EFFECTS OF EFFUSING MOLECULES ON A SUSPENDED TARGET *

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ABSTRACT

Knowledge of the angular distribution of velocity vectors of molecules streaming from an office into vacuum m effusion expenments is important in a variety of applications Particular applications are in torsion-effusion experiments, where the recoil force depends on angular distribution, and in target-collection experiments, where the fraction of effusing vapor collected by the target depends on angular distribution Recently, commercial vapor pressure balances based on the target-collection principle have become avadable We present a theoretical analysis of the needed angular distribution for the general case of effusion orifices which are frustums of right circular cones, a case which includes cylinders We apply this analysis to target-collection problems m which the target, collimator, or collector 1s circular or rmg-shaped and coaxial with the orifice Both mass effects and force effects are included We consider the case in which a fraction ν ($0 \le \nu \le 1$) of the impinging molecules stick permanently to the target and m which reflected molecules have a different "temperature" than do the impinging ones We have available a FORTRAN program which calculates angular distributions and fractions of effusing molecules which strike a given target or collimator For the special case of cyhndncal onflces, closed-form equations based on derivations by Clausing are presented for the target-collection probability

INTRODUCTION

The vapor pressure and/or chemical composition of the vapor of a substance, particularly at elevated temperatures, can be obtamed by a modification of the Knudsen-effusion method [1,2] in which a fraction of the effusing vapor lmpmges on a cold target A vanety of means are available [3] for analyzing effects on the target mass change [4], counting of radioactivity

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[5], X-ray fluorescence analysis [6], force imparted by lmpmgmg vapor [7], etc The method combined with mass spectrometry [8] renders it especially powerful Recently, an automatic target-collection apparatus mcorporatmg a microbalance has been made available commercially [9]

Onflces used m Knudsen-effusion cells are usually circular m cross section, their defining sections are frustums of right circular cones [10,11]. often the special case of a cyhnder The target used, or the collimator used to define the part of the effusing vapor which strikes the target, is circular and coaxial with the orifice for simplicity of analysis Correlation of effects on the target with the vapor pressure in the effusion cell requires knowledge of the fraction of the effusmg vapor which strikes the target. Additional factors of importance, particularly where the force exerted on the target is of interest, are the fraction of impinging molecules which stick to the target, the angular distribution of momentum vectors of the molecules striking the target and of any reboundmg from the target, the temperature of the source of the effusing molecules, and the effective temperature of any molecules rebounding from the target A theoretical analysis including these factors is available [10] and is outlined here

THEORY

We consider the model described before and Illustrated m Fig. 1 A Knudsen-effusion cell with an effusion orifice whose flow-rate limiting section is in the shape of a right circular come frustum, included would be cylindrical onfices in which the semiapex angle of the cone is zero and ideal orifices in which the height of the frustum is zero Either diverging or

Fig 1 Geometric representation of an effusion-cell and target combination Γ is one-half of the angle subtended at the orifice by the target or collimator θ is the angle representing the direction of the velocity vector of an effusing molecule

convergmg comcal orifices can be considered The target or collimator 1s circular, coaxial with the orifice, and sufficiently far removed from the onflce that the onflce appears effectively as a point source of effusmg molecules The quantities of interest are the rate at which molecules impinge on the target and the force exerted on the target Each can be related to the vapor pressure m the effusion cell

We consider the cases in which all molecules striking the target condense permanently and m whch a fraction of the molecules re-evaporate or are reflected We define the followmg terms

- Γ = half the angle subtended at the orifice by a diameter of the target or colhmator
- ν = the fraction of effused molecules striking the target which reevaporate or are reflected
- G_F = the "transmission probability" of the orifice-target system, 1 e the fraction of molecules entering the effusion onflce which effuse and stnke the target

$$
f = the force exerted on the target
$$

- θ = the angle between a line originating at the center of the orifice exit and the common axis of the onflce and the target
- $P =$ the pressure of vapor within the effusion cell
- dG_{θ} = the fraction of molecules entering the effusion orifice which effuse and strike the target within an incremental ring between θ and $\theta + d\theta$

