a–g** and **6a–f**

Analysis of compounds present in fractions **5** (6 mg) and **6** (2 mg) was made after methanolysis with 5% HCl in MeOH (3 mL) at 70 °C for 18 h (Yamada et al., 2001). The reaction mixture was extracted with hexane and concentrated in vacuo to obtain the mixtures of fatty acid methyl esters for analysis by GC-MS. The methanol layer was concentrated in vacuo and redissolved in MeOH four times in order to remove the acidic residues and obtain the sphingosine moiety and methyl galactoside. The sphingosine moiety was analyzed by GC-MS as trimethylsilyl (TMSi) ethers using *N*-trimethylsilylimidazole (merck)/pyridine (1:1) for 1 h at 70 °C for the derivatization process. Then, the TMSi ethers of sphingosine were analyzed using the same chromatographic method as for the fatty acid methyl esters. Ratios between compounds were established according to their area peaks by GC.

3.5. 1-Heptadecen-4,6-diyn-3,8 α ,9 α ,10 β -tetraol (**1**)

Pale yellow oil, $[\alpha]_D^{25} + 5.5$ (MeOH; *c* 1.08); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ) 287 (3.00), 269 (3.07), 253 (3.11); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3374, 2360, 2340, 2235, 2150; for ^1H and ^{13}C NMR spectra, see Table 1; EIMS 70 eV, *m/z* (rel. int): 238 (3), 203 (7), 195 (1), 165 (3), 149 (5), 134 (34), 119 (56), 118 (100); FABMS *m/z* 299 $[\text{M}+\text{Na}-\text{H}_2\text{O}]^+$, 277 $[\text{M}+\text{H}-\text{H}_2\text{O}]^+$.

3.6. 1-Heptadecen-4,6-diyn-8 α -acetyl-3,9 α ,10 β -triol (**2**)

Pale yellow oil, $[\alpha]_D^{25} - 6.0$ (MeOH; *c* 0.9); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ) 284 (3.99), 269 (4.08), 254 (4.08); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 2359,

2340, 2239, 2145, 1749; for ^1H and ^{13}C NMR spectra, see Table 1; EIMS 70 eV, *m/z* (rel. int): 354 (1), 319 (1), 309(1), 259(7), 207 (10), 152 (32) 150 (52), 136(57), 134 (95); FABMS *m/z* 341 $[\text{M}+\text{Na}-\text{H}_2\text{O}]^+$, 281 $[(\text{M}+\text{Na})-\text{H}_2\text{O}-\text{CH}_3\text{COOH}]^+$.

3.7. 1-Heptadecen-4,6-diyn-9 α -acetyl-3,8 α ,10 β -triol (**3**)

Pale yellow oil, $[\alpha]_D^{25} + 24.0$ (MeOH; *c* 0.2); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ) 284 (2.87), 269 (2.99), 256 (3.04); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 2359, 2341, 2239, 2147, 1749; for ^1H and ^{13}C NMR spectra, see Table 1; EIMS 70 eV, *m/z* (rel. int): 353 (1), 309 (1), 259 (9), 207 (4), 203 (5), 152 (14), 150 (9), 136 (10), 134 (41), 122 (85), 118 (100); FABMS *m/z* 337 $[\text{M}+\text{H}]^+$.

3.8. 4-Hydroxy-4-(1-hydroxy-1-methylethyl)-cyclohex-2-en-1-one (**4**)

Colorless oil, $[\alpha]_D^{25} + 5.8$ (MeOH; *c* 0.8); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ) 215 (3.99); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3405, 2362, 2338, 1683; ^1H NMR (CDCl_3 , 400 MHz, TMS) δ 7.06 (1H, *d*, *J* = 10.3 Hz, H-3), 6.06 (1H, *d*, *J* = 10.3 Hz, H-2), 2.79 (1H, *ddd*, *J* = 16.8, 10.6, 10.6 Hz, H-6a), 2.46 (1H, *dt*, *J* = 16.8, 4.2 Hz, H-6b), 2.12 (1H, *ddd*, *J* = 14.8, 10.6, 4.2 Hz, H-5a), 2.10 (1H, *ddd*, *J* = 14.8, 10.6, 4.2 Hz, H-5b); ^{13}C NMR (CDCl_3 , 100 MHz, TMSi) δ 199.6 (C-1), 148.3 (C-3), 130.3 (C-2), 74.4 (C-7), 73.1 (C-4), 29.7 (C-5), 33.7 (C-6), 24.1 (C-8), 25.0 (C-9). EIMS 70 eV, *m/z* (rel. int.): 166 (33), 151 (22), 137 (9), 126 (100), 123 (28), 108 (46), 81 (59), 79 (64), 59 (93); HRFABMS *m/z* $[\text{M}+\text{Na}]^+$ 193.0864 (calcd. for $\text{C}_9\text{H}_{14}\text{O}_3\text{Na}$, 193.0841).

