UNITED STATES PATENT OFFICE.

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PROCESS OF PHOTOGRAPHIC PRINTING.

SPECIFICATION forming part of Letters Patent No. 345,938, dated July 20, 1886.

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To all whom it may concern:

Be it known that I, REDFIELD B. WEST, of Guilford, in the county of New Haven and State of Connecticut, have invented a new Improvement in Processes of Photographic Printing; and I do hereby declare the following to be a full, clear, and exact description of the

This invention relates to an improvement 10 in the process of photographic printing described in another application, Serial No. 167,289. In that application the paper is sensitized by subjection to a bath consisting of a solution in water of potassium bichromate and 15 mercuric chloride. After the paper has been properly dried it is printed upon by exposure to light under a negative, in the usual manner. The photographic prints thus produced are of a yellow color, which not being always 20 desirable, the object of my present invention is to change this yellow tint to a more desirable violet-black; and the invention consists in the treatment of the prints, as hereinafter described, and particularly recited in the claims. 25 After washing the prints in clear water I subject them to a developing-bath produced by a solution of the following compound in water. The developing compound consists of pyro-

gallol, two parts; gallic acid, eight parts; fer-30 rous sulphate (dried) or other ferrous salt soluble in water, ten parts; sodium hyposulphite, (dried,) eighty parts. This compound is prepared in mass, and sold for use when required. The solution is made by dissolving the com-35 pound in water in the proportion of two per

cent. on the compound to the water. By employing the pyrogallol in the solution I am enabled to change the color which it produces to a brown or any intermediate shade by sub-10 jecting the prints to a toning-bath containing

lead nitrite. Without the pyrogallol a violetblack is produced by an aqueous solution of the three other ingredients; but the color is not subject to change in the toning bath. The

45 ferrous salt referred to is preferably ferrous sulphate; but I sometimes use ferrous tartrate or other ferrous salt, or produce these salts in the bath by adding to the compound an alkaline tartrate, &c.

The advantage in substituting other iron salts for the sulphate is that by so doing I prints which are obtained by the action of

can vary the color developed within certain limits, according to the ferrous salt used and the proportion of pyrogallol, from the violetblack produced by the sulphate alone to a 53 warmer tone, using the toning-bath only when it is necessary to obtain well-marked browns.

In my previous processes I added alum to the gallic acid, ferrous sulphate, and sodium hyposulphite for the purpose of obtaining 60 prints having clear whites; but I find this addition unnecessary with the method of bleaching which I now use, and objectionable, as it prevents the development of a good color. The print should be left in the developing bath 65 from five to ten minutes, then removed from the bath, then washed in clear water, and then may be bleached by the action of a dilute solution of chlorine or bromine. bleaching action of the halogens should be 70 modified by the presence of an acid, as it is then more even.

The chlorine bath is best prepared by dissolving one part of calcium hypochlorite in three hundred parts of water and adding about 75 two parts of an acid, preferably phosphoric.

After the prints have been bleached they may be mounted in the usual manner; or, if a brown color is desirable, they are toned by subjecting them to a bath consisting of a one 80 or two per cent. solution of lead nitrite or a mixture of potassium or sodium nitrite and lead nitrate; but should a cold gray be preferred to the brown, then add to the toningbath from two to five per cent. of mercuric 8: chloride. The prints are then washed in clear

The proportions which I have mentioned for the ingredients in the several compounds and baths are such as in practice have pro- 90 duced the best results; but they may be varied somewhat without departing from my inven-

I would state that in order to attain the best results the paper should possess a consider- 95 able degree of porosity and be thickly coated with starch on both sides, and the paper should be free from alkaline particles.

I claim-

1. The herein-described composition for 100 bath for the development of photographic

light upon paper sensitized with potassium bichromate and mercuric chloride, said compound consisting of pyrogallol, gallic acid, a ferrous salt soluble in water, and sodium hy-5 posulphite, in the proportions and substan-

tially as described.

2. The process herein described for developing photographic prints which are obtained by the action of light upon paper sensitized ic with potassium bichromate and mercuric chloride, consisting in subjecting the print to a bath composed of a two per cent. solution of pyrogallol, gallic acid, a ferrous salt soluble in water, and sodium hyposulphite, in the pro-15 portions and substantially as described.

3. The herein-described improvement in bleaching photographic prints, consisting in

subjecting the prints to a solution of calcium hypochlorite and phosphoric acid, in the proportions and substantially as described.

4. The herein described improvement in toning photograpic prints which are produced by the action of light upon paper sensitized with potassium bichromate and mercuric chloride where a cold gray color is desired, con- 25 sisting in subjecting the prints to a toningbath composed of a solution of lead nitrite and mercuric chloride, in the proportions substantially as described.

REDFIELD B. WEST.

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Witnesses:

G. B. SPENCER, CHRISTOPHERUS SPENCER.