It is convenient to express dG_{θ} by

$$
dG_{\theta} = 2Q(\theta) \sin \theta \cos \theta d\theta \tag{1}
$$

The function $Q(\theta)$ has been derived as a function of θ and tabulated [10] or can be calculated with an available computer program [14] G_r is then expressed as

$$
G_{\Gamma} = 2 \int_0^{\Gamma} Q(\theta) \sin \theta \cos \theta \, d\theta \tag{2}
$$

The same program [14] calculates G_{Γ}

Equation (2) for the special case of a cylindrical effusion orifice was treated by Clausmg [12] and Freeman and Searcy [13] but numerical integrations were required Edwards [10] showed the linear approximations of Clausmg [12] to be bghly accurate and obtained a closed-form expression of G_r for cylindrical orifices, that expression is given here in the Appendix

The Knudsen equation [1,2,6] modified by G_r is used to calculate the vapor pressure *P* m the effusion cell

$$
P = (dg/dt)(2\pi RT/M)^{1/2}/(G_{\Gamma}A)
$$
 (3)

in which dg/dt is the rate of accumulation of mass on the target, A is the area of the entrance to the orifice from the effusion cell, T is the tempera-

$$
(1)
$$

ture of the cell, and M is the molecular weight of the effusing vapor If ν is other than zero, then the right side of eqn (3) would be divided by $(1 - \nu)$ The force f_s exerted on the target when $\nu = 0$ is expressed by

$$
f_{s} = (3/2) PA \int_{0}^{\Gamma} Q(\theta) \sin \theta \cos^{2} \theta d\theta
$$
 (4)

If the force on the target 1s measured, say by a torsion balance [3,7], the vapor pressure m the cell can be expressed by rearrangement of eqn (4)

When ν is greater than zero, the effect on the force exerted is complicated and dependent, among other things, on the distributions of speeds and of direction of momentum vectors of the reflected or re-evaporated molecules In the case in which the molecules leaving the target have momentum vectors distributed accordmg to the Knudsen cosine law [3,15] and a Boltzmann speed distribution, so they can be assigned an effective temperature T' , the expression for the force f_t is

$$
f_{\rm t} = (1/2)PA\bigg[\Big(3\int_0^{\Gamma} Q(\theta) \sin \theta \cos^2 \theta \, \mathrm{d}\theta\Big) + \nu (T'/T)^{1/2} G_{\Gamma}\bigg] \tag{5}
$$

Assumption of the Knudsen cosine law 1s usually accurate Equation (5) 1s most useful in cases in which the effusate possesses significant vapor pressure at the temperature of the target and thermal accommodation 1s sufficient to assign to the molecules leavmg the target the temperature of the target.

The force exerted by molecules impinging on a target can be important even m cases m whch the rate of mass accumulation IS measured by a mucrobalance from which the target is suspended [3] That force can become apparent when the temperature of the effusion cell 1s changed or when the nature of the vaporization reaction changes For instance, if the furnace heatmg the effusion cell were turned off, effusion would stop, and the mass of the target would appear to increase due to loss of the upward force of impinging molecules A similar effect would be observed if a sudden decrease m pressure m the effusion cell occurred when a volatile phase m the cell was exhausted

APPENDIX

The transmission probability, G_r of a cylindrical-orifice and target system can be expressed accurately in closed form [10] Consider a cylindrical effusion orifice with length of L and radius of r Let

$$
c = 2r/L \tag{A1}
$$

Further, consider the angle

$$
\alpha = \tan^{-1}c \tag{A2}
$$

When the target subtends at the orifice an angle greater than 2α , then some parts of the target at its rim are invisible to the interior of the Knudsen cell, all vapor stnkmg that part of the target comes by reflection off the walls of the orifice Hence, two expressions for G_r are needed, one when $\Gamma \le \alpha$ and another when $\Gamma > \alpha$

We define

$$
a = 1 - \left[(c+1)(c^2+1)^{1/2} - 1 \right] / \left[c(c^2+1)^{1/2} + c^2 \right]
$$
 (A3)