3.9. Leucoceramides A–G (**5a–g**)

Amorphous solid, $[\alpha]_D^{25} + 11.8$ (Pyridine; *c* 0.35); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3330, 2917, 1622; for ^1H (*d*₅-pyridine, 400 MHz) and ^{13}C NMR (*d*₅-pyridine, 100 MHz) spectra, see Table 2; 2-hydroxyfatty acid methyl esters EIMS 70 eV, *m/z* (rel. int.): C-20 342 $[\text{M}]^+$; C-21 356 $[\text{M}]^+$; C-22 370 $[\text{M}]^+$; C-23 384 $[\text{M}]^+$; C-24 398 $[\text{M}]^+$; C-25 412 $[\text{M}]^+$; C-26 426 $[\text{M}]^+$. Sphingosine EIMS 70 eV, *m/z* (rel. int): 331(100) $[\text{M}+\text{TMSi}-\text{H}]^+$, 417 (13). FABMS positive *m/z* **5a** $\text{C}_{34}\text{H}_{67}\text{NO}_5$ 570, 610; **5b** $\text{C}_{35}\text{H}_{69}\text{NO}_5$, 584, 624; **5c** $\text{C}_{36}\text{H}_{71}\text{NO}_5$, 598, 638; **5d** $\text{C}_{37}\text{H}_{73}\text{NO}_5$, 612, 652; **5e** $\text{C}_{38}\text{H}_{75}\text{NO}_5$, 626, 666; **5f** $\text{C}_{39}\text{H}_{77}\text{NO}_5$, 640, 680; **5g** $\text{C}_{40}\text{H}_{79}\text{NO}_5$, 654, 694; corresponding to $[\text{M}+\text{H}]^+$ and $[\text{M}+\text{Na}+\text{H}_2\text{O}]^+$, respectively.

3.10. Leucocerebrosides A–F (**6a–f**)

Amorphous solid, $[\alpha]_D^{25} - 1.3$ (Pyridine; *c* 0.28); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3378, 2920, 1630; for ^1H (*d*₅-pyridine, 400 MHz) and ^{13}C (*d*₅-pyridine, 100 MHz) spectra, see Table 2; fatty acid methyl esters EIMS 70 eV, *m/z* (rel. int): C-18 298 $[\text{M}]^+$; C-24 370 $[\text{M}]^+$. 2-hydroxyfatty acid methyl esters EIMS 70 eV, *m/z* (rel. int): C-16 286 $[\text{M}]^+$; C-22 370 $[\text{M}]^+$; C-23 384 $[\text{M}]^+$; C-24 398 $[\text{M}]^+$. Sphingosine EIMS 70 eV,

m/z (rel. int): 429 (100) $[M+TMSi]^+$. FAB-MS m/z **6a** $C_{45}H_{87}NO_9$ m/z 768 $[M-H_2O+H]^+$; **6b** $C_{51}H_{99}NO_{10}$ m/z 852 $[M-H_2O+H]^+$; **6c** $C_{43}H_{83}NO_{10}$ m/z 754 $[M-H_2O+H]^+$, not detected $[M+Na]^+$; **6d** $C_{49}H_{95}NO_{10}$ m/z not detected $[M-H_2O+H]^+$, 880 $[M+Na]^+$; **6e** $C_{50}H_{97}NO_{10}$ m/z 852 $[M-H_2O+H]^+$, not detected $[M+Na]^+$; **6f** $C_{51}H_{99}NO_{10}$ m/z 866 $[M-H_2O+H]^+$, 908 $[M+Na]^+$.

3.11. Bioassays for cytokine activity

The inhibition of cytokine production was measured for human peripheral blood mononuclear (PBMC)s cells (1×10^6 /ml) which were isolated from heparinized venous blood by Ficoll–Hypaque gradient density centrifugation according to standard procedures (Boyum, 1968). After isolation, the cells were treated with 100 ng/ml of lipopolysaccharide (LPS) derived from *Escherichia coli* 055:B5 (Sigma, St Louis, MO) in the absence or presence of the test compounds (10 μ g/ml) for 48 h at 37 °C. Prednisolone (0.3 μ g/ml) (Sigma) was used as reference sample for inhibition of LPS-induced cytokine production. Then, cytokines in the supernatants from these cultures were analyzed by commercial ELISA kits. ELISA kits for human tumor necrosis factor TNF- α , IL-10 and IL-12 were purchased from BioSource International, Inc. (Camarillo, CA).

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