Then G_r is given by the following expressions

$$
G_{\Gamma}(\Gamma < \alpha) = \sin^2 \Gamma - 4(1 - 2a) \left[\tan \Gamma(c^2 + 1)(c^2 - \tan^2 \Gamma)^{1/2} / (1 + \tan^2 \Gamma) - (c^3/2)(\sin 2\Gamma + 2\Gamma) + 2\sin^{-1}(\tan \Gamma/c) \right. \\ \left. + (c^2 - 2)(c^2 + 1)I \right] / (3\pi c^2) - 2(1 - a) \left[2(c^2 + 1)I - (c^2 + 2 + 2 \tan^2 \Gamma) \sin^{-1}(\tan \Gamma/c) / (1 + \tan^2 \Gamma) \right. \\ \left. - \tan \Gamma(c^2 - \tan^2 \Gamma)^{1/2} / (1 + \tan^2 \Gamma) \right] / (\pi c^2) \tag{A4}
$$

m whuzh

$$
I = \frac{1}{(c^2 + 1)^{1/2}} \left[\tan^{-1} \left(\frac{\tan \Gamma}{\left[(c^2 + 1)^{1/2} - c \right] \left[c + (c^2 - \tan^2 \Gamma)^{1/2} \right]} \right) + \tan^{-1} \left(\frac{\tan \Gamma}{\left[(c^2 + 1)^{1/2} + c \right] \left[c + (c^2 - \tan^2 \Gamma)^{1/2} \right]} \right) \right]
$$
(A5)

and

$$
G_{\Gamma}(\Gamma > \alpha) = a \, \sin^2 \Gamma - 4(1 - 2a) \left[\pi + (c^2 - 2)(c^2 + 1)I' \right] \\
- (c^3/2)(\sin 2\Gamma + 2\Gamma) \left] / (3\pi c^2) - 2(1 - a) \left[2(c^2 + 1)I' \right] \\
- (\pi/2)(c^2 + 2) \right] / (\pi c^2)
$$
\n(A6)

m whch

$$
I' = \pi/2(c^2 + 1)^{1/2}
$$
 (A7)

Table Al gives a test of eqns (A4) and (A6) The first column gives the value of L/r of the effusion orifice The second column gives the angle Γ subtended at the orifice by the target The third column gives the correct value of G_r calculated by accurate numerical integrations of exact equations with the available computer program [14] mentioned earlier The fourth column gives values of G_r calculated with eqns (A4) and (A6) The values of Γ were chosen for convenience in applying the numerical integrations as well as for the purposes of this test

It is seen that only for large values of Γ and for $L/r > 4$ do any maccuracies in equations $(A4)$ and $(A6)$ appear that could be of experimen-

TABLE

 G_{Γ} for cylindrical onfices

L/r	Γ (°)	G_{Γ} from numerical integrations	G_{Γ} from eqs $(A4)$ and $(A6)$	
1000	10 37	0 0 2 0 5 6	0 0 2 0 5 5	
	1787	0 0 4 3 2 6	0 04 3 09	
	30 98	0 0 8 5 1 0	0 0 8 4 0 2	
	37 54	0 1 0 5 6	0 1 0 3 9	
	50 65	0 1422	0 1 3 9 4	
	57 21	01573	01540	
	70 33	01793	01753	
	7688	01859	01817	
	90 00	01909	01866	
20 00	1273	0 0 1 7 1 4	001692	
	1976	002932	0 0 2 8 6 1	
	2678	0 04198	0 04061	
	4083	0 0 6 6 6 8	006382	
	4786	007777	0 0 7 4 2 1	
	61 90	009557	0 0 9 0 8 7	
	6893	01018	0 0 9 6 7 1	
	8298	01086	0 1 0 3 1	
	90 00	0 1 0 9 4	0 1 0 3 8	

TABLE Al (continued)

tal importance In any expected, practical case Γ would be less than 45° and L/r would be less than 10 In such cases, eqns $(A4)$ and $(A6)$ will always be accurate within 2% When L/r is 4 or less, these equations can be considered exact Of far greater concern should always be conformity of conditions of the experiment to the model [10] on which derivation of eqns (A4) and (A6) as well as those m the computer program [14] 1s based

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